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4 **Long-term photosynthetic CO<sub>2</sub> removal from biogas and flue-gas: exploring the**  
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6 **potential of closed photobioreactors for high-value biomass production**  
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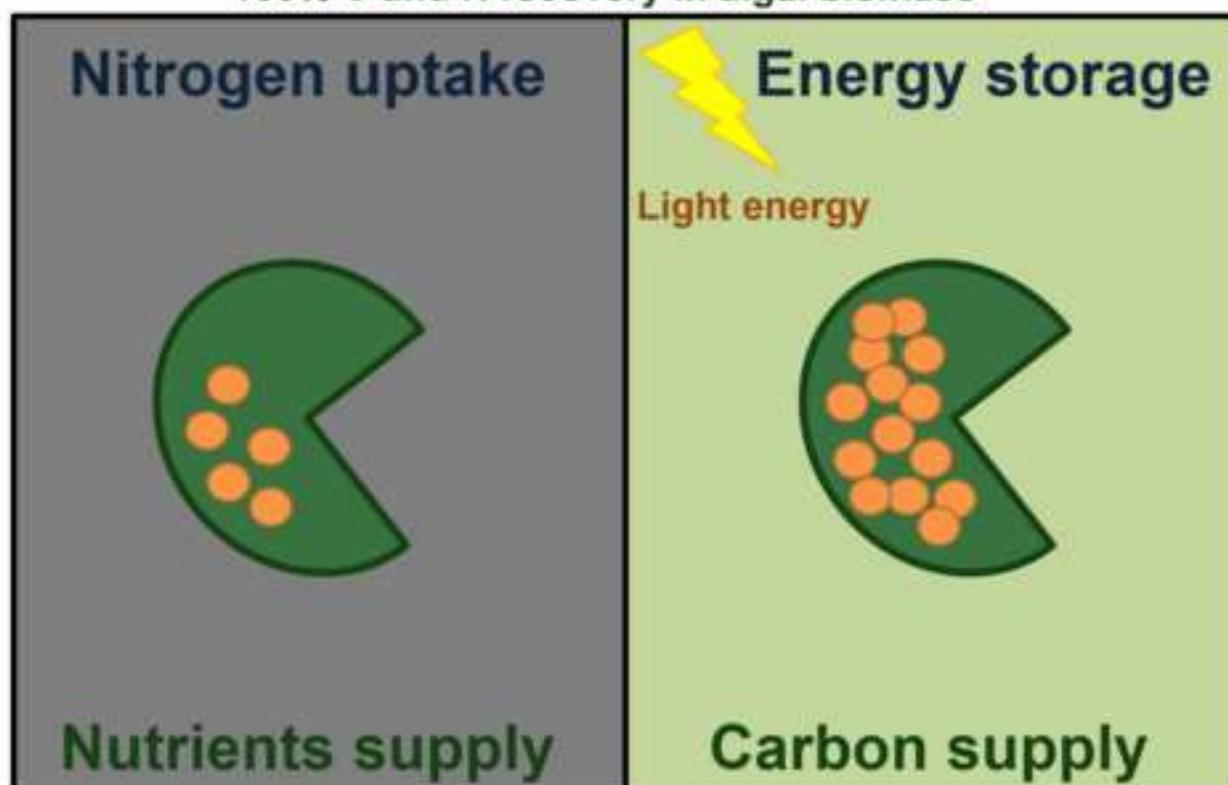
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**Continuous production of high-energy storage compounds**



100% C and N recovery in algal biomass



**Continuous CO<sub>2</sub> abatement from biogas and flue-gas**

## Highlights

- CO<sub>2</sub> abatement from biogas and flue-gas was studied in a tubular photobioreactor
- A feast-famine regime was applied for continuous production of high-energy storage compounds
- CO<sub>2</sub> removals > 98% and complete C and N recovery as biomass was achieved
- Microalgae consumed nitrogen in the dark period regardless of the N source
- The N-dark feeding strategy increased the carbohydrates productivity by 1.7 times



15 **Abstract**

16 The long-term performance of a tubular photobioreactor interconnected to a gas  
17 absorption column for the abatement of CO<sub>2</sub> from biogas and flue-gas was investigated.  
18 Additionally, a novel nitrogen feast-famine regime was implemented during the flue-gas  
19 feeding stage in order to promote the continuous storage of highly-energetic  
20 compounds. Results showed effective CO<sub>2</sub> (~98%) and H<sub>2</sub>S (~99%) removals from  
21 synthetic biogas, supported by the high photosynthetic activity of microalgae which  
22 resulted in an alkaline pH (~10). In addition, CO<sub>2</sub> removals of 99 and 91% were  
23 observed during the flue-gas operation depending on the nutrients source: mineral salt  
24 medium and digestate, respectively. A biomass productivity of ~8 g m<sup>-2</sup> d<sup>-1</sup> was  
25 obtained during both stages, with a complete nitrogen and carbon recovery from the  
26 cultivation broth. Moreover, the strategy of feeding nutrients during the dark period  
27 promoted the continuous accumulation of carbohydrates, their concentration increasing  
28 from 22% under normal nutrition up to 37% during the feast-famine cycle. This  
29 represents a productivity of ~3 g<sub>-carbohydrates</sub> m<sup>-2</sup> d<sup>-1</sup>, which can be further valorized to  
30 contribute to the economic sustainability of the photosynthetic CO<sub>2</sub> removal process.

31

32 **Keywords:** Algal-bacterial technology; Biogas upgrading; Carbohydrates production;  
33 CO<sub>2</sub> abatement; Photobioreactors.

34

## 35 **1. Introduction**

36 Carbon dioxide (CO<sub>2</sub>) represents nowadays the most important greenhouse gas (GHG),  
37 with ~77% of the total GHG emissions worldwide and an annual atmospheric  
38 concentration increase of 0.5% over the last decade (López *et al.*, 2014). In addition, the  
39 amount of CO<sub>2</sub> emitted from anthropogenic sources has increased from 22 Gt in 1990 to  
40 33 Gt in 2010, and it is expected to reach 41 Gt by 2030 (World Bank, 2014; United  
41 Nations, 2015). From these anthropogenic CO<sub>2</sub> emissions, ~93.5% are produced from  
42 the combustion of fossil fuels, with a typical concentration in the emitted gases ranging  
43 from 5 to 20% (Raesossadati *et al.*, 2014; Warmuzinski *et al.*, 2014). Energy  
44 production from biogas also constitutes an important source of anthropogenic CO<sub>2</sub>  
45 emissions (CO<sub>2</sub> content in raw biogas can vary from 15 up to 60%), which production  
46 in Europe is expected to reach 18-20 million m<sup>3</sup> by 2030 (Muñoz *et al.*, 2015). The  
47 detrimental effects of this GHG on the environment (*i.e.* global warming, modification  
48 of the pH of oceans, etc.) demand the implementation of cost-effective technologies for  
49 CO<sub>2</sub> removal from industrial emissions. In the particular case of biogas, the abatement  
50 of the CO<sub>2</sub> not only entails environmental benefits but also contributes to the upgrading  
51 of this biofuel, decreasing its transportation costs and increasing the energy content.  
52 Conventional physical/chemical technologies for CO<sub>2</sub> removal from flue-gas or biogas,  
53 such as scrubbing, adsorption, or cryogenic separation, have been widely implemented  
54 due to the extensive knowledge on their design and operation and the high removal  
55 efficiencies achieved. However, only biological technologies offer a low environmental  
56 impact, besides reducing the operating costs associated to the treatment process. In this  
57 regard, CO<sub>2</sub>-capturing biotechnologies supported by the photosynthetic activity of  
58 microalgae in photobioreactors allow for the removal of CO<sub>2</sub> in a cost-effective,  
59 environmentally friendly way (Raesossadati *et al.*, 2014; Muñoz *et al.* 2015). In this

60 microalgae-based process, the CO<sub>2</sub> is transferred from the gas to the liquid phase when  
61 the flue-gas/biogas is sparged into the cultivation broth, being subsequently fixed by  
62 microalgae during photosynthesis in the presence of light. Therefore, the CO<sub>2</sub> is not  
63 only removed from the gas preventing its emission to the atmosphere, but the C-CO<sub>2</sub> is  
64 recovered as valuable algal biomass, which can be further valorized (Raeesossadati *et*  
65 *al.*, 2014; Muñoz *et al.* 2015). Moreover, the necessary nutrients for microalgae growth  
66 can be supplemented from wastewaters, which increases the environmental  
67 sustainability of the process (Park and Craggs, 2010). However, most wastewaters are  
68 characterized by a low C/N/P ratio compared to that needed for microalgae growth  
69 (20:8:1 for urban wastewaters vs 106:16:1 to ensure balanced algae growth), therefore  
70 carbon limitation usually hinders nutrient recovery from wastewater. In this sense, CO<sub>2</sub>  
71 supply into the cultivation broth from biogas or flue-gas increases the availability of  
72 inorganic carbon, enhancing biomass productivity, ensuring complete nutrient recovery  
73 from wastewater and mitigating microalgae pH-derived inhibition (Arbid *et al.*, 2013;  
74 Posadas *et al.*, 2015).

75 The potential of algal-bacterial symbiosis for biogas (Toledo-Cervantes *et al.*, 2016;  
76 Toledo-Cervantes *et al.*, 2017b) or flue-gas (Posadas *et al.* 2015) purification combined  
77 to wastewater treatment has been already studied and demonstrated in open  
78 photobioreactors. However, few studies have focused on the implementation of this  
79 process in closed photobioreactors, which offers higher photosynthetic efficiencies by  
80 avoiding light limitation, enhanced biomass productivities and better CO<sub>2</sub> mass transfer  
81 (Chisti, 2007; Arbid *et al.*, 2013). On the other hand, this photosynthetic CO<sub>2</sub>-abatement  
82 process can be further optimized by implementing nutrient supplementation strategies to  
83 promote the production of storage compounds in the algal biomass (Mooij *et al.*, 2013).  
84 In this context, the production of a biomass with a high content in the metabolites of

85 interest will increase the economic sustainability of the process (Toledo-Cervantes *et*  
86 *al.*, 2017a).

87 This work aimed at evaluating the long-term performance of a tubular photobioreactor  
88 interconnected to a CO<sub>2</sub> absorption column for the abatement of CO<sub>2</sub> from biogas and  
89 flue-gas. Furthermore, a feast-famine regime was implemented in order to exploit the  
90 cyclic nitrogen absence for the continuous production of high-energy storage  
91 compounds.

92

## 93 **2. Materials and methods**

### 94 **2.1 Experimental system**

95 The experimental system consisted of a tubular photobioreactor interconnected to a  
96 mixing chamber and a CO<sub>2</sub> absorption column (AC) (Figure 1). The tubular  
97 photobioreactor was composed of 12 tubes of 6 cm inner diameter and 94 cm of length,  
98 with a total volume of 45.5 L. The mixing chamber (60 cm height, 50 cm width and 35  
99 cm length) had a working volume of 60 L. The absorption column was 2 m height (1.73  
100 m water column) with an internal diameter of 5 cm and a working volume of 3.5 L. Two  
101 sets of high intensity LED PCBs were placed at both sides of the photobioreactor to  
102 provide a photosynthetic active radiation (PAR) of  $\sim 1100 \mu\text{mol m}^{-2} \text{s}^{-1}$ . Light:dark  
103 cycles of 12:12 h of the PAR were fixed. The cultivation broth was re-circulated  
104 through the tubular photobioreactor and the mixing chamber at a linear velocity of 0.5  
105 m s<sup>-1</sup>. The absorption column was operated by supplying co-currently the cultivation  
106 broth from the mixing chamber and biogas/flue-gas (through a stainless steel diffuser of  
107 2  $\mu\text{m}$  pore size) at the bottom of the column. The operating parameters such as liquid  
108 and gas flow rates of the absorption column and nutrients/digestate solution flow rates  
109 are described in section 2.2.

110

## 111 **2.2 Experimental system operation**

### 112 **2.2.1 Operation with biogas (A):**

113 Prior operation, an abiotic CO<sub>2</sub>/H<sub>2</sub>S removal test was performed in order to determine  
114 the optimum liquid to gas flow rates (L/G) ratio in the AC, which maximizes the CO<sub>2</sub>  
115 and H<sub>2</sub>S removal from biogas without compromising the CH<sub>4</sub> content and the quality of  
116 the upgraded biogas due to N<sub>2</sub> and O<sub>2</sub> desorption (Toledo-Cervantes *et al.*, 2016). The  
117 biogas used was a synthetic mixture of 29.5% CO<sub>2</sub>, 0.5% H<sub>2</sub>S and 70% CH<sub>4</sub>. The liquid  
118 phase was a modified Bristol medium (final pH = 7.5) (g L<sup>-1</sup>): NaNO<sub>3</sub> (1.5), CaCl<sub>2</sub>  
119 2H<sub>2</sub>O (0.025), MgSO<sub>4</sub>·7H<sub>2</sub>O (0.075), K<sub>2</sub>HPO<sub>4</sub> (0.075), KH<sub>2</sub>PO<sub>4</sub> (0.175), NaCl (0.025),  
120 and 1 mL L<sup>-1</sup> of a micronutrient solution (2.86 g L<sup>-1</sup> H<sub>3</sub>BO<sub>3</sub>, 1.81 g L<sup>-1</sup> MnCl<sub>2</sub> 4H<sub>2</sub>O,  
121 0.22 g L<sup>-1</sup> ZnSO<sub>4</sub> 7H<sub>2</sub>O, 0.39 g L<sup>-1</sup> Na<sub>2</sub>MoO<sub>4</sub> 2H<sub>2</sub>O, 0.079 g L<sup>-1</sup> CuSO<sub>4</sub> 5H<sub>2</sub>O and 49.4  
122 mg L<sup>-1</sup> Co(NO<sub>3</sub>)<sub>2</sub> 6H<sub>2</sub>O). The liquid recirculation rates tested were 60, 150, 300 and 450  
123 mL min<sup>-1</sup> while the biogas flow rate was set at 40 mL min<sup>-1</sup>. Hence, L/G ratios ranging  
124 from 1 to 11 were studied. The AC was allowed to stabilize for two times the hydraulic  
125 retention time prior monitoring the upgraded biogas composition by GC-TCD.

126

127 The system was inoculated with the microalgae *Acutodesmus obliquus* at an initial  
128 suspended solids concentration (SST) of 0.1 g L<sup>-1</sup>, and operated for biogas upgrading  
129 during 150 days. The CO<sub>2</sub> contained in the synthetic biogas previously described was  
130 used as carbon source for microalgae growth, while nutrients were supplied by means of  
131 the modified Bristol medium. During stage IA (from day 1 to 54) the synthetic biogas  
132 was fed during the illuminated period into the absorption column at a flow rate of 40  
133 mL min<sup>-1</sup> and the liquid broth was recirculated through the AC at a flow rate of 400 mL  
134 min<sup>-1</sup> (L/G ratio = 10). The modified Bristol medium was fed into the mixing chamber

135 at a flow rate of 3 mL min<sup>-1</sup> (hydraulic retention time, HRT = 50 d) during the light  
136 period. The feed flow rate was selected according to the nitrogen load needed for the  
137 complete photosynthetic fixation of the CO<sub>2</sub> contained in the biogas, assuming a  
138 biomass composition of 50 % of carbon and 10 % of nitrogen (Groobelar, 2004). From  
139 days 54 to 77 (stage IIA) the synthetic biogas was continuously fed into the absorption  
140 column (24 h gas feeding), therefore the mineral medium flow rate was increased to 6  
141 mL min<sup>-1</sup> (HRT= 25 d). During this stage, both the gas and the liquid flow rates through  
142 the AC were maintained constant (L/G ratio = 10). Finally, in Stage IIIA (days 77 to  
143 150) the synthetic biogas was only fed during the light period at 80 mL min<sup>-1</sup> (L/G ratio  
144 = 5) while the mineral medium feeding flow rate was kept at 6 mL min<sup>-1</sup> (HRT= 25 d).

145

#### 146 **2.2.2 Operation with flue-gas (B):**

147 During experiment B, a synthetic flue-gas composed of 20% of CO<sub>2</sub> and 80% of N<sub>2</sub> was  
148 used as carbon source for the growth of microalgae. During Stage IB (from day 151 to  
149 280) the synthetic flue-gas was fed at 50 mL min<sup>-1</sup> during the light period (L/G = 5).  
150 The mineral medium previously described was modified by decreasing the NaNO<sub>3</sub>  
151 concentration to 0.75 g L<sup>-1</sup> and the nitrogen load rate was adjusted to the photosynthetic  
152 fixation of the CO<sub>2</sub> contained in the flue-gas. Therefore, the nutrient solution was fed  
153 into the mixing chamber at a flow rate of 10 mL min<sup>-1</sup> (HRT= 15 d). From days 284 to  
154 297 (Stage IIB) no nitrogen source was added to the mineral medium in order to  
155 decrease the concentration of nitrogen in the cultivation broth to the non-assimilative  
156 nitrogen concentration (<2 mg-N L<sup>-1</sup>). No further modifications were implemented  
157 during this period.

158 During stage IIIB (days 297 to 336) mineral medium was fed at the same HRT (15 d)  
159 only during the dark period in order to promote the production of high-energy

160 compounds. The nitrogen load was set at  $511 \text{ mg d}^{-1}$  based on the biomass productivity  
161 observed in stage IIB and considering a biomass nitrogen content of 7% (experimental  
162 data from stage IB). No nutrient was supplemented during the light period in which the  
163 flue-gas was fed (Mooij *et al.*, 2015). From days 336 to 380 (Stage IVB) the synthetic  
164 nutrient solution was substituted by a diluted anaerobic digestate solution obtained from  
165 the wastewater treatment plant of Valladolid city (Spain), with an average composition  
166 of total nitrogen (TN), inorganic carbon (IC) and total phosphorus (TP) of  $660 \pm 46$ ,  $524$   
167  $\pm 49$  and  $48 \pm 3 \text{ mg L}^{-1}$ , respectively. The digestate feeding flow rate was adjusted to  
168 ensure the same nitrogen load as in Stage IIB.

169 The steady state biomass chemical (C, N, P, and S) and biochemical (proteins,  
170 carbohydrates, lipids, and ashes) composition were determined at the end of the light  
171 and dark periods two times a week.

172 During both operating periods (*i.e.* A and B), the biomass concentration, measured as  
173 TSS, was determined twice a week. The temperature and dissolved oxygen  
174 concentration (DO) in the cultivation broth were daily in-situ monitored. Inlet (biogas  
175 and flue-gas) and outlet (upgraded biogas and treated gas) gas samples were drawn  
176 twice a week to analyze the composition by GC-TCD. The inlet and outlet gas flow  
177 rates in the AC were also periodically measured in order to perform the gas mass  
178 balance. Samples of 100 mL of the cultivation broth and mineral medium or diluted  
179 digestate were taken twice a week to determine the pH and concentrations of TN, IC,  
180 nitrate ( $\text{NO}_3^-$ ), sulfate ( $\text{SO}_4^{2-}$ ) and phosphate ( $\text{PO}_4^{3-}$ ). The population of microalgae in  
181 the photobioreactor was identified by microscopic observation at the end of each steady  
182 state.

183

### 184 **2.3 Analytical methods**

185 Biomass concentration was determined by dry weight (105 °C, 24 h).  $\text{NO}_3^-$ ,  $\text{PO}_4^{3-}$  and  
186  $\text{SO}_4^{2-}$  concentrations were analyzed by HPLC-IC according to Serejo *et al.*, (2015).  
187 Dissolved IC and TN concentrations were determined using a Shimadzu TOC-VCSH  
188 analyzer (Japan) equipped with a TNM-1 chemiluminescence module. The PAR was  
189 measured with a LI-190 quantum sensor and recorded with a LI-250A light meter  
190 (Lincoln, Nebraska, USA). The pH was monitored with a pH meter Eutech Cyberscan  
191 pH 510 (Eutech instruments, The Netherlands), while the DO concentration was  
192 measured with an Oxi 330i oximeter (WTW, Germany). The gas composition ( $\text{CO}_2$ ,  
193  $\text{H}_2\text{S}$ ,  $\text{O}_2$ ,  $\text{N}_2$ , and  $\text{CH}_4$  concentrations) was analyzed by GC-TCD according to Posadas  
194 *et al.* (2015). Microalgae identification was performed by microscopic observations  
195 (OLYMPUS IX70, USA) after sample fixation with 5% of lugol acid.

196 The carbohydrate and protein content of the biomass was determined according to the  
197 methodology described in Dubois *et al.* (1956) and Lowry *et al.* (1951), respectively.  
198 For carbohydrates determination, 1.5 mL of cultivation broth (biomass concentration  
199  $\sim 0.2 \text{ g L}^{-1}$ ) was mixed with 4 mL of  $\text{H}_2\text{SO}_4$  1 M. Afterwards, the sample was heated for  
200 20 min at 100 °C and centrifuged for 5 min at 10000 rpm. A volume of 0.5 mL of the  
201 supernatant was mixed with 0.5 mL of a 5% phenol solution and stood for 40 min. After  
202 that period, 2.5 mL of concentrated  $\text{H}_2\text{SO}_4$  were added, and then the optical density was  
203 determined at 485 nm. Protein content was measured by mixing 1 mL of the cultivation  
204 broth and 1 mL of NaOH 1 N and heated at 100 °C for 20 min. After centrifugation (5  
205 min at 10000 rpm), 0.4 mL of the supernatant were mixed with 2 mL of a solution  
206 composed of 1:25 (v/v) of 5% (w/v)  $\text{Na}_2\text{CO}_3$  and 0.5% (w/v)  $\text{CuSO}_4$  in 1% (p/v) sodium  
207 potassium tartrate. The mixture was stood for 10 min. Subsequently, 0.4 mL of 1 N  
208 Folin & Ciocalteu's phenol reagent was added and kept in dark for 30 min. The optical

209 density of the preparation was then read at 750 nm. Total lipids were determinate by  
210 direct extraction in an automatic Soxhlet extraction unit (SER 148 Series, Velp  
211 Scientifica) using hexane as solvent. The extraction conditions were set as follows:  
212 extraction temperature 130°C, immersion time 60 min, and solvent recovery time 120  
213 min. The ashes content was determine as volatile solids according to Standard methods  
214 (Eaton *et al.*, 2005). Finally, the elemental composition of biomass was determined  
215 using a CHNS analyzer (LECO CHNS-932) for C and N content, while an Inductively  
216 Coupled Plasma-Optical Emission Spectrometer (ICP-OES, Varian 725-ES) was used  
217 for P and S content determination.

218

### 219 **3. Results and Discussions**

#### 220 **3.1. Biogas upgrading**

##### 221 **3.1.1. Abiotic removal of CO<sub>2</sub> and H<sub>2</sub>S**

222 Table 1 shows the upgraded biogas composition and removal efficiencies obtained at  
223 different liquid to biogas flow rate ratios during the preliminary abiotic test. Maximum  
224 mass transfer efficiencies for both CO<sub>2</sub> and H<sub>2</sub>S were obtained at an L/G of 11 (~91 and  
225 99%, respectively). Similar studies have reported higher CO<sub>2</sub> and H<sub>2</sub>S removal  
226 efficiencies (REs) of  $98.8 \pm 0.2$  and  $97.1 \pm 1.4\%$ , respectively, regardless of the L/G  
227 ratio tested but using the algal-bacterial broth at a pH of 10 (Toledo-Cervantes, 2016).  
228 Furthermore, Serejo *et al.* (2015) obtained a CO<sub>2</sub>-RE of  $95 \pm 2\%$  at L/G ratios above 15  
229 because of the lower pH ( $\approx 7.9$ ) of the cultivation broth. During these studies, higher N<sub>2</sub>  
230 (7-25%) and O<sub>2</sub> (3-7%) concentrations were observed in the upgraded biogas as a result  
231 of the photosynthetic activity ( $DO \geq 8 \text{ mg-O}_2 \text{ L}^{-1}$ ) and the nitrogen concentration in the  
232 liquid broth ( $\sim 14 \text{ mg-N}_2 \text{ L}^{-1}$ ), and its subsequent stripping from the cultivation broth. In  
233 this context, closed photobioreactors are considered a viable alternative to open systems

234 for preventing desorption of nitrogen in the absorption column, since the cultivation  
235 broth is not in contact with the atmosphere. However, to the best of our knowledge, this  
236 is the first study reporting the upgraded biogas composition in a closed tubular  
237 photobioreactor (Table 1). Maximum methane concentrations of ~85% were achieved in  
238 the abiotic test since methane content in the upgraded biogas is compromised between  
239 the low nitrogen desorption and the low CO<sub>2</sub> removal reached at L/G ratios <11 and a  
240 pH of the cultivation broth of 7.5. In this sense, it is worth noticing that, since H<sub>2</sub>S and  
241 CO<sub>2</sub> are acidic gases, higher absorption of these components from the biogas is  
242 expected under biotic conditions as a result of the increase in pH by algal  
243 photosynthetic activity.

244

### 245 **3.1.2. Photosynthetic CO<sub>2</sub> and H<sub>2</sub>S removal from biogas**

246 During stage IA, CO<sub>2</sub> and H<sub>2</sub>S were effectively removed from biogas at  $97.6 \pm 0.4$  and  
247  $98.3 \pm 0.0\%$ , respectively (Figure 2a). As previously discussed, the high removals here  
248 observed were supported by the photosynthetic activity of microalgae, which allowed  
249 for a dissolved oxygen concentration of  $8.1 \pm 1.1$  mg-O<sub>2</sub> L<sup>-1</sup> and a pH of  $10.7 \pm 0.5$ .  
250 Some studies have previously reported efficient biogas upgrading by alkalophilic  
251 microalgae cultivation or by using highly alkaline digestate (Franco-Morgado *et al.*,  
252 2017; Toledo-Cervantes *et al.*, 2016, 2017b). However, during this study, only the high  
253 photosynthetic activity of microalgae supported the alkaline pH needed for the effective  
254 transfer of CO<sub>2</sub> and H<sub>2</sub>S from the gas phase into the cultivation broth. Under these  
255 conditions, the upgraded biogas had a composition of CO<sub>2</sub>  $0.4 \pm 0.4$  %, H<sub>2</sub>S  $0.01 \pm$   
256  $0.01\%$ , O<sub>2</sub>  $8.3 \pm 2.9\%$ , N<sub>2</sub>  $7.7 \pm 2.9\%$  and CH<sub>4</sub>  $83.6 \pm 1.8\%$ , which is suitable for  
257 electricity production in motor generators (Figure 2b).

258 During stage IIA, the operating strategy of feeding the biogas continuously decreased  
259 the pH of the cultivation broth to  $7.2 \pm 0.7$ . This acidic condition and the likely toxic  
260 effect of the  $\text{H}_2\text{S}$  inhibited the microalgae activity, which was confirmed by the low DO  
261 concentration observed,  $3.6 \pm 1.8 \text{ mg-O}_2 \text{ L}^{-1}$ . Despite some studies have demonstrated  
262 that biogas containing up to 0.5% of  $\text{H}_2\text{S}$  (5000 ppm<sub>v</sub>) does not inhibit microalgae  
263 growth, González-Sánchez and Posten (2017) have recently reported inhibitory effects  
264 at concentrations higher than 200 ppm<sub>v</sub>. These results were associated to the closed  
265 configuration of the photobioreactor, which likely induced the accumulation of  $\text{HS}^-$  in  
266 the cultivation broth during the dark period when dissolved oxygen concentration  
267 decreases, thus preventing further  $\text{HS}^-$  oxidation. However, the  $\text{H}_2\text{S}$  removal remained  
268 similar to that observed in stage IA at  $99.7 \pm 0.0 \%$ , due its higher solubility compared  
269 to that of  $\text{CO}_2$  (Henry law constants:  $\text{H}_2\text{S} = 1 \times 10^{-3}$  vs.  $\text{CO}_2 = 3.3 \times 10^{-4} \text{ mol m}^{-3} \text{ Pa}^{-1}$ )  
270 (Sanders, 1999). This promotes the mass transfer of  $\text{H}_2\text{S}$  to the liquid phase, leading to a  
271 toxic effect at low DO concentrations. In contrast, because of the decrease in pH driven  
272 by the low photosynthetic activity, the  $\text{CO}_2$  removal decreased to  $57.0 \pm 0.1 \%$  (Figure  
273 2a). It is worth noticing that similar photosynthetic biogas upgrading studies have  
274 reported  $\text{CO}_2$  removals in the range of 50–98.8% depending on the alkalinity of the  
275 cultivation broth and the environmental conditions in high rate algal ponds (both  
276 indoors and outdoors) (Franco-Morgado *et al.*, 2017; Posadas *et al.*, 2017; Toledo-  
277 Cervantes *et al.*, 2017). These findings highlight the need of pH control in this  
278 bioreactor configuration to avoid the deterioration of the  $\text{CO}_2$  removal performance.

279 In order to recover the cultivation broth conditions suitable for biogas upgrading, the  
280 biogas inlet flow was doubled during the illuminated period (stage IIIA). This operating  
281 strategy allowed increasing the pH up to  $10.0 \pm 0.2$  and the  $\text{CO}_2$  and  $\text{H}_2\text{S}$  removals  
282 stabilized at  $98.3 \pm 0.0$  and  $99.9 \pm 0.0\%$ , respectively (Figure 2a). Under these favorable

283 conditions, the upgraded biogas had a similar composition of that obtained in stage IA:  
284 CO<sub>2</sub> 1.8 ± 3.4%, H<sub>2</sub>S 0.00 ± 0.00, O<sub>2</sub> 9.6 ± 3.3%, N<sub>2</sub> 6.0 ± 2.2% and CH<sub>4</sub> 82.6 ± 3.8%  
285 (Figure 2b). The slightly higher oxygen concentration recorded in the upgraded biogas  
286 was correlated with the higher DO concentration in the cultivation broth (10.8 ± 1.2 mg-  
287 O<sub>2</sub> L<sup>-1</sup>) when compared to stage IA.

288 Regarding algal biomass production, the photobioreactor operation at a HRT = 50 d  
289 during stage IA lead to a biomass productivity of 2.5 ± 0.2 g m<sup>-2</sup> d<sup>-1</sup>, which entailed  
290 nitrogen and carbon recoveries of 56.6 ± 3.1% and 50.5 ± 4.5%, respectively. In stage  
291 IIA, the lower HRT of 25 days resulted in an increase in biomass concentration from 1.6  
292 ± 0.1 to 2.2 ± 0.1 g L<sup>-1</sup>, that corresponded to a biomass productivity of 7.2 ± 0.3 g m<sup>-2</sup> d<sup>-1</sup>.  
293 During this period, the mass balance showed that 86.2 ± 2.6% of the C-CO<sub>2</sub> removed  
294 from biogas and 81.4 ± 3.2% of the nitrogen fed were recovered as biomass. Finally, in  
295 stage IIIA, the doubling of the carbon load during the illuminated period allowed  
296 increasing the biomass concentration to 2.5 ± 0.1 g L<sup>-1</sup> together with a biomass  
297 productivity of 8.0 ± 0.2 g m<sup>-2</sup> d<sup>-1</sup>. Under these conditions, a complete nitrogen and  
298 carbon recovery as algal biomass was observed. These results confirm the potential of  
299 tubular photobioreactors for effective C-CO<sub>2</sub> recovery from biogas and nutrients  
300 removal. Furthermore, closed photobioreactors are recognized for the higher biomass  
301 productivities achieved in comparison with open systems. However, due to the lack of  
302 standardization of the reported values, volumetric productivities (g L<sup>-1</sup> d<sup>-1</sup>) are often  
303 used for closed photobioreactors instead of areal productivity (g m<sup>-2</sup> d<sup>-1</sup>), which hampers  
304 a fair comparison between both configurations. In this sense, while productivities of  
305 0.06 g L<sup>-1</sup> d<sup>-1</sup> have been reported for closed photobioreactors treating biogas (Meier *et*  
306 *al.*, 2016), productivities in the range of 2.2 – 15 g m<sup>-2</sup> d<sup>-1</sup> are commonly achieved in  
307 open systems, which in fact represents volumetric productivities between 0.015 and 0.1

308 g L d<sup>-1</sup> (Toledo-Cervantes *et al.*, 2016; Posadas *et al.*, 2017; Toledo-Cervantes *et al.*,  
309 2017). In this study, the biomass productivity of 8.0 g m<sup>-2</sup> d<sup>-1</sup> was equivalent to a  
310 volumetric biomass productivity of 0.18 g L d<sup>-1</sup>, which exceeds previous values  
311 reported for open systems.

312

### 313 **3.2. High-value algal biomass production from flue-gas**

314 Microalgae are capable of producing high-energy compounds, which can contribute to  
315 the economic viability of the photosynthetic CO<sub>2</sub> removal processes either from biogas  
316 or flue-gas. Carbohydrates accumulation triggered by nitrogen starvation is one of the  
317 most effective ways to obtain added-value biomass (Ho *et al.*, 2015). It is important to  
318 highlight that this operation is performed batch-wise, since a previous biomass  
319 production stage is typically required before inducing such accumulation due to the  
320 different nutrient requirements of both biochemical processes. In this sense, the concept  
321 of “survival of the fittest” introduced by Mooij *et al.* (2013) was here applied as a  
322 strategy to induce the continuous accumulation of high-energy storage compounds in  
323 the produced microalgae while cleaning flue-gas.

324 In stage IB, the biogas fed during operation stages IA-III A was replaced by a synthetic  
325 flue-gas containing 20% of CO<sub>2</sub>; therefore, the mineral medium was modified  
326 accordingly in order to balance the carbon/nitrogen load to keep the same assimilative  
327 nutrient removal reached in *section 3.1.2*. Consequently, the nutrient solution was fed at  
328 an HRT of 15 days and the system was operated until constant biomass concentration of  
329  $1.5 \pm 0.0$  g L<sup>-1</sup> was achieved. Under steady conditions, ~100% of the N-NO<sub>3</sub> fed and the  
330 C-CO<sub>2</sub> removed from flue-gas were recovered as biomass (Table 2). The harvested  
331 biomass, corresponding to the total effluent obtained at the end of the alimentation

332 period (*i.e.* after the light period), reached  $8.3 \pm 0.2 \text{ g m}^{-2} \text{ d}^{-1}$ , with a composition of  
333 ~22.1% carbohydrates, 48.3 % proteins and 14.6 % ashes (Table 3).

334 During stage IIB, the nitrogen source was removed from the mineral medium while  
335 maintaining the same nutrients load (Figure 3). The latter strategy was implemented in  
336 order to decrease the nitrogen concentration in the cultivation broth to a non-  
337 assimilative concentration of  $\sim 1.3 \text{ mg-N L}^{-1}$  in which the accumulation of high-energy  
338 compounds such as lipids and/or carbohydrates can occur (Figure 3).

339 Once N concentrations  $< 2 \text{ mg-N L}^{-1}$  were achieved in the cultivation broth, the mineral  
340 medium was supplemented with  $\text{N-NO}_3$  and fed only during the dark period at the  
341 required nitrogen load to keep the same biomass productivity of that recorded in stage  
342 IIB (Table 2). As can be observed from Figure 3, microalgae were initially not able to  
343 consume the nitrogen in the absence of light (days 297-320 of stage IIB). This can be  
344 explained by the fact that nitrogen assimilation requires the fixed  $\text{CO}_2$  and the energy  
345 generated in the photosynthetic process. Moreover, to assimilate nitrate, the molecule  
346 has to be transported across the membrane and be reduced to ammonia, consuming in  
347 the process large amounts of energy, carbon, and protons (Perez-Garcia *et al.*, 2011).

348 After this initial adaptation period of  $\sim 20$  days, consumption of the supplied nitrogen  
349 during the dark phase was observed from day 320 onwards. This fact was attributed to  
350 the concomitant degradation of storage starch in the dark period. This phenomenon  
351 would require a regenerative cycling of adenine nucleotides and phosphate that can be  
352 supported by chlororespiration, which plays an important role in the dark recovery of  
353 plants from photoinhibition through *de novo* protein synthesis (Beardall *et al.*, 2003). It  
354 has been suggested that chlororespiration supplies ATP for maintenance and synthetic  
355 processes in chloroplasts in the dark, supplementing ATP from glycolysis in the plastids  
356 (Raven and Beardall, 2003). Therefore, the accumulated high-energy molecules in the

357 form of glucose-based carbohydrates might be oxidized through the Embden–Meyerhof  
358 pathway and/or the Pentose Phosphate pathway, the energy production routes (NADPH,  
359 ATP), during the dark period. In that way, enzymes involved in nitrate assimilation  
360 (nitrate reductase and nitrite reductase) that work sequentially, had the required energy  
361 to catalyze nitrate to ammonium in the dark period; while during the light period CO<sub>2</sub> is  
362 reduced to carbohydrates through the Calvin cycle. This hypothesis was supported by  
363 the higher carbohydrate content recorded by the end of the light period, *i.e.* the 12 h  
364 nitrogen famine period, in contrast to that recorded by the end of the dark period, *i.e.* the  
365 12 h nitrogen supplementation period (Table 3).

366 Similar results were observed during stage IVB, when the mineral medium was replaced  
367 by an anaerobic digestate but keeping the same nitrogen (N-NH<sub>4</sub><sup>+</sup>) load. At this point, it  
368 is worth noticing that the variation in biomass productivity observed in stage IVB was  
369 likely due to the decrease in CO<sub>2</sub> removal down to 91.6± 11.3%, driven by the lower pH  
370 as a result of ammonium feeding. Furthermore, during this period the occurrence of  
371 *Pseudanabaena* sp. (12%) was recorded which was attributed to lack of aseptic  
372 conditions of the digestate. This fact is frequently reported in open systems where rapid  
373 variations in microalgae population are expected. Moreover, the appearance of this  
374 cyanobacterium has been previously reported in wastewater treatment processes  
375 coupled to biogas upgrading (Serejo *et al.*, 2015).

376 Finally, carbohydrates productivities ~3 g m<sup>-2</sup> d<sup>-1</sup> were recorded under the N-dark  
377 feeding strategies, which represents 1.7 times the productivity reached under normal  
378 nutrition conditions (Figure 4). The high concentration of carbohydrates reached is  
379 preferred for its chemical or biological valorization, for instance as the substrate for  
380 biohydrogen by dark-fermentation (Chen *et al.*, 2016), ethanol (John *et al.*, 2011) or  
381 biogas production (Zamalloa *et al.*, 2011). Furthermore, the biomass production through

382 wastewater treatment significantly contributes to the flue-gas or biogas cleaning process  
383 (Toledo-Cervantes *et al.*, 2017a.). These results confirm the feasibility of applying this  
384 novel strategy for inducing the accumulation of high-energy storage compounds during  
385 the photosynthetic abatement of CO<sub>2</sub> coupled with wastewater treatment, since it allows  
386 for a continuous production of added-value algal biomass.

387

#### 388 **4. Conclusions**

389 To the best of our knowledge, this is the first experimental study reporting the long-term  
390 performance of a tubular photobioreactor for the abatement of CO<sub>2</sub> from exhaust gases  
391 (biogas and flue-gas) coupled with algal biomass production. The system here proposed  
392 showed an efficient removal of CO<sub>2</sub> from gas streams (>98%), the upgraded biogas  
393 composition meeting the required standards for electricity production. Moreover, the  
394 innovative nutrient supplementation strategy, *i.e.* feeding nutrients during the dark  
395 period, allowed enhancing the carbohydrates content in the produced biomass by 1.7  
396 times regardless of the nitrogen source. In summary, this study confirmed the potential  
397 of the photosynthetic CO<sub>2</sub> removal process in closed photobioreactors to support  
398 nutrient recovery from digestate and production of added-value biomass with high  
399 carbohydrates content, resulting in a cost-efficient and environmentally-friendly  
400 technology.

401

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407

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500 **Figure captions**

501

502 **Figure 1.** Schematic diagram of the experimental system used for the photosynthetic  
503 CO<sub>2</sub> removal from biogas and flue-gas.

504 **Figure 2.** Time course of **a)** the CO<sub>2</sub> (○) and H<sub>2</sub>S (▲) removal efficiencies; and **b)** the  
505 upgraded biogas composition CH<sub>4</sub> (●), CO<sub>2</sub> (■), N<sub>2</sub> (□) and O<sub>2</sub> (+).

506 **Figure 3.** Time course of the total nitrogen (TN) concentration in the cultivation broth.  
507 Open circles represent the nitrogen concentration at the end of the dark period where the  
508 nitrogen supplementation took place (8:00 am) and solid squares represent the nitrogen  
509 concentration at the end of the illuminated period (8:00 pm).

510 **Figure 4.** Biomass concentration (■) and carbohydrates productivity (white bars)  
511 achieved under different N-supplementation strategies.

**Table 1.** Abiotic removal efficiencies and upgraded biogas composition obtained at different liquid to biogas flow rate ratios (L/G).

L/G	Removal efficiencies (%)		Upgraded biogas composition (%)				
	H <sub>2</sub> S	CO <sub>2</sub>	CH <sub>4</sub>	H <sub>2</sub> S	CO <sub>2</sub>	N <sub>2</sub>	O <sub>2</sub>
1	88.2 ± 1.3	61.6 ± 3.1	84.1 ± 0.4	0.05 ± 0.0	10.6 ± 1.1	4.1 ± 1.0	2.2 ± 0.1
4	93.2 ± 0.4	76.6 ± 1.5	85.9 ± 0.6	0.03 ± 0.0	6.7 ± 0.1	6.0 ± 0.7	3.5 ± 0.2
7	95.5 ± 1.3	89.6 ± 2.3	81.3 ± 0.3	0.02 ± 0.0	3.1 ± 0.4	14.3 ± 2.5	6.5 ± 0.9
11	98.6 ± 0.9	90.8 ± 3.0	82.8 ± 0.1	0.01 ± 0.0	2.7 ± 1.2	13.2 ± 2.5	7.1 ± 0.5

**Table 2.** Average values for operating parameters recorded during flue-gas cleaning.

<b>Stage</b>	<b>pH</b>	<b>DO (mg-O<sub>2</sub> L<sup>-1</sup>)</b>	<b>Productivity (g m<sup>-2</sup> d<sup>-1</sup>)</b>	<b>CO<sub>2</sub> removal (%)</b>
<b>IB</b>	10.1 ± 0.1	10.7 ± 0.8	8.3 ± 0.2	99.8 ± 0.7
<b>IIB</b>	10.5 ± 0.4	8.7 ± 0.3	6.7 ± 0.1	99.3 ± 0.1
<b>IIIB</b>	10.4 ± 0.2	6.0 ± 0.2	6.8 ± 0.3	99.3 ± 0.0
<b>IVB</b>	9.3 ± 0.4	6.2 ± 0.3	5.8 ± 0.4	91.1 ± 1.4

**Table 3.** Biochemical composition of algal biomass under different nutrition strategies

Stage	I-IIB (Regular nutrition)	IIIB		IVB	
		N-famine - light	N-NO <sub>3</sub> <sup>-</sup> - dark	N-Famine - light	N-NH <sub>4</sub> <sup>+</sup> - dark
<b>Carbohydrates (mg g<sub>b</sub><sup>-1</sup>)</b>	221.4 ± 56.2	439.9 ± 32.3	369.7 ± 27.9	453.6 ± 58.1	378.3 ± 11.9
<b>Proteins (mg g<sub>b</sub><sup>-1</sup>)</b>	482.9 ± 57.7	349.4 ± 53.0	526.0 ± 18.0	424.4 ± 54.7	476.8 ± 26.6
<b>Lipids (mg g<sub>b</sub><sup>-1</sup>)</b>	48.5 ± 3.9	40.0 ± 4.2	48.0 ± 2.5	31.5 ± 5.7	31.6 ± 3.8
<b>Ashes (mg g<sub>b</sub><sup>-1</sup>)</b>	146.0 ± 1.2	70.0 ± 0.0	40.0 ± 0.3	76.0 ± 0.1	56.8 ± 0.0

Figure 1  
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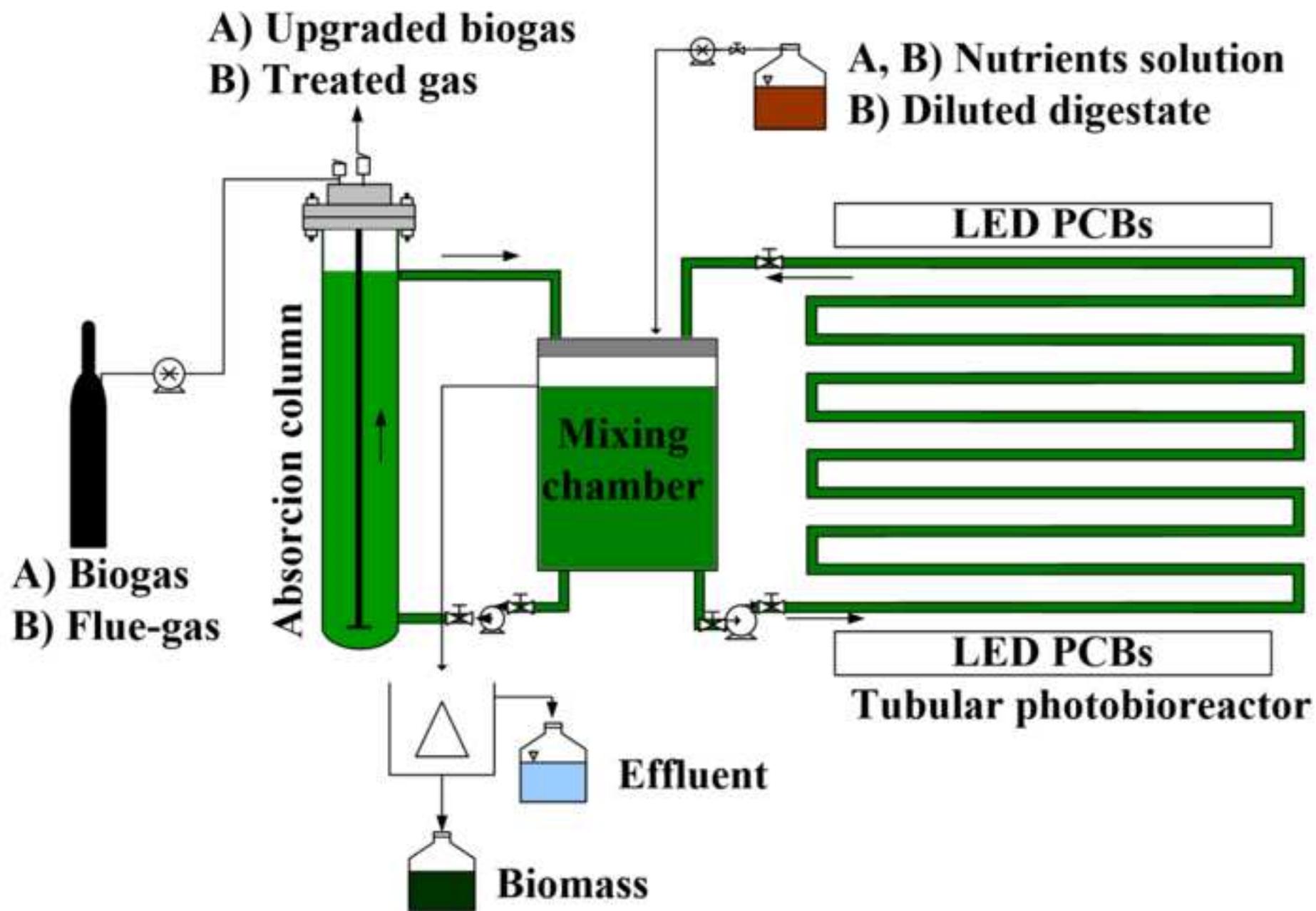


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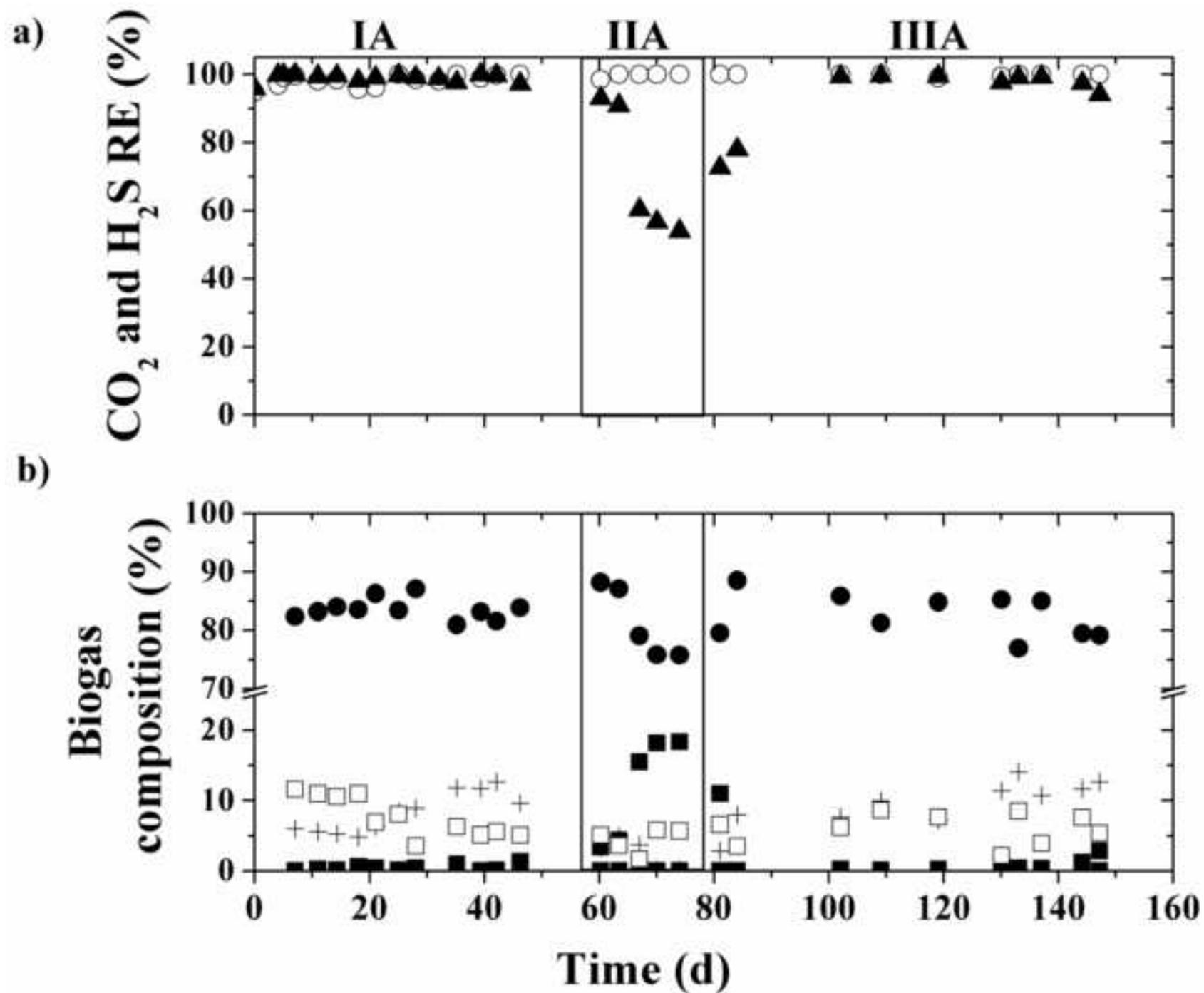


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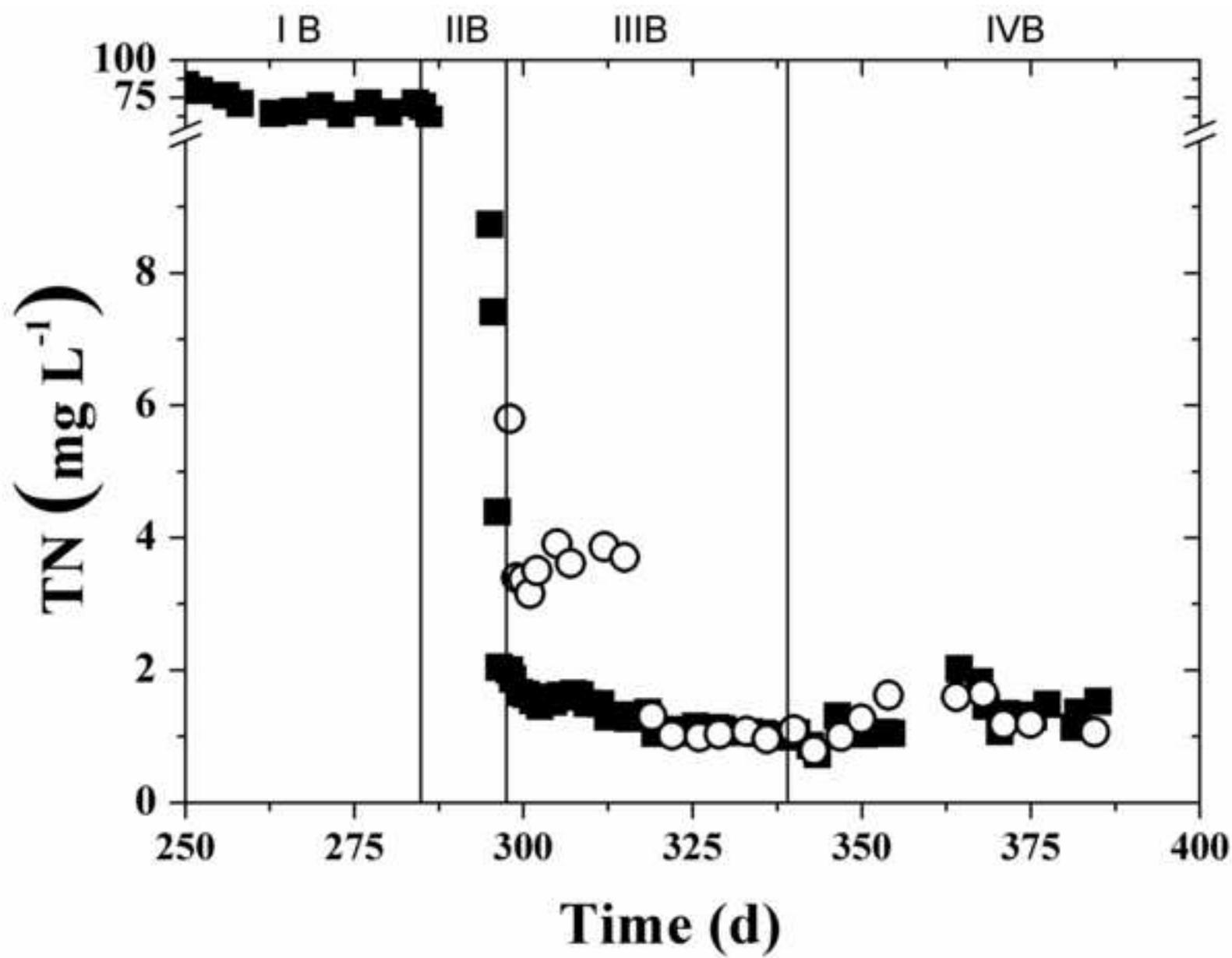
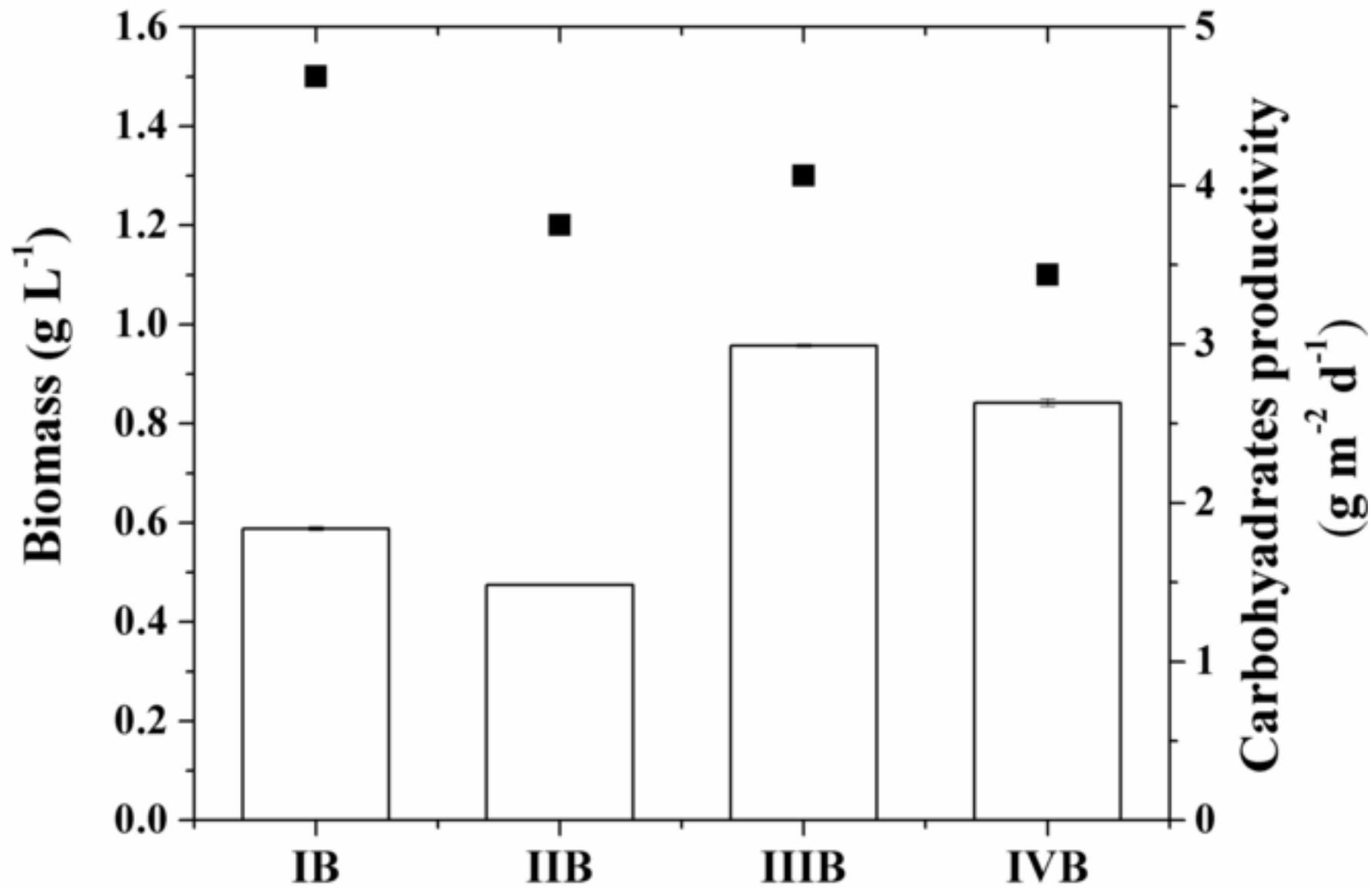


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