



Ru-SUPPORTED ON MESOPOROUS SILICA AS ACTIVE CATALYST FOR THE HYDROLYSIS OF THE HEMICELLULOSIC FRACTION OF WHEAT BRAN

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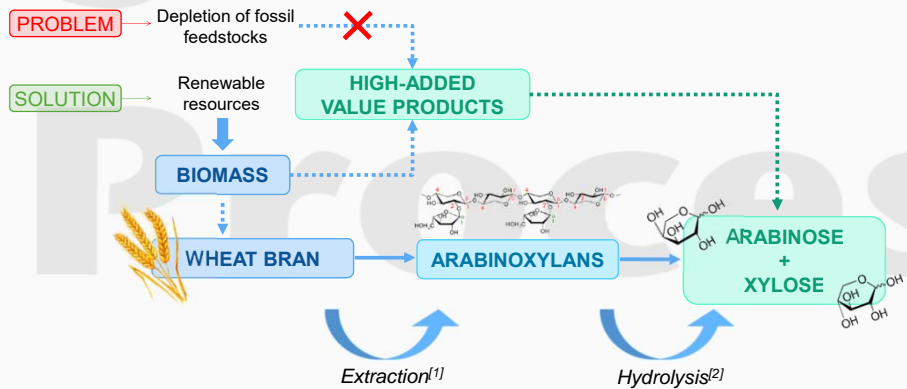
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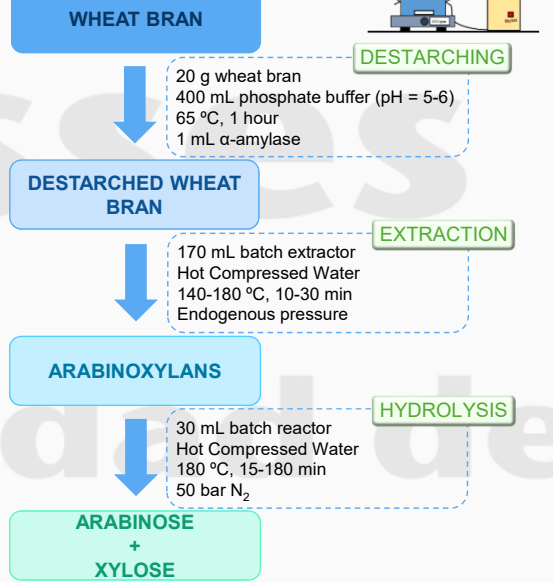
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1 Introduction

The current depletion of fossil resources is forcing society to look for renewable alternatives for energy and chemicals production. In this context, biomass is considered a sustainable and renewable feedstock suitable for the production of high-added value products (C5 sugars in this work). The production of C5 sugars (arabinose and xylose) from wheat bran can be divided into two main steps: 1) Extraction of arabinoxylans, mainly as oligomers and 2) Hydrolysis of arabinoxylans into monomers. In order to overcome the drawbacks of conventional methods with acids or enzymes, both steps have been studied by using Ru-supported catalysts.



2 Experimental

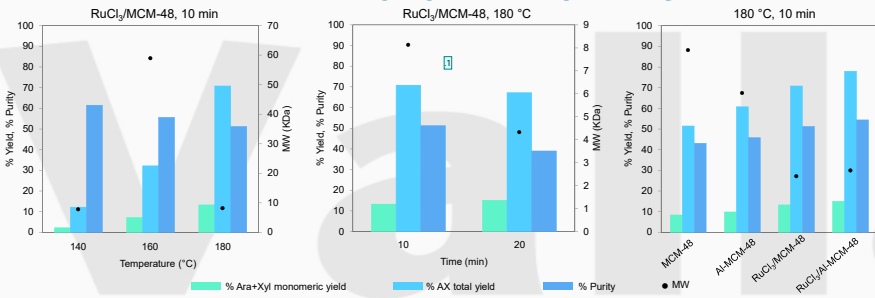


3 Results

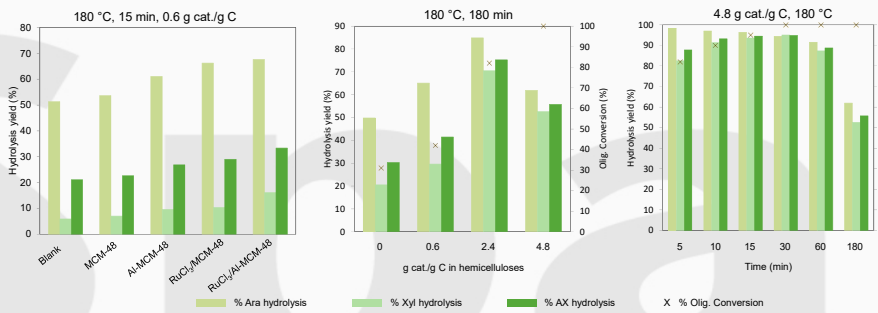
CATALYST CHARACTERIZATION

Catalyst	Ru (%)	S _{BET} (m ² ·g ⁻¹)	V _{pore} (cm ³ ·g ⁻¹)	D _{pore} (nm)	Acidity (mEq H ⁺ /g cat)
MCM-48	-	1298	0.87	2.2	0.293
Al-MCM-48	-	1352	0.81	2.5	0.598
RuCl ₃ /MCM-48	4	1032	0.63	2.2	0.738
RuCl ₃ /Al-MCM-48	4	1017	0.63	2.7	1.130

EXTRACTION OF ARABINOXYLANS

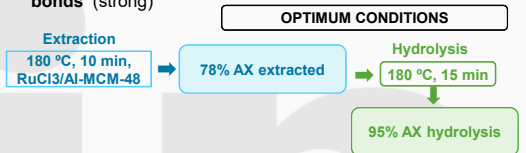


HYDROLYSIS OF ARABINOXYLANS



4 Conclusions

- Extraction + Hydrolysis of arabinoxylans have been studied using RuCl₃ catalysts over different mesoporous silica supports
- Acidity of the catalyst is a key parameter for these processes:
 - ↑ Acidity, ↑ Yield
- Ru³⁺ has demonstrated to be active, as it is a moderate Lewis acid
- Arabinose is always faster released than xylose:
 - Arabinose belongs to side chains and it is linked by α-glycosidic bonds (weak)
 - Xylose belongs to the backbone and it is linked by β-glycosidic bonds (strong)



REFERENCES

- [1] Sánchez-Bastardo, N., Romero, A., Alonso, E., 2017. Carbohydrate Polymers. 160, 143-152.
[2] Sánchez-Bastardo, N., Alonso, E., 2017. Bioresource Technology. 238, 379-388.

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