

1 **Valorization of CH₄ emissions into high-added-value products: Assessing**
2 **the production of ectoine coupled with CH₄ abatement**

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5 **Abstract:**

6 This study assessed an innovative strategy for the valorization of dilute methane emissions based on the bio-
7 conversion of CH₄ (the second most important greenhouse gas (GHG)) into ectoine by the methanotrophic
8 ectoine-producing strain *Methylobacterium alcaliphilum* 20 Z. The influence of CH₄ (2-20 %), Cu²⁺ (0.05-
9 50μM) and NaCl (0-9 %) concentration as well as temperature (25-35 °C) on ectoine synthesis and specific
10 CH₄ biodegradation rate was evaluated for the first time. Concentrations of 20 % CH₄ (at 3 % NaCl, 0.05 μM
11 Cu²⁺, 25 °C) and 6 % NaCl (at 4 % CH₄, 0.05 μM Cu²⁺, 25 °C) supported the maximum intra-cellular ectoine
12 production yield (31.0 ± 1.7 and 66.9 ± 4.2 mg g biomass⁻¹, respectively). On the other hand, extra-cellular
13 ectoine concentrations of up to 4.7 ± 0.1 mg L⁻¹ were detected at high temperatures and Cu²⁺ concentrations
14 (30 °C and 50 μM), despite this methanotroph has not been previously classified as an ectoine-excreting
15 strain. This research demonstrated the feasibility of the bio-conversion of dilute emissions of methane into
16 high-added value products in an attempt to develop a sustainable GHG bioeconomy.

17 **Keywords:** Ectoine, Greenhouse Gas, Methane treatment, Methane biorefinery

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20 1. Introduction

21 Methane (CH₄) is the second most important greenhouse gas (GHG) emitted nowadays as a
22 result of its high global warming potential (25 times higher than that of CO₂) and emission
23 rates (United States Environmental Protection Agency, 2015). Despite CH₄ can be used as
24 an energy vector for electricity and heat generation when concentrations are higher than 20
25 %, more than 56 % of anthropogenic CH₄ emissions worldwide contain concentrations
26 lower than 5 %. When applied to these dilute emissions (such as exhaust gases from
27 landfills or coal mines), current CH₄ abatement technologies are neither environmentally
28 friendly nor cost-effective (Avalos Ramirez et al., 2012).

29 Nowadays, the lack of a suitable approach to prevent the adverse environmental effects of
30 CH₄ has encouraged both political initiatives to control these GHG emissions and an
31 intensive research on novel strategies for CH₄ abatement (European Environmental
32 Agency, 2015). In this regard, the biological abatement of dilute CH₄ emissions combined
33 with the production of high-added value products represents, if properly tailored, a cost-
34 effective alternative to mitigate CH₄ emissions. This CH₄ biorefinery approach would avoid
35 the negative environmental effects of methane emissions while turning its treatment into a
36 profitable process.

37 Ectoine (1,4,5,6-tetrahydro-2-methyl-4-pyrimidinecarboxylic acid) is one of the most
38 valuable microbial protective compounds against osmotic dehydration, as well as an
39 efficient stabilizer for enzymes and nucleic acids (Pastor et al., 2010). This compound has
40 attracted recent attention based on the high retail value that purified ectoine reaches in the
41 cosmetic industry (approximately \$1300 kg⁻¹) (Strong et al., 2015). In 1999, Khmelenina et
42 al. demonstrated that some moderate halophilic methanotrophs such as *Methylobacterium*

43 *alcaliphilum* 20Z were able to produce and accumulate ectoine inside the cell (Kaluzhnaya
44 et al., 2001; Khmelenina et al., 2000, 1999). These studies, conducted at high CH₄
45 concentrations, represented the first proof of the ability of CH₄-oxidizing bacteria to
46 produce ectoine. However, little is known about the influence of environmental conditions
47 on the bioproduction of this secondary metabolite when combined with the abatement of
48 dilute CH₄ emissions. Furthermore, no studies on the production of extra-cellular ectoine
49 (naturally excreted to the medium by specific excreting strains) by methanotrophs have
50 been carried out to date.

51 The present study aimed at systematically elucidating the influence of CH₄, copper (Cu²⁺)
52 and NaCl concentrations, as well as temperature, on the extra and intra-cellular ectoine
53 production using the strain *Methylobacterium alcaliphilum* 20Z.

54

55 **2. Materials and Methods**

56 *2.1. Chemicals and mineral salt medium*

57 The mineral salt medium (MSM) used was a modified Brunner medium prepared according
58 to Kalyuzhnaya et al. (2008) with a final pH of 9.1. NaCl and CuCl₂·2H₂O were
59 supplemented to the MSM at the different concentrations tested (Table 1). All chemicals
60 and reagents were purchased from Panreac (Barcelona, Spain) with a purity higher than
61 99.0 %. CH₄ was purchased from Abello-Linde, S.A. (Barcelona, Spain) with a purity of at
62 least 99.5 %.

63 *2.2. Microorganisms and inoculum preparation*

64 The methanotrophic strain used in this study, *Methylobacterium alcaliphilum* 20Z
65 (Kalyuzhnaya et al., 2008), was purchased from DSMZ (Leibniz-Institut).
66 *Methylobacterium alcaliphilum* 20Z is an halophilic alkalitolerant methanotrophic strain
67 able to produce ectoine in the presence of NaCl (Khmelenina et al., 2000). Briefly, a 10×
68 dilution of the liquid *Methylobacterium alcaliphilum* 20Z stock culture from DSMZ was
69 grown at 25 °C in 120 mL glass bottles containing 90 mL of MSM at 3 % of NaCl and 0.05
70 μM Cu²⁺. The bottles were closed with gas-tight butyl septa and metallic caps and 50 % v/v
71 of the air headspace was replaced by CH₄. The inoculum was ready to use in the batch
72 cultivation tests when a bacterial biomass concentration of 0.1 ± 0.06 g L⁻¹ was achieved.

73 2.3. Batch cultivation tests

74 Five series of 13-day tests (TS) were performed in duplicate to evaluate the influence of
75 different environmental factors (CH₄, NaCl, Cu²⁺, T) on the production of extra and intra-
76 cellular ectoine by *Methylobacterium alcaliphilum* 20 Z. Sterile batch gas-tight reactors
77 (1.2 L) containing 190 mL of MSM and inoculated with 10 mL of the inoculum above
78 described (to an initial concentration of 0.05 ± 0.001 g L⁻¹) were used in each tests series.
79 The reactors were closed with gas-tight butyl septa and plastic screw caps. Unless
80 otherwise specified, all tests were initially supplied with a CH₄ headspace concentration of
81 25 g CH₄ m⁻³ (4 %), 3 % of NaCl, 0.05 μM Cu²⁺ and incubated at 25 °C under a continuous
82 magnetic agitation of 600 rpm. The parameter evaluated in each specific test series was:
83 TS1: methane concentration (2, 4 and 20 %), TS2: copper concentration (0.05, 25 and 50
84 μM), TS3: NaCl concentration (0, 3, 6 and 9 %), and TS4: temperature (25, 30, 35 °C). A
85 final test (TS5) combining the optimum conditions for ectoine production obtained in the

86 previous test series (20 % CH₄, 50 μM Cu²⁺, 6 % NaCl and 30 °C) was also carried out
87 (Table 1).

88 <Table 1>

Table 1. Cultivation conditions evaluated during *Methylobacterium alcaliphilum* 20Z batch cultivation tests.

Test series (TS)	Operating conditions			
	CH ₄ (%)	Cu ²⁺ (μM)	NaCl (%)	T (°C)
TS1	2, 4, 20	0.05	3	25
TS2	20	0.05, 25, 50	3	25
TS3	20	0.05	0, 3, 6, 9	25
TS4	20	0.05	3	25, 30, 35
TS5	20	50	6	30

89

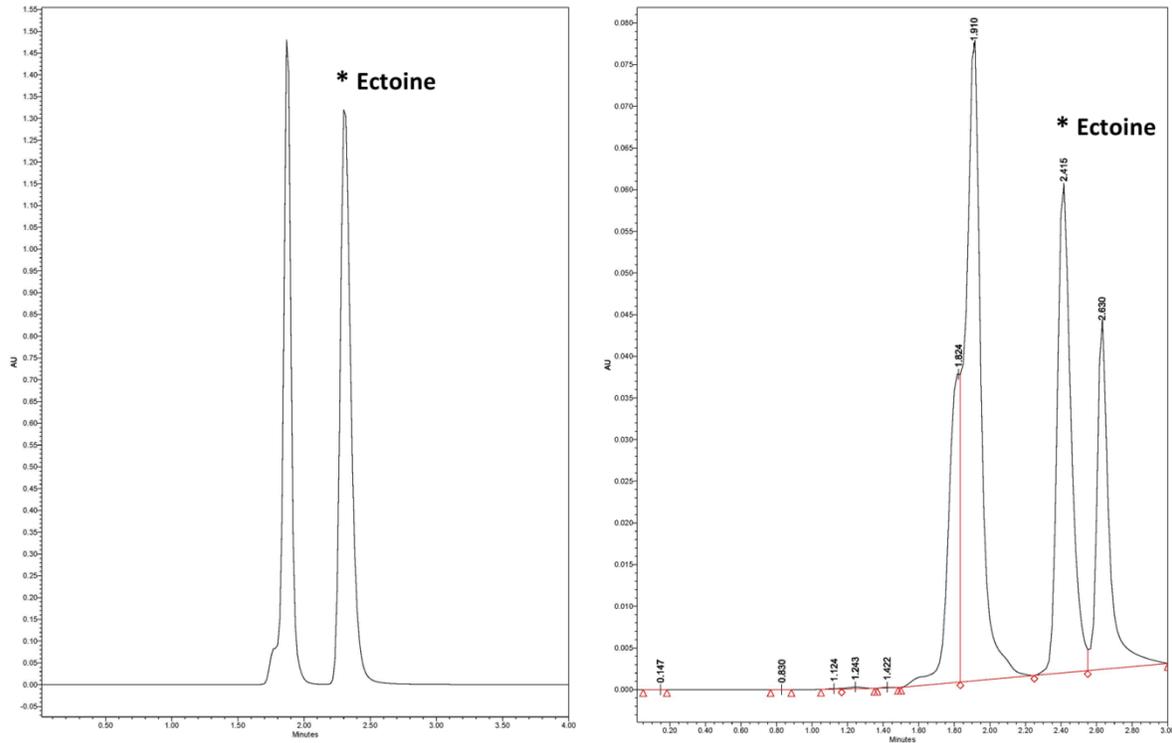
90 Aliquots of pure CH₄ were supplied to the headspace of the reactors in TS1 using a 100 mL
91 gas tight syringe. Copper and NaCl were supplied by addition of the corresponding amount
92 of salt to the cultivation broths in TS2 and TS3, respectively, while the different
93 temperatures used in TS4 (25, 30 or 35° C) were maintained using thermostatic baths
94 (Digiterm-S-150 20).

95 The O₂, CO₂ and CH₄ headspace concentrations were daily monitored. Aliquots of 10 mL
96 from the cultivation broth were also daily drawn with a liquid syringe to determine biomass
97 concentration and the intra and extra-cellular ectoine concentration. Biomass concentration
98 was estimated via culture absorbance measurements at 650 nm, which were previously

99 correlated to dry biomass concentrations (g L^{-1}) determined as total suspended solids (TSS)
100 concentration.

101 *2.4 Analytical procedures*

102 The intra-cellular ectoine concentration was determined using 2 mL of cultivation broth
103 centrifuged at 9000 g and 4 °C for 15 min. Then, 2 mL of 80 % ethanol and 25 ± 5 mg of
104 0.1-mm-diameter zirconia/silica beads (BioSpec, Spain) were added to the Eppendorf tube
105 containing the pellet. Microbial cells were then disrupted in a Mini-BeadBeater-16
106 (BioSpec, Spain) at 1048 g for 1 min and the suspension was kept overnight at room
107 temperature (modified from Lang et al., 2011). The supernatant of these suspensions was
108 used for ectoine analysis prior centrifugation at 9000 g and 4 °C for 15 min and filtration
109 through 0.22 μm filters (Filter-lab, Barcelona). The specific intra-cellular ectoine
110 concentration ($\text{g ectoine g biomass}^{-1}$) was calculated using the TSS concentration (g L^{-1}) of
111 the corresponding cultivation broth. An aliquot of 1 mL of cultivation broth was also drawn
112 and filtered through 0.22 μm filters (Filter-lab, Barcelona) to measure the extra-cellular
113 ectoine concentration. The concentration of ectoine was measured by HPLC-UV in a HPLC
114 717 plus auto-sampler (Waters, Bellefonte, USA) coupled with a UV Dual λ Absorbance
115 detector (Waters, Bellefonte, USA detector) at 210 nm using a LC-18 AQ + C Supelcosil
116 column (Waters, Bellefonte, EEUU) and a C18 AQ + pre-column (Waters, Bellefonte,
117 EEUU). A phosphate buffer, consisting of 0.8 mM K_2HPO_4 and 6.0 mM Na_2HPO_4 , was
118 used as a mobile phase at 25 °C and a flow rate of 1 mL min^{-1} (Tanimura et al., 2013).
119 Ectoine quantification was carried out using external standards of commercially available
120 ectoine ((S)-b-2-methyl-1,4,5,6-tetrahydro-pyrimidine-4-carboxylic acid, purity 95 %,
121 Sigma Aldrich, Spain) (Figure 1).



123

124 **Figure 1.** HPLC chromatograms a) Standard of ectoine at 100 mg L⁻¹ in MSM b) ethanol
 125 extracts of *Methylobacterium alcaliphilum* 20Z cultivated at 3% NaCl, 25 °C, 0.05 μM
 126 Cu²⁺ and 4 % CH₄ in MSM.

127

128 CH₄, O₂ and CO₂ gas concentrations were determined in a Bruker 430 GC-TCD (Palo Alto,
 129 USA) equipped with a CP-Molsieve 5A (15 m × 0.53 μm × 15 μm) and a CP-PoraBOND
 130 Q (25 m × 0.53 μm × 10 μm) column. The oven, injector and detector temperatures were
 131 maintained at 45 °C, 150 °C and 200 °C, respectively. Helium was used as the carrier gas at
 132 13.7 mL min⁻¹.

133 Culture absorbance measurements at 650 nm were conducted using a Shimadzu UV-2550
 134 UV/Vis spectrophotometer (Shimadzu, Japan). TSS concentration was measured according
 135 to standard methods (American Water Works Association, 2012).

136 2.5. Data analysis

137 The specific CH₄ degradation rate (SDR, g CH₄ g⁻¹_{biomass} h⁻¹) was calculated from the slope
138 of the time course plot of methane concentration within the linear range in the batch
139 cultivation tests carried out. The statistical data analysis was performed using SPSS 20.0
140 (IBM, USA). The results are given as the average ± standard deviation. The homogeneity
141 of the variance of the parameters was evaluated using a Levene test. Significant differences
142 were analysed by ANOVA and post-hoc analysis for multiple group comparisons.
143 Differences were considered to be significant at $p \leq 0.05$.

144

145 3. Results

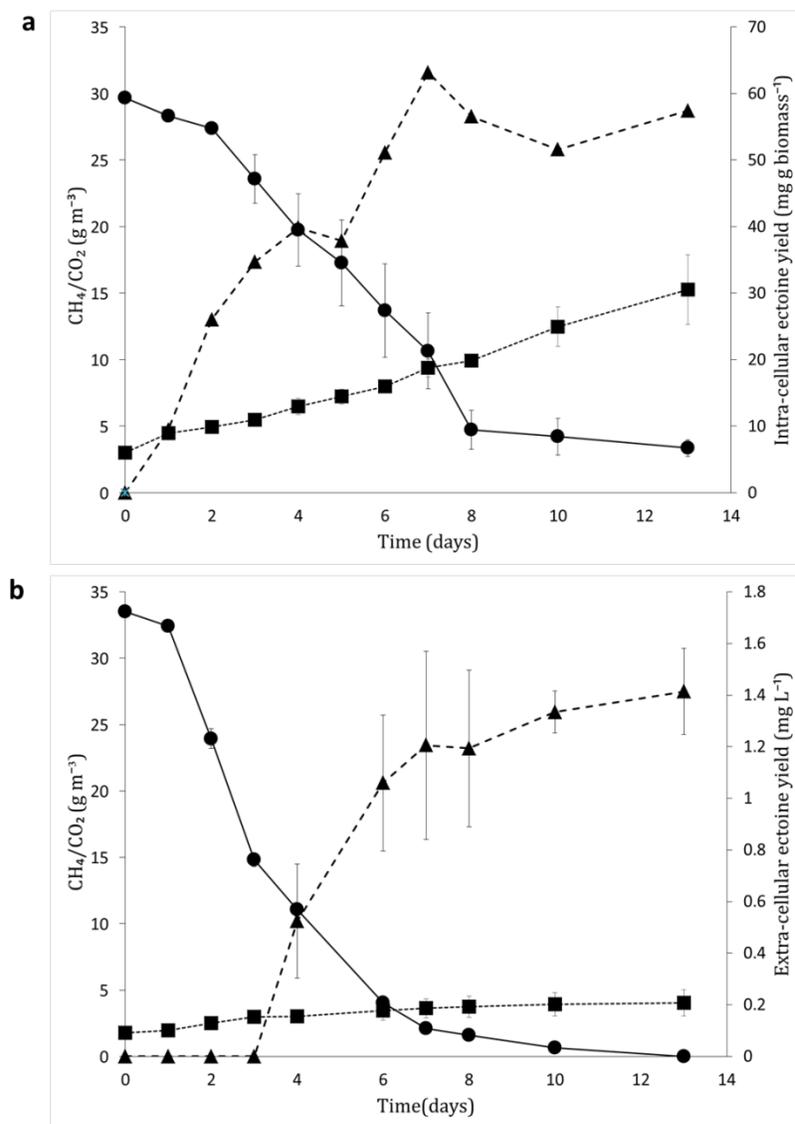
146 3.1. Influence of cultivation conditions on intra-cellular ectoine production

147 Intra-cellular ectoine reached its maximum concentration between days 5 and 7 of
148 cultivation regardless of the conditions tested. Subsequently, the concentration of intra-
149 cellular ectoine remained constant until the end of the assay (Figure 2a). No significant
150 difference was recorded in the intra-cellular ectoine concentration at 4 % CH₄, 3 % NaCl,
151 0.05 μM Cu²⁺ and 25 °C in TS1-TS4, which confirmed the reproducibility and consistency
152 of the results here obtained. Both CH₄ and NaCl concentrations had a significant influence
153 on the production of intra-cellular ectoine (Figure 3). A CH₄ concentration of 20 %
154 supported maximum specific yields of 31.0 ± 1.7 mg ectoine g biomass⁻¹ by the end of the
155 cultivation, while the maximum yields obtained at CH₄ concentrations of 2 and 4 % were
156 9.9 ± 0.6 and 13.6 ± 3.8 mg ectoine g biomass⁻¹, respectively. A NaCl concentration of 6 %
157 was identified as the optimum value for the accumulation of intra-cellular ectoine, which

158 reached a maximum concentration of 66.9 ± 4.2 mg ectoine g biomass⁻¹. Higher or lower
 159 salt concentrations supported lower ectoine yields (30.4 ± 7.5 mg ectoine g biomass⁻¹ at 9
 160 % NaCl; 12.5 ± 3.9 mg ectoine g biomass⁻¹ at 3 % NaCl; 1.2 ± 0.5 mg ectoine g biomass⁻¹ at
 161 0 % NaCl). On the contrary, no significant effect ($p < 0.05$) of temperature or Cu²⁺
 162 concentration was observed on the production of intra-cellular ectoine (Figure 3).

163

<Figure 2>



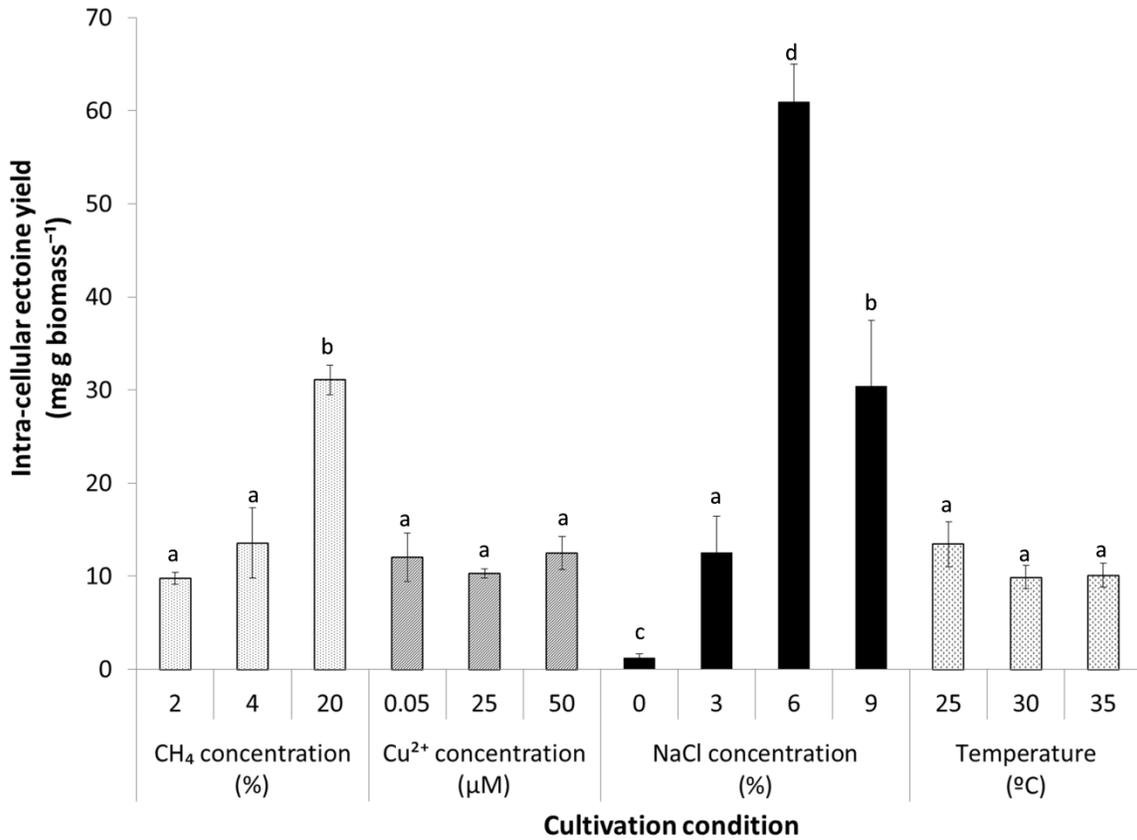
164

165 **Figure 2.** Time course of the concentration of CH₄ (●, continuous line), CO₂ (■, dotted
 166 line) and intra-cellular (a) or extra-cellular (b) ectoine (▲, dashed line) at a) during
 167 *Methylobacterium alcaliphilum* 20Z cultivation at 6 % NaCl, 0.05 μM Cu²⁺, 25 °C and 4 %
 168 CH₄, and b) at 3 % NaCl, 50 μM Cu²⁺, 25 °C and 4 % CH₄.

169

170

<Figure 3>



171

172 **Figure 3.** Maximum intra-cellular ectoine yield under different cultivation conditions.
 173 Vertical lines represent standard deviations from replicates. Columns inter/intra-groups
 174 with different letters were significantly different at p<0.05.

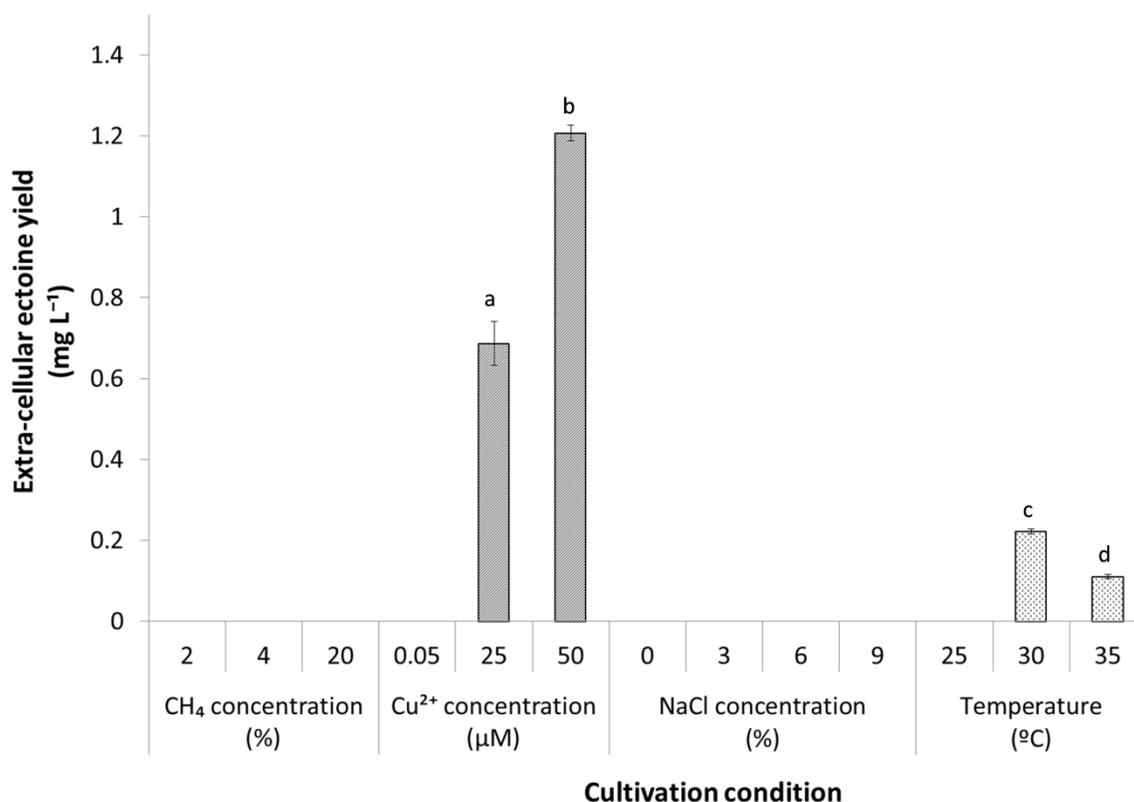
175 The maximum specific intra-cellular ectoine yields recorded at the different Cu²⁺
 176 concentrations tested were 12.0 ± 2.6 mg ectoine g biomass⁻¹ at 0.05 μM Cu²⁺, 10.3 ± 0.5
 177 mg ectoine g biomass⁻¹ at 25 μM Cu²⁺ and 12.4 ± 0.7 mg ectoine g biomass⁻¹ at 50 μM

178 Cu^{2+} . Similarly, no significant effect of temperature on ectoine accumulation was observed
179 within the tested T range, with an average yield of 11.70 ± 1.1 mg ectoine g biomass⁻¹.

180 3.2. Influence of cultivation conditions on ectoine excretion

181 While no extra-cellular ectoine was detected during the 13 days of cultivation under the
182 different concentrations of CH_4 and NaCl tested, ectoine excretion was detected at high
183 temperature and Cu^{2+} concentration (Figure 4).

184 <Figure 4>



185

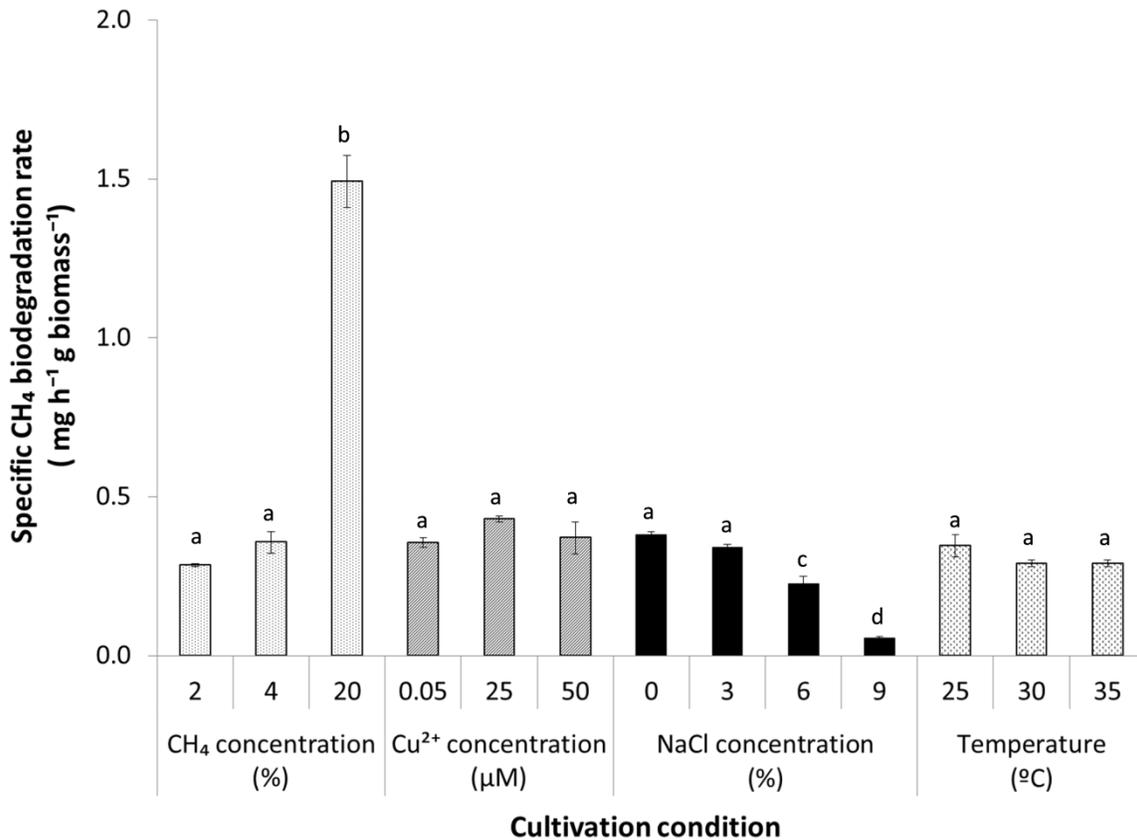
186 **Figure 4.** Extra-cellular ectoine excreted under different cultivation conditions. Vertical
187 lines represent standard deviations from replicates. Columns inter/intra-groups with
188 different letters were significantly different at $p < 0.05$.

189 Ectoine excretion was observed by day 4 in tests supplemented with high Cu^{2+}
190 concentrations (Figure 2b). The maximum concentrations recorded were 0.7 ± 0.05 and 1.2
191 ± 0.01 mg extra-cellular ectoine L^{-1} at 25 and 50 μM of Cu^{2+} , respectively. Excretion of
192 ectoine was also observed at 30 and 35 °C, although lower maximum concentrations were
193 detected under these particular cultivation conditions (0.2 ± 0.005 mg extra-cellular ectoine
194 L^{-1} at 30°C and 0.1 ± 0.005 mg extra-cellular ectoine L^{-1} at 35°C).

195 *3.3. Influence of cultivation conditions on the specific CH_4 degradation rate*

196 The results showed that a CH_4 headspace concentration of 20 % supported a significantly
197 ($p < 0.05$) higher SDR (1.50 ± 0.08 g CH_4 h^{-1} g biomass $^{-1}$) compared to the SDRs recorded at
198 4 and 2 % of CH_4 (0.33 ± 0.05 and 0.29 ± 0.03 g CH_4 h^{-1} g biomass $^{-1}$, respectively) (Figure
199 5). On the contrary, the specific CH_4 oxidation rates decreased at higher NaCl
200 concentrations, with CH_4 SDRs of 0.05 ± 0.005 , 0.22 ± 0.02 , 0.34 ± 0.02 and 0.38 ± 0.02 g
201 CH_4 h^{-1} g biomass $^{-1}$ at 9, 6, 3 and 0 % NaCl, respectively. Neither temperature (25, 30 and
202 35 °C supported SDRs of 0.34 ± 0.04 , 0.30 ± 0.01 and 0.30 ± 0.01 g CH_4 h^{-1} g biomass $^{-1}$,
203 respectively) nor Cu^{2+} concentration (0.05, 25 and 50 μM of Cu^{2+} supported SDRs of 0.35
204 ± 0.01 , 0.43 ± 0.01 and 0.37 ± 0.05 g CH_4 h^{-1} g biomass $^{-1}$, respectively) showed a
205 significant effect on the specific CH_4 degradation rate.

206 <Figure 5>



207

208 **Figure 5.** Specific CH₄ biodegradation rate under different cultivation conditions. Vertical
 209 lines represent standard deviations from replicates. Columns inter/intra-groups with
 210 different letters were significantly different at p<0.05.

211

212 3.4. Production of extra and intra-cellular ectoine under optimum cultivation conditions

213 A final study was carried out combining the optimum parameters from previous tests TS1-
 214 TS4 in order to determine the production of extra and intra-cellular ectoine (20 % CH₄, 6 %
 215 NaCl, 30 °C, 50 µM Cu²⁺) (Table 2). These cultivation conditions promoted the excretion
 216 of ectoine to the extra-cellular medium (4.7 mg L⁻¹, which would correspond to 33.3 mg

217 ectoine g biomass⁻¹) and resulted in a high production of intra-cellular ectoine (40.7 ± 0.02
 218 mg ectoine g biomass⁻¹).

219 <Table 2>

Table 2. Maximum values of ectoine concentration during *Methylobacterium* *alcaliphilum* 20Z batch cultivation tests.

Test	Maximum intra-cellular ectoine [Ectoine] (mg g biomass ⁻¹)	Maximum extra-cellular ectoine [Ectoine] (mg L ⁻¹)
TS1		
20% CH ₄ , 25°C, 0.05μM Cu ²⁺ , 3% NaCl	31.0 ± 1.7	N/D
TS2		
4% CH ₄ , 25°C, 0.05μM Cu ²⁺ , 6% NaCl	66.9 ± 4.2	N/D
TS3		
4% CH ₄ , 25°C, 50μM Cu ²⁺ , 3% NaCl	12.4 ± 0.7	1.2 ± 0.01
TS4		
4% CH ₄ , 30°C, 0.05μM Cu ²⁺ , 3% NaCl	10.6 ± 0.15	0.2 ± 0.005
TS5		
20 % CH ₄ , 30°C, 50μM Cu ²⁺ , 6% NaCl	40.7 ± 0.02	4.7 ± 0.05

220

221

222 4. Discussion

223 This research investigated for the first time the effect of 4 environmental parameters (i.e.
 224 CH₄ concentration, Cu²⁺ concentration, temperature and NaCl concentration) on the

225 production and excretion of ectoine and on the specific CH₄ degradation by
226 *Methylobacterium alcaliphilum* 20 Z in order to elucidate the optimum operational
227 conditions to maximize ectoine production during the abatement of dilute CH₄ emissions.

228 A high concentration of CH₄ in the GHG emission significantly enhanced the production of
229 intra-cellular ectoine likely due to an increase in substrate availability for the bacterial
230 community, which induced high growth rates and therefore a high metabolite production
231 (Estrada et al., 2014). In this sense, an increase in CH₄ concentration from 4 to 20 %
232 enhanced the production of intra-cellular ectoine by a factor of 2.7 (from 9.9 ± 0.6 to 30.4 ±
233 7.5 mg ectoine g biomass⁻¹). However, no significant effect was observed on the
234 accumulation of ectoine within the lower range of CH₄ concentrations tested (2 and 4 %).
235 The key role of CH₄ concentration in the intra-cellular ectoine accumulation is in
236 agreement with the results observed by Khmelena et al. (2000), who recorded a
237 maximum ectoine concentration of 200 mg ectoine g biomass⁻¹ in *Methylobacterium*
238 *alcaliphilum* 20 Z under a CH₄ concentration of 50 % (v/v) in a MSM with 6 % of NaCl,
239 0.05 μM Cu²⁺ at 29 °C (Khmelenina et al., 2000; Trotsenko et al., 2005). However, only
240 dilute CH₄ emissions (<20 %) not suitable for energy recovery can be considered as a
241 substrate of this novel CH₄ bio-refinery.

242 The salinity of the cultivation medium exhibited a positive effect on the intra-cellular
243 ectoine yield up to a concentration of 6 % NaCl (66.9 ± 4.2 mg ectoine g biomass⁻¹),
244 whereas higher salt concentrations resulted in lower ectoine yields (30.4 ± 7.5 mg ectoine g
245 biomass⁻¹ at 9 % of NaCl). Some authors have also observed a decrease in ectoine
246 accumulation at increasing external salinity due to the regulation of the ectoine biosynthesis
247 at the enzyme activity level (Reshetnikov et al., 2005). It is noteworthy that the

248 concentration of ectoine herein obtained at 6 % NaCl does not differ much from the values
249 commonly encountered during the industrial production of ectoine using the glucose
250 fermentative microorganism *Halomonas elongate* (yielding ectoine at an average value of
251 150.5 mg ectoine g biomass⁻¹ when reused 9 times) (Strong et al., 2015). Nowadays, the
252 process implemented at industrial scale (bacterial milking) involves the cyclic increase and
253 decrease of the salt concentration in the cultivation broth up to 12 % NaCl. This process
254 involving salt shocks increases reactor corrosion and hinders the downstream processing of
255 ectoine due to the discontinuous nature of the production procedure and the high
256 concentrations of salt. Alternatively, *Methylobacterium alcaliphilum* 20Z can continuously
257 synthesize a comparable high yield of ectoine in a less harsh medium coupled with CH₄
258 abatement from dilute emissions. Surprisingly, ectoine production was also observed in the
259 absence of NaCl, although the concentrations detected were 55 times lower than the
260 maximum ectoine yield recorded at 6 % of NaCl. The presence of a basal activity of the
261 specific enzymes responsible for ectoine biosynthesis was likely related to the constitutive
262 transcription of the ectoine gene cluster by this strain as previously observed by
263 Khmelenina et al. (2000) and Reshetnikov et al. (2006). Likewise, Reshetnikov et al. (2006)
264 confirmed that the optimum temperature for the enzymes catalyzing the key reactions of
265 ectoine biosynthesis in *Methylobacterium alcaliphilum* 20Z was 20 °C, while temperatures
266 higher than 30 °C inhibited this reaction. However, no pernicious effect of temperature on
267 the production of intra-cellular ectoine was observed in the present study. Moreover, no
268 significant differences occurred among the ectoine yields recorded in the cultivation media
269 containing different Cu²⁺ concentrations.

270 Ectoine excreting bacterial strains can accumulate ectoine and excrete it into the cultivation
271 medium, thus enhancing the economics of the industrial production process of ectoine
272 (Lang et al., 2011). Up to date, *Methylobacterium alcaliphilum* 20Z has never been
273 described as a strain able to excrete ectoine to the extra-cellular medium (Trotsenko et al.,
274 2005). Whereas no extra-cellular ectoine was detected at the tested concentrations of NaCl
275 and CH₄, cultivation at high temperatures and high Cu²⁺ concentrations promoted the
276 excretion of a significant fraction of the intra-cellular ectoine. This excretion could be
277 associated to the activation of unspecific channels or specific transporters able to excrete
278 ectoine as a result of passive diffusion or an active transport of the cation Cu²⁺. A research
279 carried out by Schubert et al. (2007) observed that a transgenic *E.coli* genetically modified
280 with the genes for ectoine biosynthesis, *ectABC*, from the halophilic bacterium
281 *Chromohalobacter salexigens*, was able to excrete ectoine via expression of a specific
282 transporter (Schubert et al., 2007). In this context, the presence of high Cu²⁺ concentrations
283 in the cultivation medium could affect the activation of some specific transporters. Gram
284 negative bacteria can express the tubular trans-membrane proteins porins, which allow the
285 diffusion of small solutes and constitute a likely route for the simultaneous uptake of
286 unchelated Cu²⁺ and excretion of intra-cellular ectoine in methanotrophs (Balasubramanian
287 et al., 2011). Another plausible explanation for the excretion of intra-cellular ectoine might
288 be the increase in cell membrane permeability, which is affected by temperature as
289 observed by Kropinski et al. (1987) .

290 Along with the optimization of ectoine production, the maintenance of an efficient CH₄
291 abatement from dilute emissions of this GHG is also of key importance in the context of
292 climate change mitigation. Hence, the highest SDRs were obtained at a CH₄ concentration

293 of 20 % as a result of the enhanced gas-liquid concentration gradients, which likely induced
294 higher aqueous CH₄ concentrations and therefore higher specific growth rates (Cantera et
295 al., 2016). However, no SDR enhancement was observed in the low range of CH₄
296 concentrations (2 and 4 %), as previously reported by Cantera et al. (2016). High salt
297 concentrations negatively affected methane biodegradation in *Methylobacterium*
298 *alcaliphilum* 20Z, the SDR obtained at 9 % NaCl being 7.7 times lower than that obtained
299 at 3 % NaCl. This strain is an halotolerant alkaliphilic methanotroph (Khmelenina et al.,
300 1997) that can tolerate higher salt concentrations than other methanotrophs, but highly
301 saline environments are not its optimum habitat. Thus, despite higher NaCl concentrations
302 mediated the highest ectoine yields (6 and 9 %), a reduced CH₄ abatement performance was
303 recorded. On the other hand, temperature and Cu²⁺ did not affect the SDR.

304 Finally, the optimum conditions (30°C, 50 µM Cu²⁺, 20 % of CH₄ and 6 % of NaCl) were
305 combined in a test in order to maximize both extra and intra-cellular ectoine production. In
306 this particular assay, the production of extra-cellular ectoine was 4 times higher than that
307 obtained in TS2 (25°C, 50 µM Cu²⁺, 4 % of CH₄ and 3 % of NaCl), while the intra-cellular
308 ectoine yield was lower than that recorded in TS3 (6 % of NaCl, 0.05 µM Cu²⁺ and 4 % of
309 CH₄). Thus, the combination of these parameters favored the excretion of 44.4 % of the
310 total intra-cellular ectoine produced. The total (intra-cellular + extra-cellular) ectoine
311 produced would account for 73 mg ectoine g biomass⁻¹ if not excreted, which was similar to
312 the intracellular ectoine concentration (66.9 ± 4.2 mg g biomass⁻¹) in TS3 at 6 % NaCl.

313 In summary, a proper selection of the environmental parameters (temperature, Cu²⁺, NaCl
314 and CH₄ concentration) for *Methylobacterium alcaliphilum* 20Z cultivation is crucial to
315 simultaneously maximize both the intra-cellular production and excretion of ectoine and

316 CH₄ abatement. The promising results here obtained support the further development of
317 CH₄ biorefineries capable of creating value out of GHG mitigation using extremophilic
318 methanotrophs.

319

320 **5. Acknowledgements**

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325

326 **6. References**

327 American Water Works Association, 2012. Standard Methods for the Examination of Water and Wastewater,
328 American Water Works Association/American Public Works Association/Water Environment
329 Federation.

330 Avalos Ramirez, A., García-Aguilar, B.P., Jones, J.P., Heitz, M., 2012. Improvement of methane biofiltration
331 by the addition of non-ionic surfactants to biofilters packed with inert materials. *Process Biochem.* 47,
332 76–82.

333 Balasubramanian, R., Kenney, G.E., Rosenzweig, A.C., 2011. Dual pathways for copper uptake by
334 methanotrophic bacteria. *J. Biol. Chem.* 286, 37313–37319. doi:10.1074/jbc.M111.284984

335 Cantera, S., Lebrero, R., García-encina, P.A., Mu, R., 2016. Evaluation of the influence of methane and
336 copper concentration and methane mass transport on the community structure and biodegradation
337 kinetics of methanotrophic cultures 171, 11–20. doi:10.1016/j.jenvman.2016.02.002

338 Estrada, J.M., Lebrero, R., Quijano, G., Pérez, R., Figueroa-González, I., García-Encina, P.A., Muñoz, R.,

339 2014. Methane abatement in a gas-recycling biotrickling filter: Evaluating innovative operational
340 strategies to overcome mass transfer limitations. *Chem. Eng. J.* 253, 385–393.
341 doi:10.1016/j.cej.2014.05.053

342 European Environmental Agency, 2015. Atmospheric greenhouse gas concentrations (CSI 013/CLIM 052) -
343 Assessment published Feb 2015. [WWW Document]. [http://www.eea.europa.eu/data-and-](http://www.eea.europa.eu/data-and-maps/indicators/atmospheric-greenhouse-gas-concentrations-4/assessment)
344 [maps/indicators/atmospheric-greenhouse-gas-concentrations-4/assessment](http://www.eea.europa.eu/data-and-maps/indicators/atmospheric-greenhouse-gas-concentrations-4/assessment).

345 Kaluzhnaya, M., Khmelenina, V., Eshinimaev, B., Suzina, N., Nikitin, D., Solonin, a, Lin, J.L., McDonald,
346 I., Murrell, C., Trotsenko, Y., 2001. Taxonomic characterization of new alkaliphilic and alkalitolerant
347 methanotrophs from soda lakes of the Southeastern Transbaikal region and description of
348 *Methylomicrobium buryatense* sp.nov. *Syst. Appl. Microbiol.* 24, 166–176. doi:10.1078/0723-2020-
349 00028

350 Kalyuzhnaya, M.G., Khmelenina, V., Eshinimaev, B., Sorokin, D., Fuse, H., Lidstrom, M., Trotsenko, Y.,
351 2008. Classification of halo(alkali)philic and halo(alkali)tolerant methanotrophs provisionally assigned
352 to the genera *Methylomicrobium* and *Methylobacter* and emended description of the genus
353 *Methylomicrobium*. *Int. J. Syst. Evol. Microbiol.* 58, 591–596. doi:10.1099/ijs.0.65317-0

354 Khmelenina, V.N., Kalyuzhnaya, M.G., Sakharovsky, V.G., Snzina, N.E., Trotsenko, Y. a., Gottschalk, G.,
355 1999. Osmoadaptation in halophilic and alkaliphilic methanotrophs. *Arch. Microbiol.* 172, 321–329.
356 doi:10.1007/s002030050786

357 Khmelenina, V.N., Kalyuzhnaya, M.G., Starostina, N.G., Suzina, N.E., Trotsenko, Y. a., 1997. Isolation and
358 characterization of halotolerant alkaliphilic methanotrophic bacteria from Tuva soda lakes. *Curr.*
359 *Microbiol.* 35, 257–261. doi:10.1007/s002849900249

360 Khmelenina, V.N., Sakharovskii, V.G., Reshetnikov, a S., Trotsenko, I. a, 2000. Synthesis of osmoprotectors
361 by halophilic and alkalophilic methanotrophs. *Mikrobiologiya* 69, 465–470.

362 Kropinski, A.M.B., Lewis, V., Berry, D., 1987. Effect of growth temperature on the lipids, outer membrane
363 proteins, and lipopolysaccharides of *Pseudomonas aeruginosa* PAO. *J. Bacteriol.* 169, 1960–1966.

364 Lang, Y. jun, Bai, L., Ren, Y. nan, Zhang, L. hua, Nagata, S., 2011. Production of ectoine through a combined

365 process that uses both growing and resting cells of *Halomonas salina* DSM 5928T. *Extremophiles* 15,
366 303–310. doi:10.1007/s00792-011-0360-9

367 Pastor, J.M., Salvador, M., Argandoña, M., Bernal, V., Reina-Bueno, M., Csonka, L.N., Iborra, J.L., Vargas,
368 C., Nieto, J.J., Cánovas, M., 2010. Ectoines in cell stress protection: Uses and biotechnological
369 production. *Biotechnol. Adv.* 28, 782–801. doi:10.1016/j.biotechadv.2010.06.005

370 Reshetnikov, A.S., Khmelenina, V.N., Trotsenko, Y.A., 2006. Characterization of the ectoine biosynthesis
371 genes of haloalkalotolerant obligate methanotroph “*Methylobacterium alcaliphilum* 20Z.” *Arch.*
372 *Microbiol.* 184, 286–297. doi:10.1007/s00203-005-0042-z

373 Reshetnikov, A.S., Mustakhimov, I.I., Khmelenina, V.N., Trotsenko, Y.A., 2005. Cloning, purification, and
374 characterization of diaminobutyrate acetyltransferase from the halotolerant methanotroph
375 *Methylobacterium alcaliphilum* 20Z. *Biochem.* 70, 878–883. doi:10.1007/s10541-005-0197-x

376 Schubert, T., Maskow, T., Benndorf, D., Harms, H., Breuer, U., 2007. Continuous synthesis and excretion of
377 the compatible solute ectoine by a transgenic, nonhalophilic bacterium. *Appl. Environ. Microbiol.* 73,
378 3343–3347. doi:10.1128/AEM.02482-06

379 Strong, P., Xie, S., Clarke, W.P., 2015. Methane as a resource: can the methanotrophs add value? *Environ.*
380 *Sci. Technol.* 49, 4001–4018. doi:10.1021/es504242n

381 Tanimura, K., Nakayama, H., Tanaka, T., Kondo, A., 2013. Ectoine production from lignocellulosic biomass-
382 derived sugars by engineered *Halomonas elongata*. *Bioresour. Technol.* 142, 523–529.
383 doi:10.1016/j.biortech.2013.05.004

384 Trotsenko, Y.A., Doronina, N. V., Khmelenina, V.N., 2005. Biotechnological potential of aerobic
385 methylotrophic bacteria: A review of current state and future prospects. *Appl. Biochem. Microbiol.* 41,
386 433–441. doi:10.1007/s10438-005-0078-5

387 United States Environmental Protection Agency., 2013. Global Greenhouse Gas Emissions Data [WWW
388 Document]. United States Environ. Prot. Agency. URL
389 <http://www.epa.gov/climatechange/ghgemissions/gases/ch4.html>

390

Table 1.

Table 1. Cultivation conditions evaluated during *Methylobacterium alcaliphilum* 20Z batch cultivation tests.

Test series (TS)	Operating conditions			
	CH ₄ (%)	Cu ²⁺ (μM)	NaCl (%)	T (°C)
TS1	2, 4, 20	0.05	3	25
TS2	20	0.05, 25, 50	3	25
TS3	20	0.05	0, 3, 6, 9	25
TS4	20	0.05	3	25, 30, 35
TS5	20	50	6	30

Table 2.

Table 2. Maximum values of ectoine concentration during <i>Methylomicrobium alcaliphilum</i> 20Z batch cultivation tests.		
Test	Maximum intra-cellular ectoine	Maximum extra-cellular ectoine
	[Ectoine] (mg g biomass ⁻¹)	[Ectoine] (mg L ⁻¹)
TS1		
20% CH ₄ , 25°C, 0.05μM Cu ²⁺ , 3% NaCl	31.0 ± 1.7	N/D
TS2		
4% CH ₄ , 25°C, 0.05μM Cu ²⁺ , 6% NaCl	66.9 ± 4.2	N/D
TS3		
4% CH ₄ , 25°C, 50μM Cu ²⁺ , 3% NaCl	12.4 ± 0.7	1.2 ± 0.01
TS4		
4% CH ₄ , 30°C, 0.05μM Cu ²⁺ , 3% NaCl	10.6 ± 0.15	0.2 ± 0.005
TS5		
20 % CH ₄ , 30°C, 50μM Cu ²⁺ , 6% NaCl	40.7 ± 0.02	4.7 ± 0.05

Figure 1.

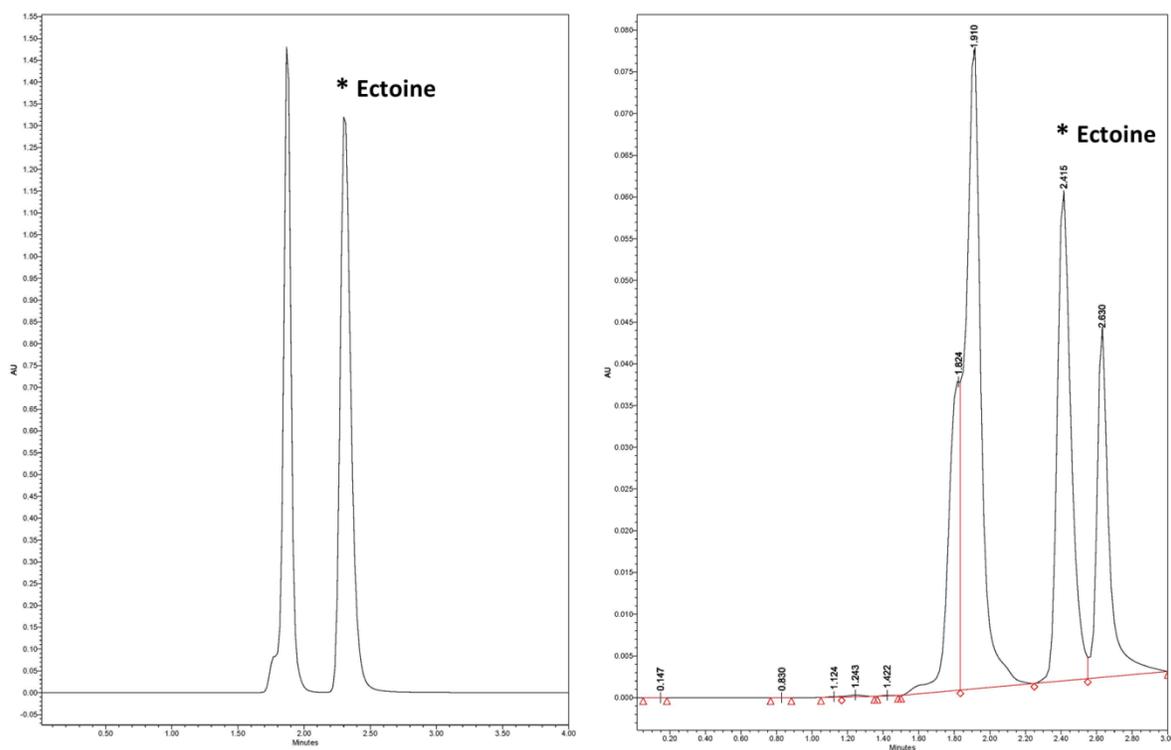


Figure 1. HPLC chromatograms a) Standard of ectoine at 100 mg L⁻¹ in MSM b) ethanol extracts of *Methylobacterium alcaliphilum* 20Z cultivated at 3% NaCl, 25 °C, 0.05 μM Cu²⁺ and 4 % CH₄ in MSM.

Figure 2.

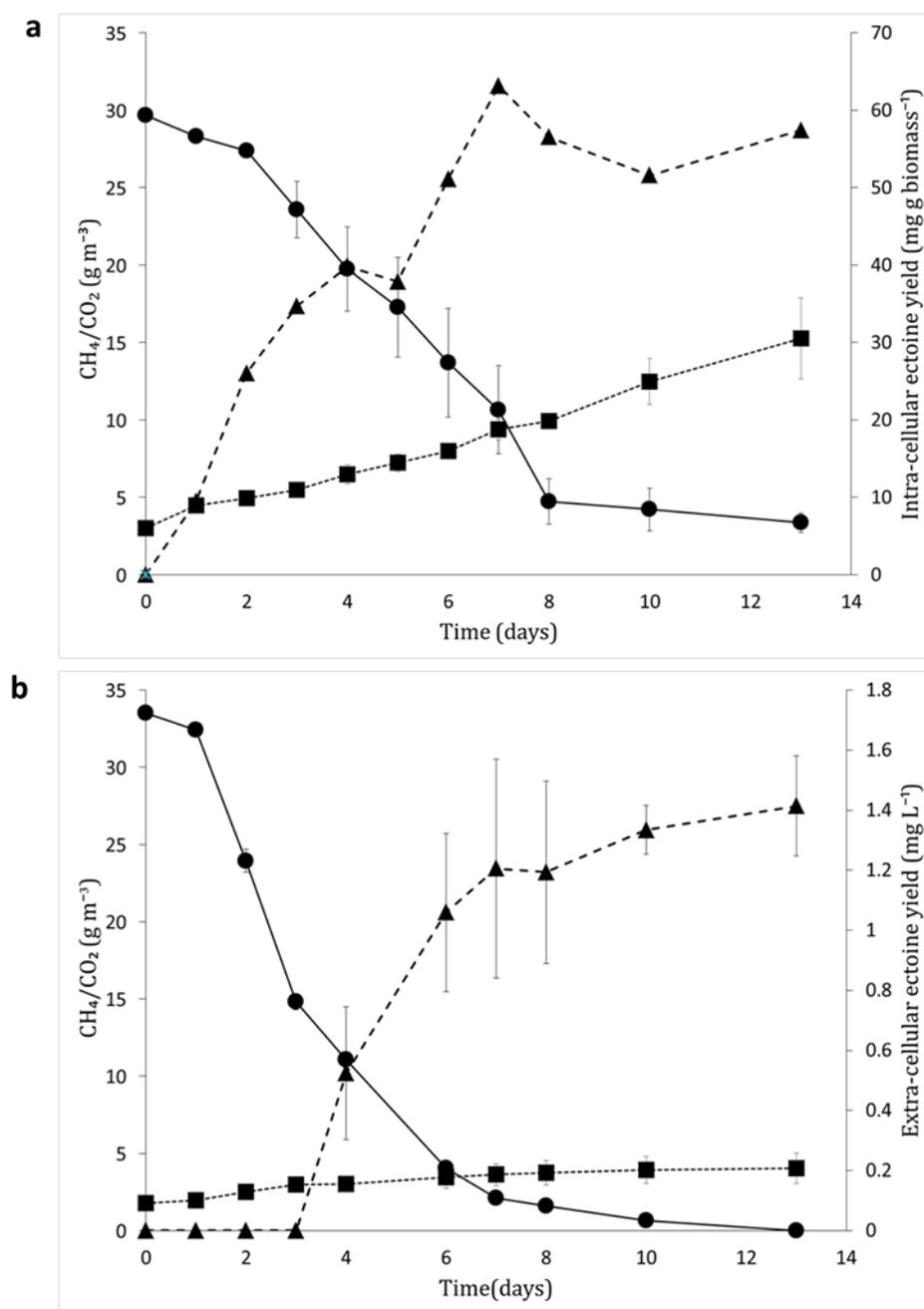


Figure 2. Time course of the concentration of CH₄ (●, continuous line), CO₂ (■, dotted line) and intra-cellular (a) or extra-cellular (b) ectoine (▲, dashed line) at a) during *Methylomicrobium alcaliphilum* 20Z cultivation at 6 % NaCl, 0.05 μM Cu²⁺, 25 °C and 4 % CH₄, and b) at 3 % NaCl, 50 μM Cu²⁺, 25 °C and 4 % CH₄.

Figure 3.

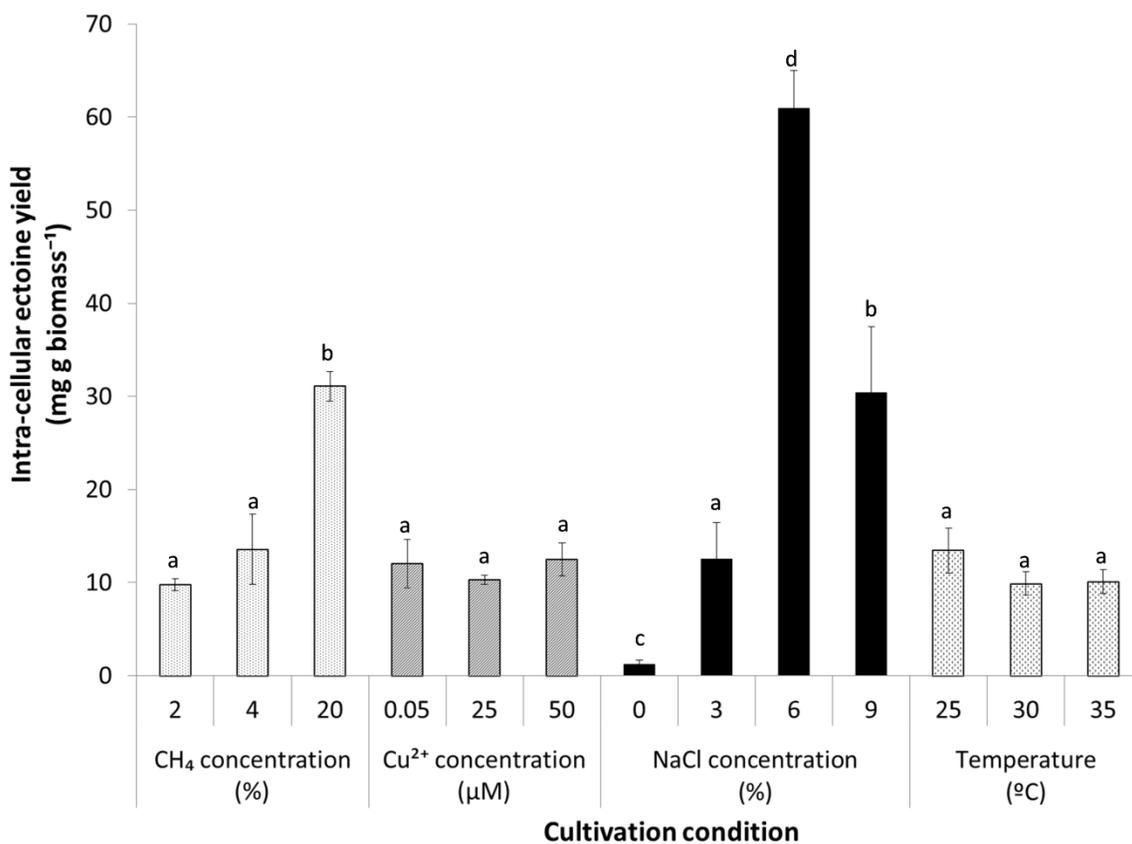


Figure 3. Maximum intra-cellular ectoine yield under different cultivation conditions. Vertical lines represent standard deviations from replicates. Columns inter/intra-groups with different letters were significantly different at $p < 0.05$.

Figure 4.

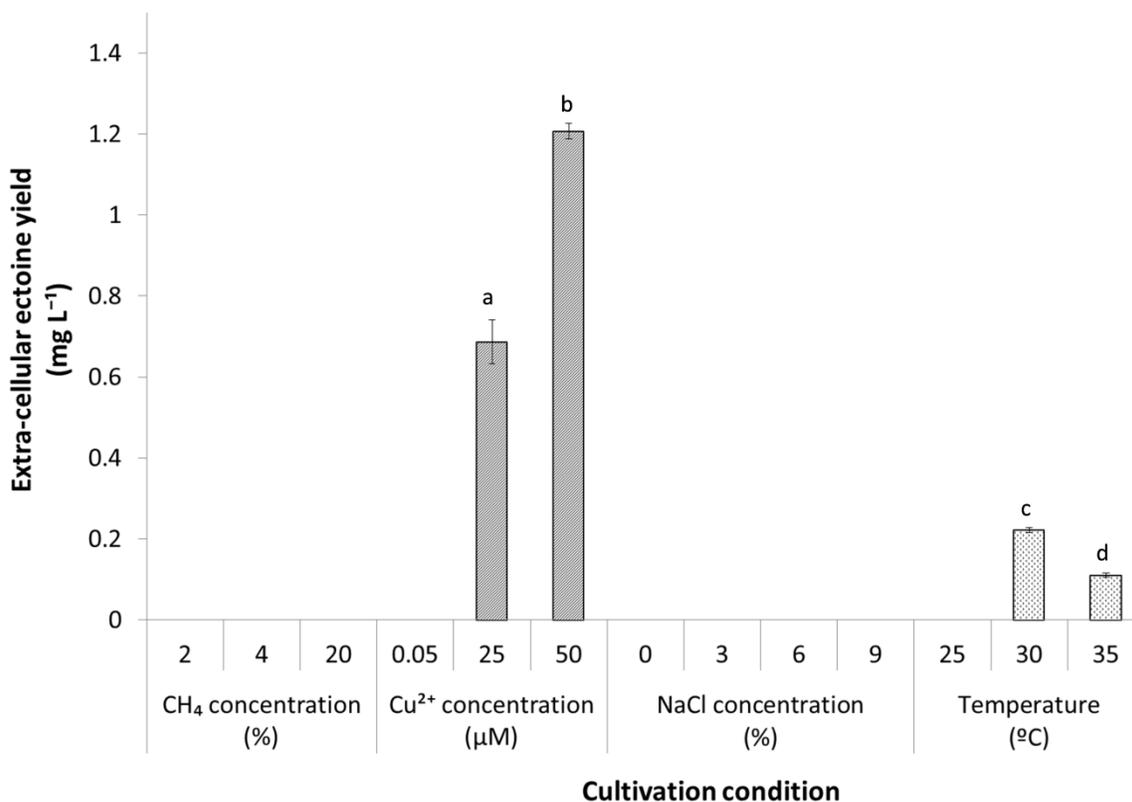


Figure 4. Extra-cellular ectoine excreted under different cultivation conditions. Vertical lines represent standard deviations from replicates. Columns inter/intra-groups with different letters were significantly different at $p < 0.05$.

Figure 5.

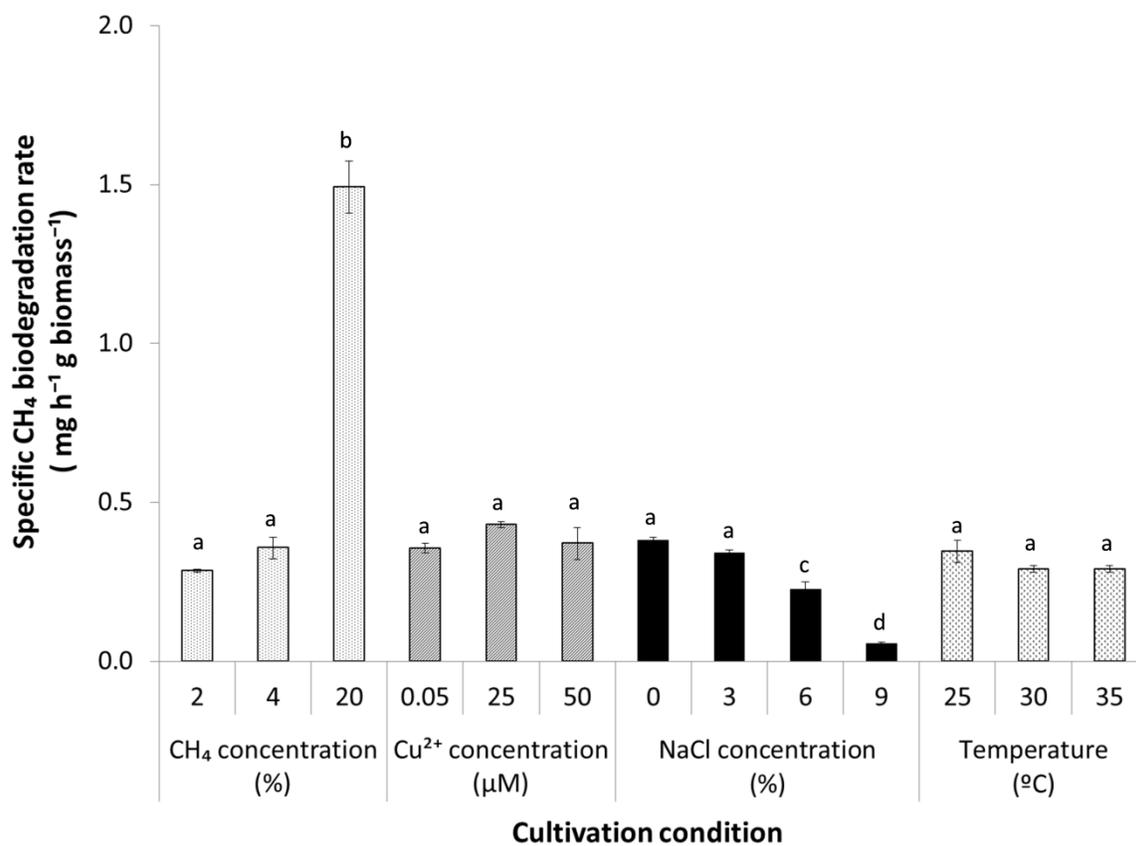


Figure 5. Specific CH₄ biodegradation rate under different cultivation conditions. Vertical lines represent standard deviations from replicates. Columns inter/intra-groups with different letters were significantly different at $p < 0.05$.