

Quantitative assessment of energy and resource recovery in evolutionary wastewater treatment plants based on Plant-Wide simulations

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Abstract

The growing development of technologies and processes for resource treatment and recovery is offering endless possibilities for creating new plant wide configurations or modifying existing ones. However, the configurations' complexity, the interrelation between technologies and the influent characteristics turned decision-making in a complex or unobvious process. In this frame, the Plant-Wide Model library presented in this paper allows a thorough, comprehensive and accurate analysis of different plant configurations that are basic aspects in decision-making from an energy and resource recovery perspective. In order to demonstrate the potential of the library and the need to run simulation analyses, this paper carries out a comparative analysis of evolutionary WWTP, from a techno-economic point of view. The selected layouts were (1) a conventional WWTP based on a modified version of the Benchmark Simulation Model No. 2, (2) an upgraded or retrofitted WWTP, and (3) a new WRRF concept denominated as C/N/P decoupling WWTP. The study was based on a preliminary analysis of the organic matter and nutrient energy use and recovery options, a comprehensive mass and energy flux distribution analysis in each configuration in order to compare and identify areas for improvement, and a cost analysis of these plants for different influent COD/TN/TP ratios. The plant layouts proposed in this paper are just a sample of the possibilities offered by current technologies. Even so, the library presented here is generic and can be used to construct any other plant layout, provided that a model is available.

Keywords

Energy and Resource recovery; Global mass and energy balance; Model-based assessment; Plant-Wide simulations;

1. INTRODUCTION

The purpose of the design and upgrade of conventional waste(water) treatment plants (WWTPs) has traditionally been to remove the residual organic compounds and nutrients contained in the water to fulfil quality standards. Resource or energy recovery was focused exclusively on obtaining energy from the biogas produced in anaerobic sludge digestion. This biogas production can supply from a quarter to half of the energy requirements for a WWTP with activated sludge (AS) process (Crawford *et al.*, 2010; McCarty *et al.*, 2011; Puchongkawarin *et al.*, 2015), which needs between 0.3 and 0.6 kWh m⁻³ treated water (Foley *et al.*, 2010) to fulfil the energetic needs of the plant. Nevertheless, this value is only one tenth of that associated to the heat of combustion of organic compounds contained in the wastewater (McCarty *et al.*, 2011; Shoener *et al.*, 2014; Kokabian *et al.*, 2015). Hence, if a greater proportion of this energy were recovered, treatment plants could become self-sufficient and producers of energy (Logan, 2004; Guest *et al.*, 2009).

Recent concerns about climate change or sustainability have led to an increasing awareness of the importance of resource recovery, energy minimization and environmental impact assessment, which in turn has resulted in tightening effluent standards. Under this changing context, a new paradigm

61 has emerged in which municipal wastewater (MWW), traditionally considered to be a pollution
62 problem and an energy- and chemical-intensive activity with excess sludge disposal issues (Gude,
63 2015), is starting to be thought of as a continuous and sustainable source of chemical energy and
64 resources (Frijns *et al.*, 2013). As a result, WWTPs are now considered to be Wastewater Resource
65 Recovery Facilities (WRRF) from which valuable products like chemicals, nutrients (mainly
66 phosphorus, P), bioenergy (methane from anaerobic digestion) and bio-products can be obtained
67 (Keller, 2008, Guest *et al.*, 2009). To make this change possible, the water sector is developing new
68 and innovative treatment technologies, such as energy-efficient nutrient removal or recovery
69 technologies with Anammox, struvite crystallisers, phototropic bacteria, high rate algae systems,
70 sludge pre-treatment processes, or systems for the production of microbial polymers.

71
72 The most immediate step for getting this goal is the updating of existing plants in order to reduce
73 overall operating costs and recover resources. Thanks to the incorporation of new technologies or
74 different plant layouts, energy self-sufficient WWTPs is a feasible goal (Jeppsson *et al.*, 2007).
75 Proof of this comes from the Strass and Wolfgangsee-Ischl WWTPs in Vienna (Wett *et al.*, 2007;
76 Nowak *et al.*, 2011). As stated in the work of Batstone *et al.* (2015), currently there are two
77 extended philosophies to address the transition from WWTPs to WRRF's. One is the low energy
78 mainline (LEM) configuration, which focuses on using low strength anaerobic digestion processes
79 for treating raw domestic sewage, followed by nutrient removal processes (McCarty *et al.*, 2011).
80 The other is the Partition-Release-Recover (PRR) configuration, which focuses on a first stage of
81 COD and nutrient accumulation in the solids, a second stage of release through the digestion
82 process, and a final stage of digestate treatment (Verstraete *et al.*, 2009).

83
84 In the literature there are numerous studies comparing different plant layouts and analysing the
85 energy consumption of conventional WWTPs (Nowak, 2003; Gude, 2015, Tchobanoglous *et al.*,
86 2014, Mininni *et al.*, 2015), and fewer studies analysing advanced WWTPs (Garrido *et al.*, 2013;
87 Batstone *et al.*, 2015; Khiewwijit *et al.*, 2015), many of which use life cycle analysis (LCA)
88 methods and decision support system (DSS) tools (Foley *et al.*, 2010; Garrido-Baserba *et al.*, 2014;
89 Bisinella de Faria, *et al.*, 2015; Castillo *et al.*, 2016). Even so, virtually all these studies are based
90 on operating cost analysis and/or are largely dependent on the quality of the data used and their
91 specifications. However, one of the main problems found in these energy assessments is the limited
92 information available to reproduce disturbances or unusual situations (Jenkins *et al.*, 2014). It is
93 for this reason that the best tool for overcoming all these obstacles is to conduct mass balances for
94 COD, N and P for the whole plant (Spindler & Vanrolleghem, 2012; Jenkins *et al.*, 2014). The
95 detailed analysis of each stream allows a better understanding of the process, identifying areas for
96 improvement and opportunity for resource and energy recovery. Among existing approaches in the
97 literature, the Plant-Wide Modelling (PWM) methodology proposed by Ceit-IK4 (Grau *et al.*, 2007;
98 Fernández-Arévalo *et al.*, 2014; Lizarralde *et al.*, 2015) constitutes a very suitable tool for
99 rigorously and globally assessing the incorporation of new leading-edge technologies in
100 conventional plant layouts (Fernández-Arévalo *et al.*, 2015) or selecting the most appropriate
101 operating strategies at existing full-scale facilities (Fernández-Arévalo *et al.*, 2016).

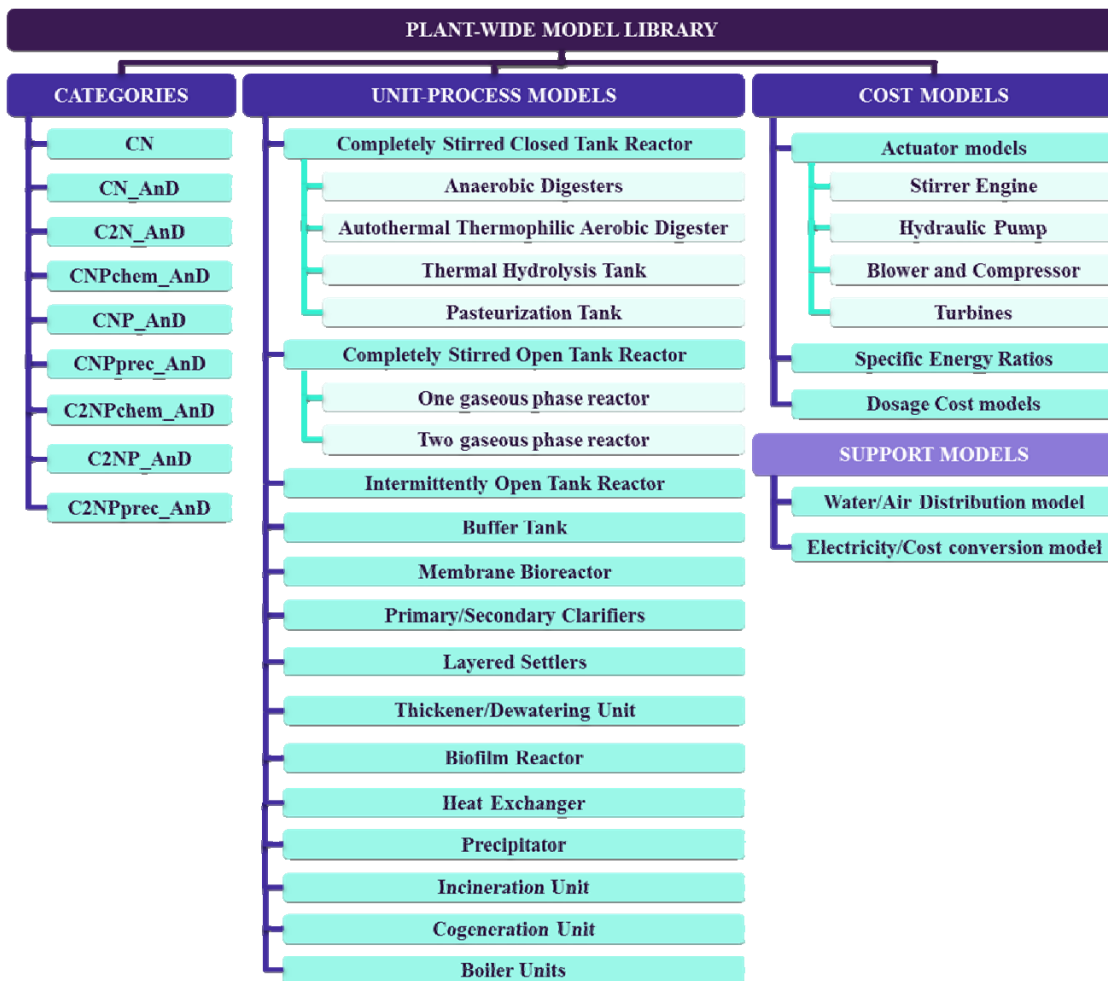
102
103 The main objective of this paper is to conduct a comparative analysis of evolutionary WWTPs,
104 from a techno-economic point of view, analysing in turn organic matter and nutrient energy use and
105 recovery options. To do this, an upgraded plant and a newly designed plant have been analysed and
106 compared against a conventional plant (based on the BSM2 configuration; Jeppsson *et al.*, 2007). In
107 the upgraded or retrofitted WWTP, thermal hydrolysis (TH) technology and a nitrification/Anammox
108 process have been incorporated into the reference plant, and the new plant is a C/N/P decoupling
109 WWTP, which is based on the PRR configuration proposed by Batstone *et al.*, (2015).

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2. MODELLING TOOL: Plant-Wide Model Library

The Plant-Wide Modelling (PWM) methodology (Grau *et al.*, 2007; Fernández-Arévalo *et al.*, 2014; Lizarralde *et al.*, 2015) allows the rigorous and systematic construction of compatible unit-process models (UPM) in order to describe the dynamic behaviour of different processes and technologies in the water and sludge lines in an integrated way. This PWM methodology is based on selecting, from a global list, the set of process transformations required to describe all unit-processes incorporated into each specific WWTP. Thus, the model will be constituted by a unique set of transformations and components vector that will allow the description of all relevant processes occurring in the plant. An accurate definition of the stoichiometry and the enthalpies of formation ensures the elemental mass, charge and energy continuity through the whole plant (Fernández-Arévalo *et al.*, 2014). This methodology allows the straightforward construction of different plant-wide models, which is especially suitable for the comparative assessment of any combination of existing technologies and configurations or those that are under development.

Following the guidelines proposed by Grau *et al.* (2007), Fernández-Arévalo *et al.* (2014) and Lizarralde *et al.* (2015) and with the goal of simplifying plant-wide models construction, this paper presents a new model library (Figure 1). The users can construct their own plant wide models by means the appropriate selection of the category, unit-process models and cost models depending on the case under study (Fernández-Arévalo *et al.*, 2017).



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Figure 1. Schematic representation of the Ceit Plant-Wide Model library.

2.1. Category selection

134 Each category includes equations describing a set of biochemical, chemical and physico-chemical
 135 transformations. Depending on the complexity of the WWTP and the goals and scope of the
 136 modelling study users could select one category or another. The model categories have been
 137 developed combining conventional biological processes described in ASM (Henze *et al.*, 2000) and
 138 ADM (Batstone *et al.*, 2002) models with chemical and physico-chemical processes. All of them
 139 are represented by means of the definition of a stoichiometric matrix and kinetics vector (Grau *et al.*,
 140 2007; Lizarralde *et al.*, 2015). The nomenclature used to define the categories is as follows:
 141 “C”, “N” and “P” describe biological organic matter biodegradation, N biodegradation, and
 142 biological and chemical P removal, respectively, all of them in aerobic and anoxic conditions at low
 143 and high temperatures (TH reactions); “2N” specifies two-step N removal and Anammox reactions;
 144 “chem” denotes chemical P but not biological P removal; “prec” includes precipitation reactions;
 145 and finally, “AnD” describes anaerobic conditions at low and high temperatures (fermentation and
 146 digestion).

147
 148 The organized structure that the methodology presents enables the straightforward development of
 149 categories, allowing the library to be continuously updated.

151 2.2. Unit-process model selection

152 The Ceit PWM library contains a set of unit-process models that describes the mass and energy
 153 transport in each unit. According to the features of novel technologies and processes analysed in
 154 advanced WWTPs, these models consider in most of the cases aqueous, gas and solid phases and
 155 mass and energy exchange among them (Fernández-Arévalo *et al.*, 2014; Lizarralde *et al.*, 2015).

157 2.3. Cost model selection

158 Lastly, the library includes a set of actuator models, specific energy ratios and dosage cost models
 159 in order to estimate in detail the costs of each element (see Table 1). All actuator models are
 160 developed based on engineering expressions instead of directly using cost curves or fixed values.
 161 The models are standardized, so they can be used interchangeably in any category.

162
 163 **Table 1.** Description of cost models (Actuators, Specific Costs and Dosage Cost models)

| Actuator Models | Equations |
|-----------------------------|--|
| Stirrer Engine Model | <p>For maintaining solids in suspension</p> $W_{stir} = \frac{G^2 \eta_w V_w}{\rho_w}$ <p>For rapid mixing or flocculation</p> $W_{stir} = G^2 \eta_w V_w$ |
| Hydraulic pump model | $W_{pump} = \rho_w g Q_w H_L \eta_{pump}$ $H_{wout} = H_{win} + W_{pump} (1 - \eta_{pump})$ |
| Blower and Compressor model | $W_{blow} = \sum_{comp=1}^m \left[\frac{(\dot{m}_{gin})_{comp} R T_{gin}}{(MW)_{comp} \left(\frac{V_{g,comp} - 1}{V_{g,comp}} \right) \eta_{blow}} \right] \left[\left(\frac{P_{gout}}{P_{gin}} \right)^{\frac{V_{g,comp} - 1}{V_{g,comp}}} - 1 \right]$ $T_{gout} = \frac{T_{gin}}{\eta_{blow}} \sum_{comp=1}^m \left(\frac{(\dot{m}_{gin})_{comp}}{\sum_{comp=1}^m (\dot{m}_{gin})_{comp}} \left(\frac{P_{gout}}{P_{gin}} \right)^{\frac{V_{g,comp} - 1}{V_{g,comp}}} - (1 - \eta_{blow}) \right)$ |

| Actuator Models | Equations |
|--|---|
| Turbine model | $W_{\text{turbine}} = \sum_{\text{comp}=1}^m \left[\frac{(\dot{m}_{g,\text{in}})_{\text{comp}} R T_{g,\text{in}}}{(MW)_{\text{comp}} \left(\frac{V_{g,\text{comp}} - 1}{V_{g,\text{comp}}} \right)} \right] \left[\eta_{\text{turb}} \cdot \left(\frac{P_{g,\text{in}}}{P_{g,\text{out}}} \right)^{\frac{1 - \gamma_{g,\text{comp}}}{\gamma_{g,\text{comp}}}} \right]$ $T_{g,\text{out}} = T_{g,\text{in}} \sum_{\text{comp}=1}^m \left(\frac{(\dot{m}_{g,\text{in}})_{\text{comp}}}{\sum_{\text{comp}=1}^m (\dot{m}_{g,\text{in}})_{\text{comp}}} \left(\frac{P_{g,\text{out}}}{P_{g,\text{in}}} \right)^{\frac{V_{g,\text{comp}} - 1}{V_{g,\text{comp}}}} + (1 - \eta_{\text{turb}}) \right)$ |
| Support Models | Equations |
| Water/Air Distribution model | <p>Detailed Model</p> $HL = HL_s + HL_f + HL_l; \quad HL_f = f_{\text{moody}} \left(\frac{L_{\text{pipe}}}{D_{\text{pipe}}} \right) \left(\frac{u_w^2}{2g} \right)$ <p>Approximation</p> $P_{g,\text{out}} = \phi_w g (\text{Submergence} + 1) 10^{-5}$ |
| Electricity/cost conversion model | $\text{Cost}_{\text{actuator}} = W_{\text{actuator}} \text{ MU}$ |
| Dosage Cost Models | Equations |
| Chemically Enhanced Primary Treatment (CEPT) | $\text{Cost}_{\text{dosage}} = \text{Cost}_{\text{chem}} \left[\frac{\text{€}}{\text{kg}} \right] \cdot \left(\frac{k_{\text{chem}}^{n_{\text{CEPT}}} \left(1 - \frac{\eta_{\text{max}} \cdot \eta_{\text{CEPT}}}{\eta_{\text{max}} \cdot \eta_{\text{min}}} \right)}{\frac{\eta_{\text{max}} \cdot \eta_{\text{CEPT}}}{\eta_{\text{max}} \cdot \eta_{\text{min}}}} \right)^{1/n_{\text{CEPT}}}$ |
| Poly-electrolyte dosage costs | $\text{Cost}_{\text{dosage}} = \text{Cost}_{\text{poly}} \left[\frac{\text{€}}{\text{kg}} \right] \cdot \sum_{j=1}^{\text{N}^{\circ} \text{ of kinds of sludge}} \left(\text{TSS}_j Q_w k_{\text{poly},j} \left[\frac{\text{g poly}}{\text{kg TSS}} \right] \right)$ |

164 (Camp and Stein, 1943; CEDEX, 2004; Tchobanoglous *et al.*, 2014; Tik *et al.*, 2013; Weisbach, 1845; Zaher *et al.*, 2009; Zweitering,
165 1985)

167 3. DESCRIPTION OF THE SCENARIOS

168 The comparative analysis of the three configurations selected (conventional WWTP, upgraded
169 WWTP and C/N/P decoupling WWTP) has been based on PWM simulations. This section details
170 the description of these plant layouts and the steps followed to build the model.

172 3.1. Plant layouts definition

173 *Conventional WWTP*

174 The reference WWTP considered in the study is a biological nutrient removal plant based on the
175 Benchmark Simulation Model No. 2 (BSM2; Jeppsson *et al.*, 2007). The plant layout (Figure 3) is
176 constituted by a primary clarifier for the pre-treatment step, an AS process for C and N removal
177 based on a Modified Ludzack-Ettinger configuration (2 anoxic and 3 aerobic tanks and a secondary
178 clarifier), a dissolved air flotation (DAF) unit, an anaerobic digestion process and a dewatering step.
179 A ferric chloride dosage is delivered to the output from the third aerobic tank for chemical P
180 removal. Besides adding the chemical agents for P removal, ferric chloride can also be added to
181 enhance the settling characteristics of the primary sludge for cases in which the production of
182 primary sludge needs to be maximized. Finally, two other chemical additions are required in
183 flotation and dewatering processes.

185 *Upgraded WWTP*

186 This second layout is based on the reference case (conventional WWTP), but with two advanced
187 technologies being incorporated in the sludge line: a TH reactor and a nitrification/Anammox process
188 for treating the rejected supernatants (Figure 4).

189

249 describe the major costs of the system the following models were selected: blowers, pumps, stirrer
 250 engines, gas and water distribution systems, specific energy ratios, dosage costs and electricity/cost
 251 conversion models. The dehydration process is described from specific energy ratios and dosage
 252 costs, and in the case of flotation the process is described by specific energy ratios and pumping,
 253 aeration and dosage costs. All these cost models were calibrated from standard engineering values.
 254

255 4. SIMULATION ANALYSIS: ENERGY AND NUTRIENT MANAGEMENT 256 EXPLORATION

257 Once the models for the three scenarios proposed were constructed, (steady state) simulations were
 258 carried out to analyse the potential use of the energy contained in the organic matter and nutrients
 259 and/or its recovery. Furthermore, a cost analysis of these plants for different influent C/N/P ratios
 260 was carried out. To avoid possible interferences from other factors that affect plant operation (plant
 261 oversizing, unit-process efficiencies, environmental factors, etc.), reactor volumes and recycle flows
 262 have been optimized for each plant layout and for each influent in order to fulfil a fixed effluent
 263 quality of 10gN m⁻³, 1gP m⁻³, 125gCOD m⁻³ and 35gSS m⁻³, accordingly to the European Directive
 264 91/271/EEC.
 265

266 4.1. General considerations about the potential use of the energy contained in COD and 267 nutrient and/or its recovery options in WWT processes

268 In analysing the different conventional options of getting energy from COD removal, the most
 269 effective and typical way is to transform the organic matter into CH₄ (the $\Delta h_{r,CH_4}$ is 13.91 kJ gCOD⁻¹
 270 or 890 kJ mol⁻¹) and use its combustion to produce thermal and electric energy. The anaerobic
 271 COD biodegradation presents three advantages against aerobic or anoxic oxidation in AS: (1) the
 272 heat of reaction is higher, 11.86-13.37 kJ gCOD_{rem}⁻¹ against 4.07 kJ gCOD_{rem}⁻¹-5.27 kJ gCOD_{rem}⁻¹
 273 for aerobic and 3.52 kJ gCOD_{rem}⁻¹- 4.74 kJ gCOD_{rem}⁻¹ for anoxic biodegradation (calculated from
 274 Table 2), (2) the energy recovery from biogas combustion is more effective because of the heat
 275 dissipated when oxidation occurs in the aqueous phase, and (3) aeration costs are reduced. Thus, the
 276 clearest alternative to maximize the recovery or reuse of the COD energy potential is to minimize
 277 the COD oxidation in AS processes. This can be obtained by producing more primary sludge and
 278 working at lower SRT in the secondary biological treatment.
 279

280 **Table 2** – Specific heat yields (or energy content) estimated with the PWM methodology

| Transformations [†] | S _{SU} | S _{AA} | S _{FA} | S _{HVA} | S _{HBU} | S _{HPRO} | S _{HAC} |
|---|-----------------|-----------------|-----------------|------------------|------------------|-------------------|------------------|
| Aerobic COD removal heat (kJ gCOD _{rem} ⁻¹) | -5.27 | -4.41 | -4.07 | -4.14 | -4.17 | -4.19 | -4.32 |
| Aerobic COD removal heat (kJ mol _{rem} ⁻¹) | -1011 | -2997 | -2997 | -862 | -667 | -469 | -276 |
| Anoxic COD removal heat (kJ gCOD _{rem} ⁻¹) | -4.74 | -3.89 | -3.55 | -3.52 | -3.64 | -3.67 | -3.79 |
| Anoxic COD removal heat (kJ mol _{rem} ⁻¹) | -910 | -520 | -2610 | -753 | -583 | -410 | -243 |
| Anaerobic COD removal heat (kJ gCOD _{rem} ⁻¹) [‡] | -1.01 | -0.17 | 0.14 | 0.08 | 0.04 | 0.00 | -0.15 |
| Anaerobic COD removal heat (kJ mol _{rem} ⁻¹) [‡] | -195 | -22 | 101 | 17 | 9 | 1 | -25 |

281 [†] CO₂ stripping is not included in the specific heat yield estimations, but the COD used for biomass growth is included.

282 [‡] The estimation of the reaction heat includes the acidogenesis, acetogenesis and methanogenesis reactions.

283
 284 Although, as it's been abovementioned, the energy recover from compounds is more efficient when
 285 they are in a gaseous phase. In the case of the ammonia, its solubility in water is very high,
 286 necessitating stripping methods for transferring from water into gas phase. This, added to the fact
 287 that ammonia requires a catalyst for its oxidation in gas phase (Jones *et al.*, 1999), makes this
 288 process economically unfeasible. Moreover, the nitrogen recovery techniques (ion exchange
 289 methods or stripping processes) consume more energy than removal processes, with the exception
 290 of struvite recovery technologies. Consequently, from an economic perspective, the destruction of
 291 nitrogen compounds appears the most logical route (Matassa *et al.*, 2016) and low-energy
 292 alternatives can be proposed, such as the use of Anammox bacteria, anaerobic phototropic bacteria
 293 or high-rate algae (Batstone and Virdis, 2014). Comparing the N oxidation reactions in the aqueous

294 phase, the Anammox reaction is the one that release more energy to the medium ($23.32 \text{ kJ gN}_{\text{rem}}^{-1}$),
295 followed by nitrification ($15.46 \text{ kJ gN}_{\text{rem}}^{-1}$) and nitrification ($6.09 \text{ kJ gN}_{\text{rem}}^{-1}$) reactions. Thus,
296 Nitrification/Anammox reactions maximize the energy utilization of the N and minimize oxygen
297 consumption in the process that leads to a reduction in the aeration costs.

298
299 Finally, the scarcity of natural phosphorus resources converts the recovery of P into the first
300 alternative for use. Currently, P recovery methods from MWW include the agricultural use of
301 sludge, production of struvite, particularly in enhanced biological P removal (EBPR) plants, and the
302 recovery of P from ash (Wilfert *et al.*, 2015). On the other hand, phosphorus removal itself, using
303 biological or chemical processes does not exert any effect on energy balances.

304 305 **4.2. Analysis of the energy use of a conventional wastewater treatment plant**

306 To analyse the degree of utilization of thermal energy content (energy associated with the fluid
307 temperature) and mass energy content (energy associated with the composition of water), a global
308 plant-wide simulation of a conventional plant was carried out under steady-state conditions for a
309 critical temperature of 13°C .

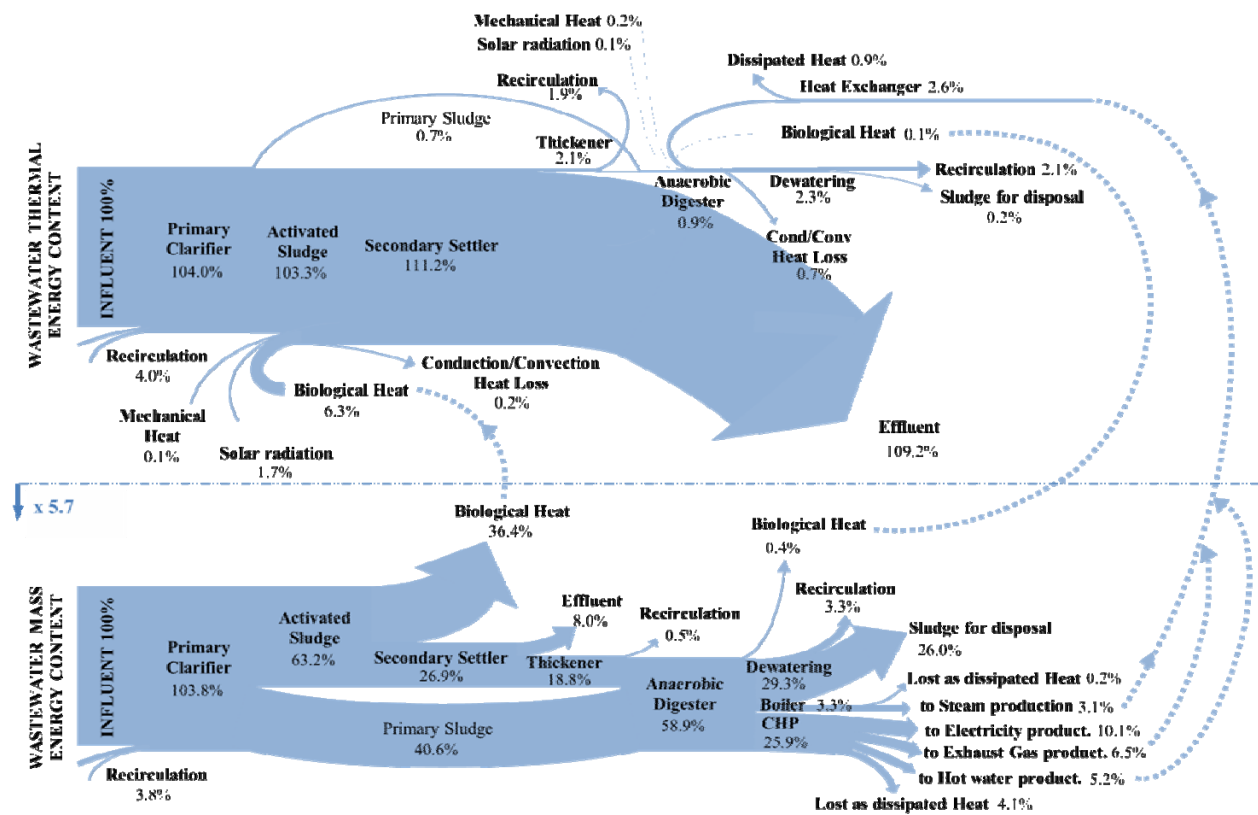
310
311 Based on the mass and energy fluxes proposed by Pagilla *et al.* (2004), Figure 6 shows the
312 maximum energy potential of the wastewater in each point of the plant. The top of the figure shows
313 the total thermal energy or enthalpy (not exergy) associated with temperature, while the bottom part
314 reflects the maximum energy potential of the constituents in the water, that is the energy released
315 upon oxidation of all water components to CO_2 , H_2O , NO_3 and H_3PO_4 .

316
317 As shown in Figure 6, the biological heat and the solar and atmospheric radiations increased the
318 temperature of the aqueous phase by 0.5-2 degrees (1.5°C for this case study) and the thermal
319 energy output of the plant by 10% (energy loss through the effluent). Heat recovery technologies
320 (Wanner *et al.*, 2005; Corbala-Robles *et al.*, 2016) could be an appropriate solution for taking
321 advantage of this thermal energy. However, the obtained heat ($55\text{-}75^\circ\text{C}$, Alekseiko *et al.*, 2014) is a
322 very low exergy stream and its application is limited to use in the plant itself or in WWTPs located
323 near a residential area or near hot water demanding areas (IWA Resource Recovery Cluster, 2015).
324 In spite of this, its high coefficient of performance (COP or the ratio of heating provided to work
325 required), which is between 1.77 and 10.63, makes it a promising technology (Hepbasli *et al.*,
326 2014).

327
328 Simulation results show that a considerable fraction of the mass energy content is released to the
329 atmosphere or aqueous phase as biological heat (35-40%) due to the transformations that occurred
330 in the system. Among these transformations, nitrification reactions brought more specific energy to
331 the system ($21.61 \text{ kJ gN}_{\text{rem}}^{-1}$), followed by the COD oxidation reactions (Table 1). Around 30% of
332 the mass energy content is converted into biogas and goes to the CHP unit. In this specific case, for
333 a mass flow of $7.7 \text{ t}_{\text{COD}} \text{ d}^{-1}$ ($42.6 \text{ kg}_{\text{COD}} \text{ m}^{-3}$) entering to the anaerobic digester and a temperature of
334 13°C , it is not possible to maintain the mesophilic temperature, and 3-5% of COD is addressed to
335 the boiler, reducing the electrical energy production. Consequently, at this particular case, only 10%
336 of the mass energy content in MWW is converted into electricity, losing the remaining energy by
337 heat dissipation (4%), through the effluent (8% mass content and 37% thermal content), through the
338 sludge (26%), and digester heating (15%). From the analysis of Figure 6, it can be said that in a
339 conventional plant, most of the influent energy potential is lost as heat (digester heating and
340 oxidation reactions).

341
342 Points highlighted in sections 4.1 and 4.2, shows that a rigorous energy and mass flow analysis is
343 crucial for assessing the potentiality of the plant in terms of (1) energy use and recovery and (2)

344 valuable recovered products production. According to this, section 4.3 will show as an example, a
 345 simulation-based analysis of the three plant layouts presented in section 3 and for different
 346 operating scenarios.
 347



348 **Figure 6.** Simulation of the wastewater mass and energy content distribution throughout the conventional WWTP.

349

350 4.3. Comparative analysis of COD and nutrient (N/P) flux distributions in a conventional, 351 upgraded and C/N/P decoupling WWTP

352 In order to analyse the potential of the wastewater mass energy content, a set of PWM simulations
 353 has been carried out. To that end, the distribution of COD, N and P flows throughout the plants was
 354 assessed in the traditional, upgraded and C/N/P decoupling plant under stationary conditions for a
 355 temperature of 18°C. At this temperature all the biogas is conducted to CHP in the three case-
 356 studies, which simplifies the comparison.
 357

358 **Conventional WWTP**

359 As discussed in the previous section, the way to use the maximum energy content of COD is to
 360 convert this organic matter into CH₄. Although in the conventional plant analysis (Figures 7, 8 and
 361 9), only 29% of the influent COD is transformed into biogas, this value turns into 43% if the
 362 influent non-biodegradable organic matter (20% of the COD (S_I, X_I)) and the non-biodegradable
 363 fraction produced in the plant (12% (S_P, X_P)) are not considered. Given this, it's clear that
 364 maximizing the biogas generation by (1) producing more primary and secondary sludge and (2)
 365 transforming part of the non-biodegradable organic matter into biodegradable, for example,
 366 by using mechanical (ultrasound treatments, high-pressure homogenisation), thermal (thermal
 367 hydrolysis), chemical (ozonation, Alkali treatments) or biological alternatives (Pérez-Elvira *et al.*,
 368 2006), could improve significantly the organic matter energetic potential.
 369

370 Regarding the total N (TN) balance of this study, 58% of the N is denitrified, 17% and 25% are
 371 extracted from the effluent and dewatered sludge, respectively, and 25% of the N is recirculated

372 back to the water line as $\text{NH}_x\text{-N}$. The N percentage extracted from this dewatered sludge is not a
373 fixed value and it is closely related to the degree of volatile solids (VS) removed in the anaerobic
374 digestion process. The volatile solids removal efficiency is approximately proportional to the degree
375 of $\text{NH}_x\text{-N}$ released. In this case, for a VS removal of 51%, a formation of 51% $\text{NH}_x\text{-N}$ with respect
376 to the TN feed to the digester has been observed.
377

378 Finally, as previously mentioned, P is a component that is extracted from the plant only within the
379 effluent and sludge. Thus, the flow of total P (TP) in the dewatered sludge (80% in this case)
380 depends on the P concentration in the influent and effluent. For a high P load influent (25gP/m^3 ,
381 Henze *et al*, 2008) the percentage of TP extracted as solids can be 92-96%, while for a low P load
382 influent (6gP/m^3 , Henze *et al*, 2008) it can be about 60-80%, which is in accordance with our
383 calculations (Figure 9).
384

385 ***Upgraded WWTP***

386 In this case, incorporating the thermal hydrolysis technology, allows the secondary sludge
387 biodegradability to be increased (by 40% in this particular study), thus converting the non-
388 biodegradable matter, X_P , into biodegradable matter (X_{CH} , X_{PR} , X_{LI}) and consequently increasing
389 biogas production (by 27% in this particular case study, and by 40% when only secondary sludge is
390 digested). This production depends mainly on the proportions of primary and secondary sludge fed
391 to the digester. The extra amount of COD transformed into methane is approximately the same as
392 the amount by which COD decreased in the dewatered sludge, in this case the extracted COD was
393 reduced by 19% and the sludge produced by 12% (as a function of the VSS/TSS ratio). The
394 degradation of this new fraction of biodegradable organic matter (part of X_P) will release 25% more
395 $\text{NH}_x\text{-N}$ and 23% more ortho-P in the digested sludge, thereby decreasing the content of TN and TP
396 in dewatered sludge and increasing the content of $\text{NH}_x\text{-N}$ and ortho-P slightly (by 25% and 23%,
397 respectively). Thus, the resulting reject water will contribute to an increase in the N load to be
398 treated in the AS process by up to 30%. This can be a problem if the biological plant does not have
399 sufficient capacity to treat this additional nitrogen load. Thus, before incorporating any technology,
400 it is useful to analyse its repercussions and viability in the plant as a whole.
401

402 With the inclusion of the nitrification/Anammox process the total N flux in the reject water stream
403 was reduced by 70% and the $\text{NH}_x\text{-N}$ flux by 92%, decreasing in turn the $\text{NH}_x\text{-N}$ to be treated in the
404 AS process by 28%. By using either energy-efficient technologies (nitrification/Anammox) or
405 conventional N removal technologies (denitrification-nitrification processes), the N gas released to
406 the atmosphere is similar in both cases (58%). In this plant layout, due to the pre-treatment
407 incorporated (TH), the plant has to treat more $\text{NH}_x\text{-N}$ or more biodegradable nitrogen. This results
408 in increased amounts of nitrogen lost by stripping (64%).
409

410 The release of these extra nutrients can increase the probability of uncontrolled precipitation of salts
411 (struvite, calcium ortho-P, etc.), if the concentration of ions (Mg^{++} , Ca^{++} , etc.) is considerable and if
412 the process conditions favour them. Thus, although the plant does not have biological P removal,
413 the P released in the digestion can be enough to generate uncontrolled precipitation problems.
414

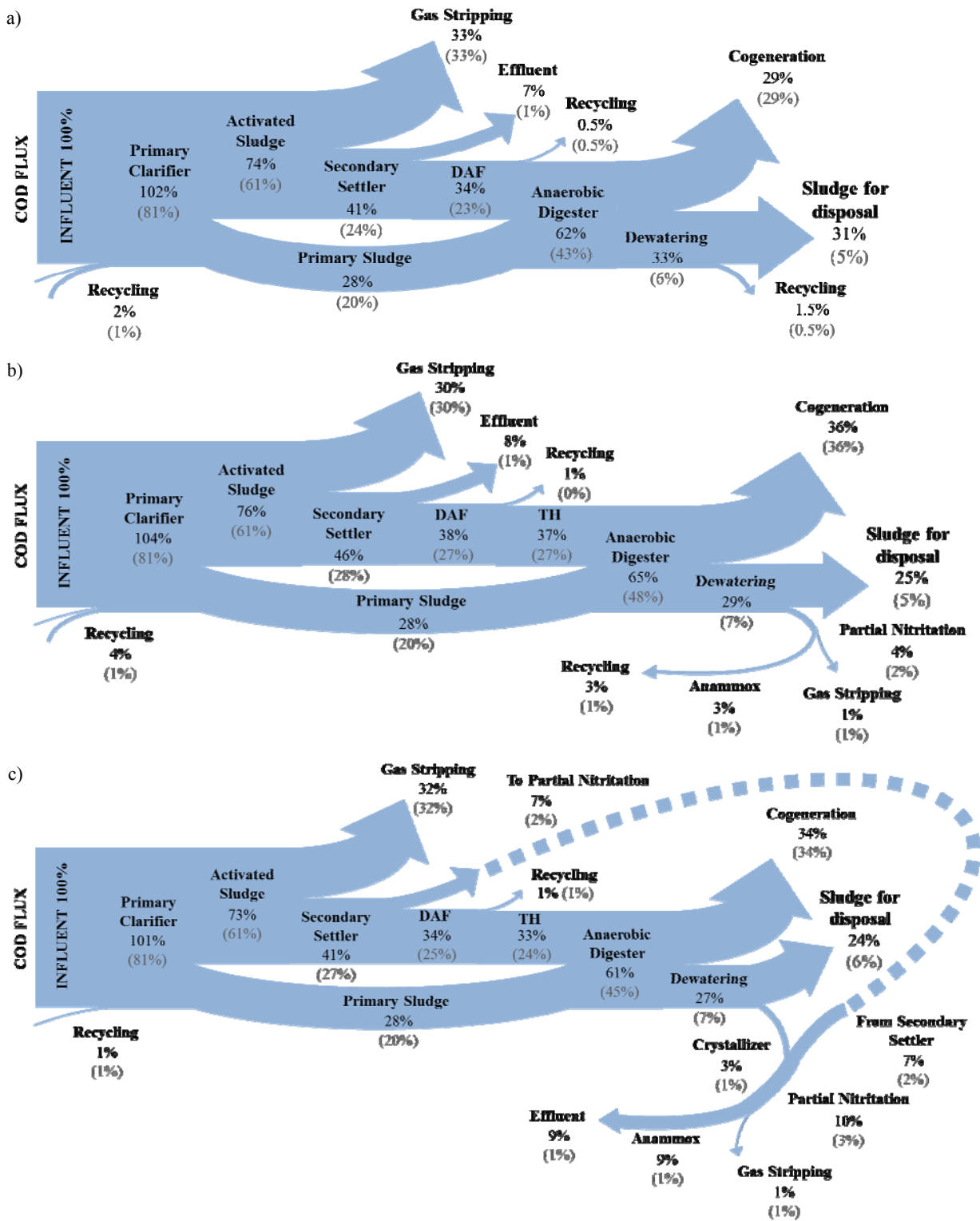
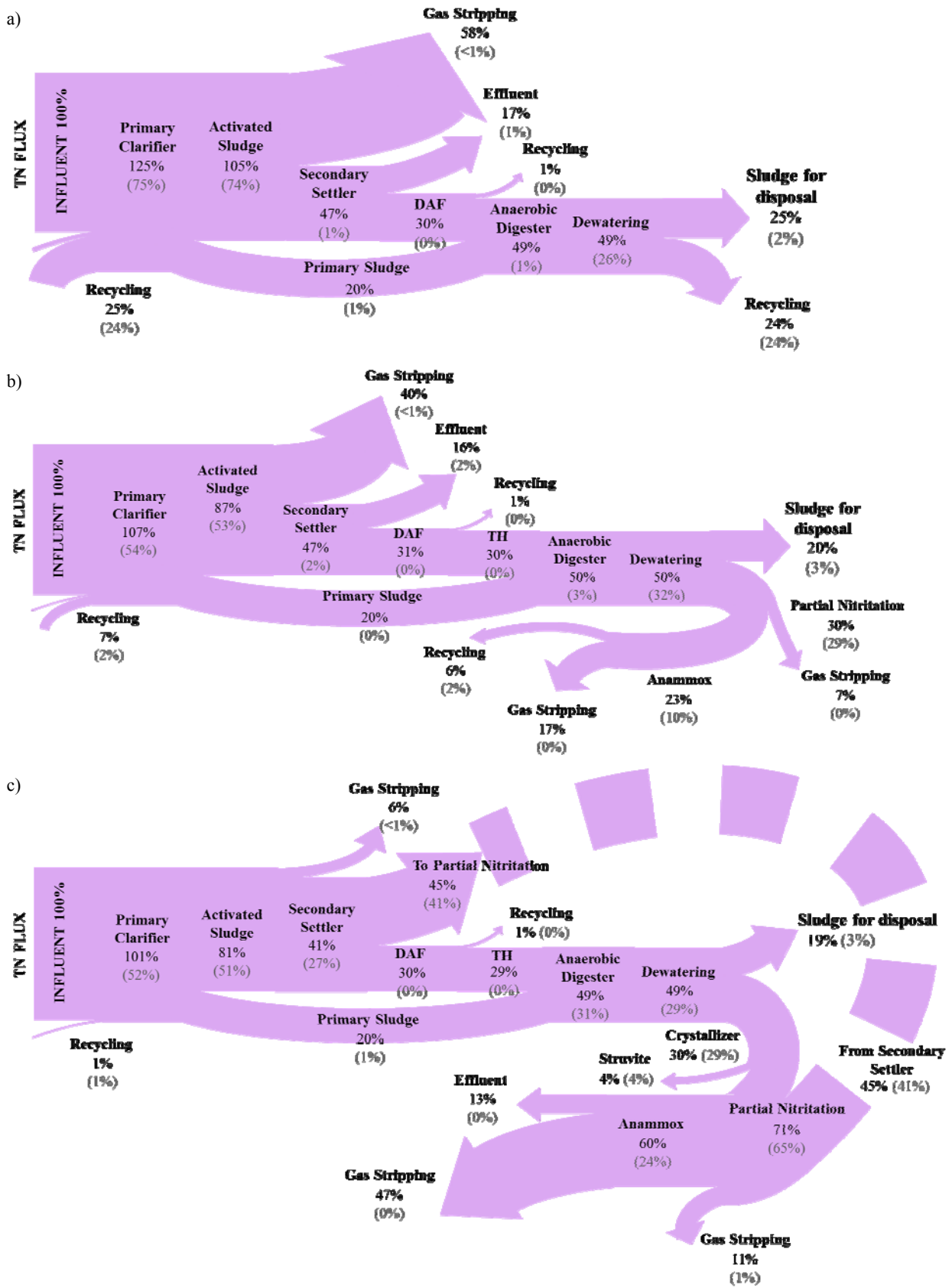
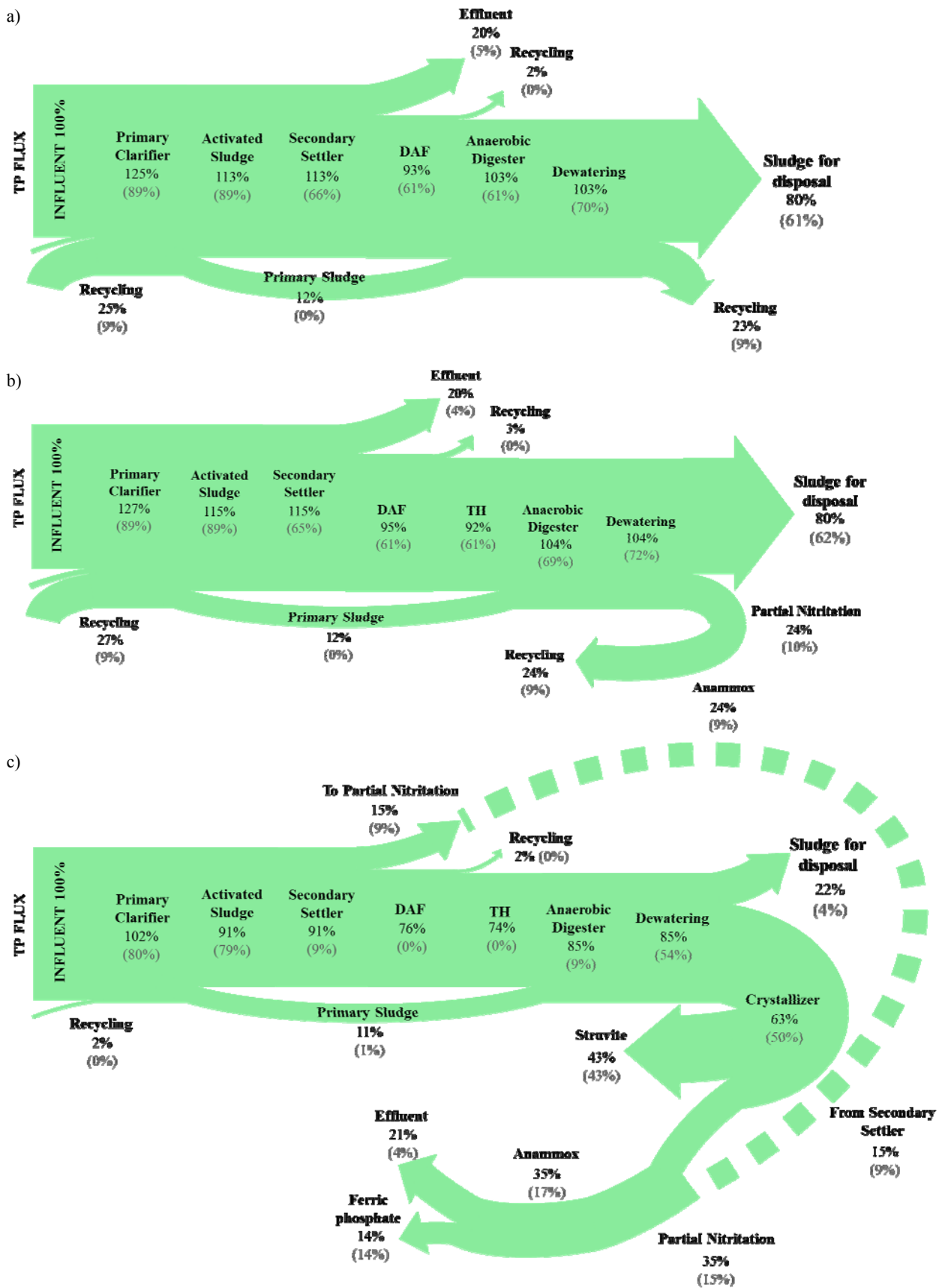


Figure 7. Simulation of the total COD and (biodegradable COD) flux distributions throughout: (a) a conventional WWTP, (b) an upgraded WWTP, and (c) a C/N/P decoupling WWTP

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418 **Figure 8.** Simulation of the TN and (NH_x-N) flux distributions throughout: (a) a conventional WWTP, (b) an upgraded
 419 WWTP, and (c) a C/N/P decoupling WWTP
 420



421 **Figure 9.** Simulation of the TP and ortho-P flux distributions throughout: (a) a conventional WWTP, (b) an upgraded
 422 WWTP, and (c) a C/N/P decoupling WWTP

423
 424 *New WRRF concept: C/N/P decoupling WWTP*

425 This new treatment concept consists of treating each compound (organic matter, N and P) in the
426 most efficient way possible, promoting recovery and maximizing energy use: organic matter is
427 valorised as biogas, the P is recovered as struvite and the N is treated with energy-efficient
428 technologies.

429
430 By working at low solids retention time of 3 days (to avoid nitrification and an excessive
431 accumulation of inerts), the production of non-biodegradable organic matter, due to decay
432 processes, is lower (12% lower than in the conventional configuration), but the same amount of
433 CO₂ is produced due to acidogenesis, PAO bacteria growth and polyP storage reactions. Therefore,
434 this plant layout should not be used with the goal of increasing biogas production. Once again, to
435 increase the biodegradability of the sludge, a thermal hydrolysis unit was introduced to the global
436 plant configuration, obtaining in this case a 21% increase in biogas production. If the objective had
437 been to only maximize the production of biogas, without paying attention to the removal and
438 recovery of P, the configuration could have been modified to a high load fully aerated configuration
439 (without anaerobic reactors to accumulate the P), and in that case, biogas production would have
440 increased up to 40% (about 20%-25% due to the thermal hydrolysis and another 15%-20% due to
441 the high-rate process).

442
443 In the anaerobic digestion process the ortho-P accumulated in PAO bacteria is released, along with
444 the ortho-P previously released into the TH process. Unlike a configuration without biological P
445 removal, in which the percentage of ortho-P at the outlet of digestion is 9-11% of the TP influent, in
446 a configuration with P accumulation this percentage can increase up to 54%. Thus, the dewatered
447 sludge will contain 72% less P, but a greater amount of ortho-P. The recovery of this ortho-P can be
448 accomplished by recovery in crystallization units. The percentage of P recovered depend on factors
449 such as the influent P and ions (uncontrolled precipitation problems) composition, the required
450 effluent quality, the P accumulation efficiency of AS processes, the need for chemical agents in the
451 water line (FeCl₃) and the efficiency of VS removal in digestion, among other things, making it
452 possible to recover 43% of P as struvite.

453
454 Finally, a large proportion of the influent N (79%) will be treated with efficient technologies, since
455 N fluxes recovered as struvite and released by stripping into the AS process were minimal (in this
456 particular case by 4% and 6%, respectively).

457
458 The mass flow analysis allows tracking of components throughout the plant. In a WRRF concept,
459 these components are associated with a source of valuable products and bioenergy. So indirectly, it
460 is an analysis of the plant recovery potential, and consequently an analysis of the plant efficiency. In
461 this frame, the aim of the last section was the estimation of the treatment costs associated with these
462 streams and the quantification of the energy produced or recovered.

463
464 **4.4. Analysis of the costs distributions in a conventional, upgraded and C/N/P decoupling**
465 **WWTP for different influent COD/N/P ratios**

466 The most influential factors on WWTP operating costs are the plant layout and the composition of
467 the MWW influent. In order to analyse the effect of these factors, a global economic analysis of
468 each plant layout was carried out for different influent COD/TN ratios (Table 3), under stationary
469 conditions and for a temperature of 18°C. The ratio TN/TP has been maintained constant. Reactor
470 volumes and operational set-points have been optimized for each particular plant layout and for
471 each influent composition.

472
473 **Table 3** – C/N ratios considered for the influent characterization
474

| Low C | Medium C | High C |
|-------|----------|--------|
|-------|----------|--------|

| | | 444 gCOD m ⁻³ | 592 gCOD m ⁻³ | 740 gCOD m ⁻³ |
|----------------|-----------------------|--------------------------|--------------------------|--------------------------|
| Low TN (LN) | 43 gN m ⁻³ | COD/TN = 10.3 | COD/TN = 13.8 | COD/TN = 17.2 |
| Medium TN (MN) | 57 gN m ⁻³ | COD/TN = 7.8 | COD/TN = 10.4 | COD/TN = 13.0 |
| High TN (HN) | 71 gN m ⁻³ | COD/TN = 6.3 | COD/TN = 8.3 | COD/TN = 10.4 |

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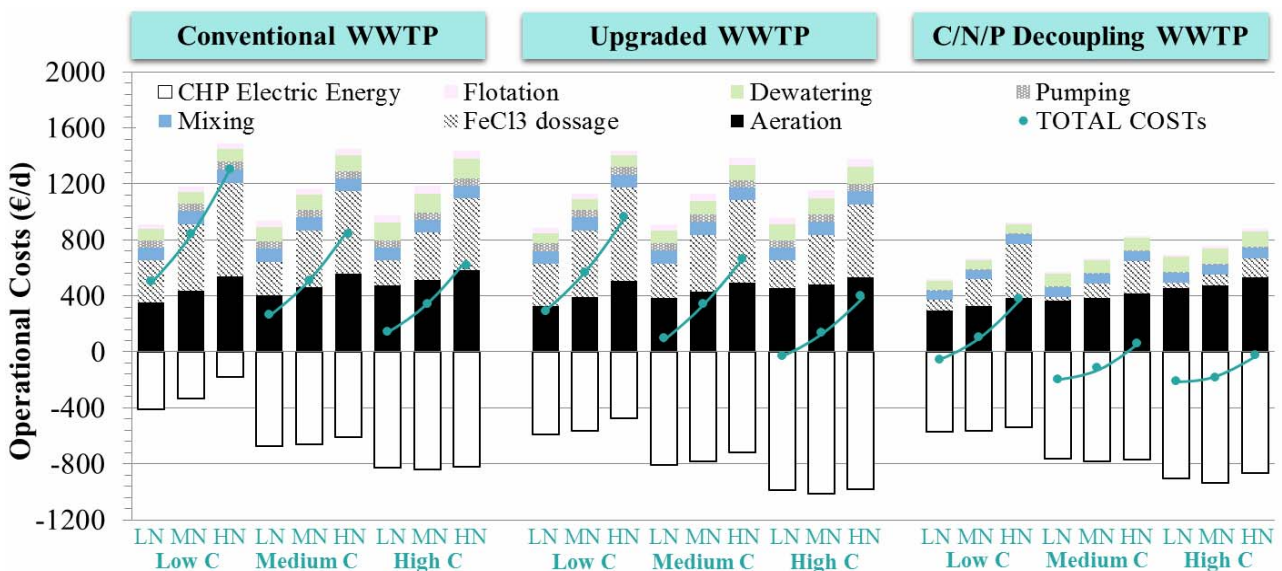
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Figure 10 summarizes the results obtained in all these optimizations. The operating cost distributions of each plant and for each influent are represented by the bars, the CHP electric energy recovery has been included as a negative cost, while the net cost is represented by blue dots. A first analysis of the cost distribution shows that positive operating costs are very similar for the conventional and upgraded WWTP, while the C/N/P decoupling WWTP reduces the expenses significantly. For all configurations, these operating costs are mainly associated with influent N concentration and show a low dependence to the variations of influent C load. In the upgraded and C/N/P decoupling plants, negative operational costs (energy recovery) are increased, due to a more efficient use of the influent COD. Contrarily to the positive costs, energy recovery is mainly associated with C load and exhibits a very low dependence with the N concentration in the influent (except for the critical case of very low C/N ratio in the conventional plant). Finally, total costs are clearly positive in a conventional plant, while the upgraded configurations could theoretically get a neutral cost balance only for high C/N load ratios. However, the C/N/P decoupling plant has a real potential for obtaining a negative cost balance for a broad range of influent characteristics.



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Figure 10. Operating cost analysis in a conventional, upgraded and C/N/P decoupled WWTP for different COD/TN ratios: Cost distribution in columns and net operating costs represented by the blue dots (€/d).

Figure 11 shows the effect of influent concentrations on the most representative costs (aeration and dosage costs and electricity production) and on the plant self-sufficiency (%) for the three plant-layers under study.

The aeration costs exhibits a logical growing trend in the three configurations for increasing N and COD loads (Figure 11a). It is also remarkable the very limited influence of N load to the aeration power in the C/N/P Decoupling WWTP, reflecting the high efficiency of this advanced configuration for the removal of N. The Upgraded plant has incorporated a Nitritation/Anammox process to treat rejected supernatants, reducing overall aeration costs around 6-15% without sludge pre-treatment processes, and somewhat lower, at around 3-11%, when a thermal hydrolysis is incorporated. For the C/N/P Decoupling plant layout, aeration savings of 16-29% are achieved for low-medium COD loads and savings of 4-8% for higher concentrations.

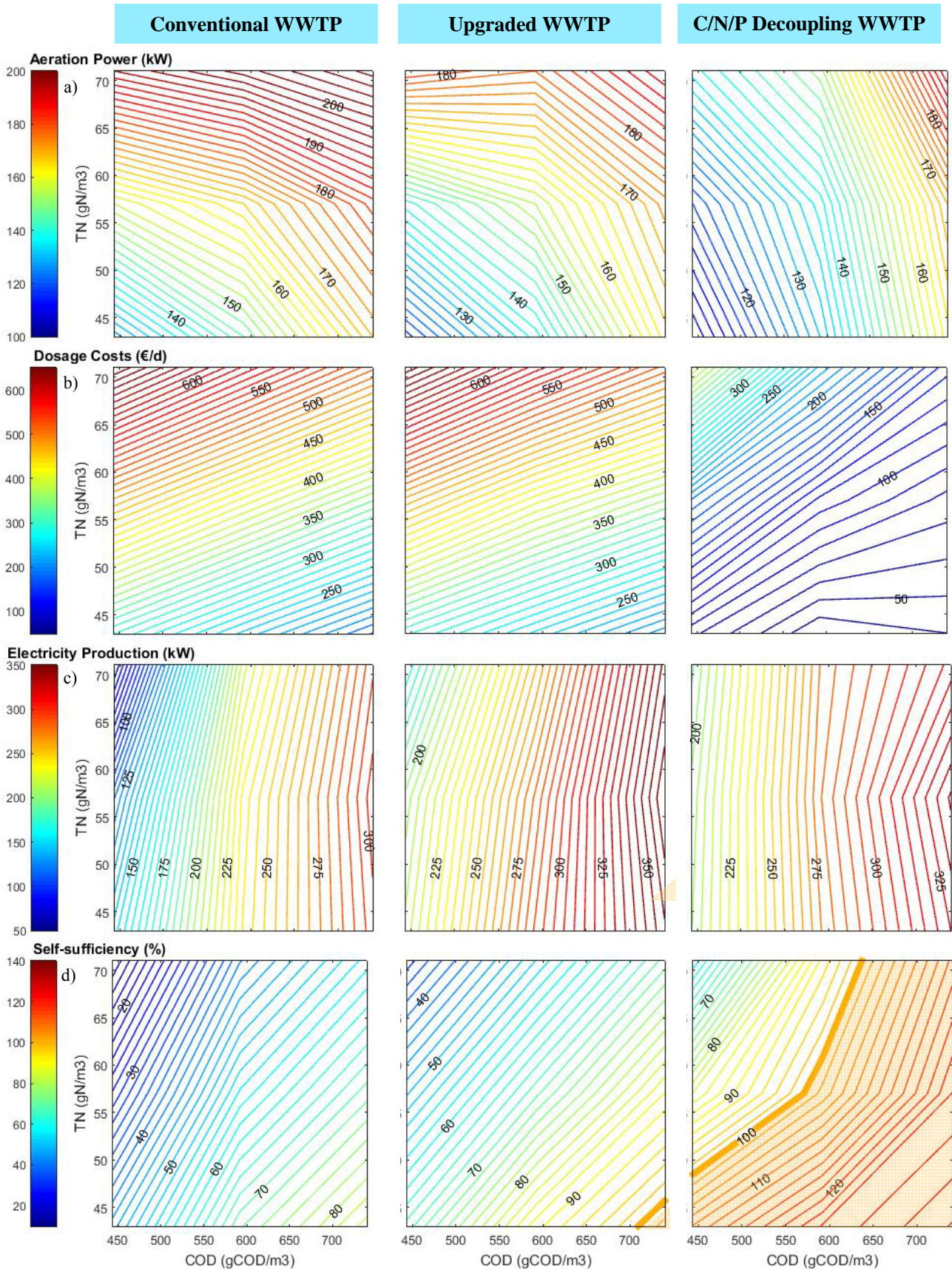


Figure 11. (a) Aeration Power, (b) dosage costs, (c) electricity production, and (d) plant self-sufficiency in a conventional, upgraded and C/N/P decoupled WWTP for different COD/TN ratios

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In dosage costs, a similar trend has been also found in the three configurations. Ferric chloride

511 dosage depends directly on the influent P content, but indirectly on the C/N ratio. In the first two
512 configurations dosage costs are similar, since in both configurations ferric chloride is used to
513 remove all the phosphorus. The third configuration in turn provides savings in chemical reagents,
514 78-80% for high C/N ratios and savings of 42-61% for low ratios. Phosphorus removal is performed
515 through biological reactions, and chemical agents are only used to adjust the water line effluent and
516 the rejected water (after recovering 85% of ortho-P as struvite) to effluent quality standards. In
517 addition to the significant reduction in operating costs this configuration provides a value-added
518 product such as the struvite. The analysis didn't consider the costs of production of struvite, but
519 neither the profit after its sale. It was considered a neutral balance. Still, using the Sankey diagrams
520 such as those used in the previous section (Figure 9c), a struvite maximum production of 3.7
521 $\text{kg}_{\text{struvite}} \text{kgP}_{\text{inf}}^{-1}$ for virtually all influents was estimated. Thus, the third plant minimizes operating
522 costs by promoting the recovery of biological products and maximizing the use of energy.

523
524 Analysing Figure 11c, it can be seen, as expected, that the biogas production depends exclusively
525 on the influent COD concentration. The incorporation of the TH process has led to increased biogas
526 production by increasing the biodegradability of the sludge. For both configurations the increased
527 production is close to 20%. Being clear the COD dependence, it is possible to set a ratio to estimate
528 the electrical energy generated in CHP per unit of COD fed to the plant: 0.46 kWh $\text{kgCOD}_{\text{inf}}^{-1}$ for
529 the conventional plant, 0.55 kWh $\text{kgCOD}_{\text{inf}}^{-1}$ for the upgraded plant and 0.52 kWh $\text{kgCOD}_{\text{inf}}^{-1}$ for
530 the C/N/P decoupling WWTP.

531
532 Finally, the Figure 11d shows a comparative picture of the self-sufficiency for the three plants.
533 Conventional WWTPs were designed based on traditional biological treatments under a “removal
534 philosophy”, being difficult to achieve the total energy self-sufficiency. As shown in Figure 11d,
535 the total self-sufficiency degree is closely linked to the C/N influent ratio, and this can vary in the
536 conventional plant from 12% for very low COD/TN ratios, up to 85% for very high ratios.
537 Consequently, the treatment plant layouts comparison with different influent ratios would not be
538 entirely correct, nor ensuring that a configuration is always self-sufficient without mentioning the
539 influent ratio of the analysed plant (Jenkins *et al.*, 2014). The philosophy of the second
540 configuration is based primarily on increasing energy production in order to achieve a net overall
541 balance closer to self-sufficiency, in this case between 33% and 103%. In this second configuration
542 it is possible to achieve the self-sufficiency but only with high COD/TN ratios (Figure 11d).
543 Finally, the main objective of the third configuration (C/N/P decoupling WWTP) is the operational
544 costs minimization, by promoting the recovery of bio-products and maximizing the use of energy.
545 With this configuration it is possible to achieve the plant self-sufficiency for almost all COD/TN
546 influent ratios (58% for low COD/N ratios and up to 130% for high ratios).

547 548 **5. DISCUSSION AND CONCLUSIONS**

549 Plant-Wide simulations allows a thorough, comprehensive and accurate analysis of different plant
550 configurations from an energy and resource recovery perspective. To demonstrate the potential of
551 the tool and the need for simulation analysis, this paper compared three different plants: (1) a
552 conventional WWTP, (2) an upgraded or retrofitted WWTP, and (3) a new WRRF concept known
553 as a C/N/P decoupling WWTP.

554
555 Analysing the layouts from a standpoint of resources and energy utilization, a low utilization of the
556 energy content of the components could be observed in all configurations. The only resource that
557 can be recovered efficiently as energy is the organic matter transmitted to the gas phase. The
558 oxidation of the components in the aqueous medium (AS process and nitrification/Anammox
559 technology) releases a large amount of energy as heat that is transmitted to the atmosphere or
560 extracted by the effluent (in these simulations about 37%). This energy is difficult to recover or the

561 recovered energy has a low exergy. Therefore, oxidations in the aqueous medium should be
562 minimized and instead oxidations should be promoted in the gaseous phase. Another key to
563 maximizing the COD energy use is to incorporate technologies that increase sludge biodegradability
564 ($X_{P \rightarrow X_{CH}}, X_{PR}, X_{LI}$), such as the TH process incorporated in the second and third layouts. In the
565 conventional plant, the COD used to produce biogas was around 29%. The TH technology
566 increased this to 36% in the upgraded plant and 34% in the C/N/P decoupling WWTP. In turn, the
567 process reduced sludge production by 12% and by 22%, respectively, in these two plants.

568
569 Regarding resource recovery methods, N recovery techniques are really expensive (ion exchange
570 methods or stripping processes), or as in the case of the technique that could compete with the
571 removal processes, struvite precipitation, the N recovered is minimal (4% estimated by the C/N/P
572 decoupling WWTP simulation, Figure 8). In the case of P, the scarcity of natural P resources
573 converts the recovery of P in the first alternative. The conventional and the upgraded plants
574 removed P by $FeCl_3$ precipitation and only the third configuration attempted to recover the P. The
575 maximum estimated struvite recovery was 43% (Figure 9) and the estimated maximum struvite
576 production was $3.7 \text{ kg}_{\text{struvite}} \text{ kgP}_{\text{inf}}^{-1}$ for virtually all influents.

577
578 Analysing the costs obtained in the study, it can be seen that WWTP self-sufficiency is closely
579 linked to the influent COD/TN/TP ratio. In all plants the trend was similar, the highest degree of
580 self-sufficiency was obtained for the higher ratio values. Achieving self-sufficiency was not
581 possible in conventional plant, in the upgraded plant it depended on the influent ratio, and in the
582 C/N/P decoupling WWTP layout self-sufficiency was feasible for almost all influents (58% for low
583 COD/TN ratios and up to 130% for high ratios). Simulations for different influents showed that, as
584 expected, operating costs increased with the influent load. Assessing costs in detail, aeration was
585 the most significant cost in all configurations (36-48% in the conventional and upgraded plants and
586 41-65% in the C/N/P decoupling WWTP) followed by the chemical dosage, especially in the first
587 and second configurations (20-48% in the conventional and upgraded plants and 5-28% in the
588 C/N/P decoupling WWTP). Regarding plant qualities, the differentiating factor of the upgraded
589 WWTP layout was the increased biogas production. The thermal hydrolysis process increased the
590 biodegradability of the secondary sludge by 40% and electricity production by 19-21% for
591 medium/high COD concentrations and by 43-162% for low COD concentrations. The decrease in
592 aeration costs was not significant in this second configuration (3-11%) due to the NH_X -N release in
593 the TH process (25% more NH_X -N), although efficient nitrification/Anammox technologies had been
594 used to treat rejected water. The fundamental feature of the C/N/P decoupling WWTP was the
595 increase in electricity production (savings of 10-20% for high COD/TN ratios and 39-198% for low
596 ratios) and the decrease in $FeCl_3$ requirements (78-80% for high COD/TN ratios and 42-61% for
597 low ratios) and aeration costs (16-29% for high COD/TN ratios and 4-8% for low ratios), three
598 qualities that enable the plant self-sufficiency.

599
600 Through simulation it has been found that each resource has its optimal way of being treated, and
601 thus the key to maximizing the recovery of resources and energy is the independent treatment of
602 nutrients and COD, valorising the organic matter, and recovering or treating the nutrients.

603
604 The plant layouts proposed in this paper are just a sample of the possibilities for upgrading or
605 designing innovative plants, but they have enabled an analysis of the current needs and challenges
606 that need to be addressed. Even so, the methodology presented here is generic and can be used for
607 any other plant. The use of plant-wide models is, in this context, very useful to ensuring that
608 complex plants featuring different technologies can be analysed reliably and that the model
609 faithfully reproduces the plant behaviour, also in terms of energy and chemical consumption.

610

NOMENCLATURE

| | |
|-------------------|---|
| $Cost_{actuator}$ | Actuator cost (€ d^{-1}) |
| $Cost_{Chem}$ | Chemical agent specific cost (€ kg^{-1}) |
| $Cost_{dosage}$ | Dosage cost (€ d^{-1}) |
| $Cost_{poly}$ | Polyelectrolyte specific cost (€ kg^{-1}) |
| d_p | Particle size (m) |
| D_{pipe} | Pipe diameter (m) |
| D_{stir} | Impeller diameter (m) |
| f_{moody} | Friction coefficient |
| $F_{oversize}$ | Oversize factor |
| g | Gravitational acceleration (m s^{-2}) |
| G | Velocity gradient (s^{-1}) |
| H_{in} | Input enthalpy (kJ d^{-1}) |
| H_{out} | Output enthalpy (kJ d^{-1}) |
| HL | Total head loss (m) |
| HL_f | Friction head loss (m) |
| HL_l | Minor losses (m) |
| HL_s | Static head (m) |
| k_{Chem} | Dosage constant ($\text{g}_{chem} \text{m}^{-3}$) |
| $k_{poly,i}$ | Polyelectrolite and Total solids concentration ratio for the sludge type i ($\text{g}_{poly} \text{kg}_{TSS}^{-1}$) |
| L_{pipe} | Pipe length (m) |
| $\bar{m}_{i,in}$ | Inlet i phase mass flux (gE d^{-1}) |
| MU | Monetary unit (€ d^{-1}) |
| MW_i | Molecular weight of i gaseous phase components |
| n_{CEPT} | Chemically Enhanced Primary Treatment constant |
| N_{js} | Impeller rotational speed required to just suspend the particles (Hz, revolutions per sec.) |
| N_p | Power number |
| $P_{g,in}$ | Absolute gas pressure at the blower/compressor inlet |
| $P_{g,out}$ | Absolute gas pressure at the blower/compressor outlet |
| Q_w | Water flow rate ($\text{m}^3 \text{d}^{-1}$) |
| R | Ideal gas constant ($\text{kJ mol}^{-1} \text{K}^{-1}$) |
| S | Impeller/tank geometry factor |
| Submergence | Submergence (m) |
| $T_{i,in}$ | i phase inflow temperature (K) |
| $T_{i,out}$ | i phase outflow temperature (K) |
| TSS_i | Total suspended solids concentration in the phase i (gSS m^{-3}) |
| u_w | Average liquid velocity (m s^{-1}) |
| V_i | Volume of the i phase (m^3) |
| $W_{actuator}$ | Electrical consumption of actuator (kJ d^{-1}) |
| W_{blow} | Electrical consumption of blower or compressors (kJ d^{-1}) |
| W_{pump} | Electrical consumption of pump (kJ d^{-1}) |
| W_{stir} | Electrical consumption of stirring (kJ d^{-1}) |
| $W_{turbine}$ | Electrical consumption of turbine (kJ d^{-1}) |
| X_{TSS} | Weight percentage of solids in the suspension |

Greek Symbols

| | |
|----------------|---|
| $\gamma_{g,i}$ | Heat capacity ratio of the i gaseous phase components |
| η_i | Dynamic viscosity ($\text{kg m}^{-1} \text{s}^{-1}$) |
| η_{blow} | Efficiency of blowers/compressors |
| η_{CEPT} | Chemically Enhanced Primary Treatment efficiency |
| η_{max} | Chemically Enhanced Primary Treatment maximum efficiency |
| η_{min} | Chemically Enhanced Primary Treatment minimum efficiency |
| η_{pump} | Efficiency of pumps |
| η_{stir} | Efficiency of agitation engines |
| η_{turb} | Efficiency of turbines |
| ν_i | Kinematic viscosity of the i phase ($\text{m}^2 \text{s}^{-1}$) |
| ρ_i | Density (g m^{-3}) |

Subscripts

| | |
|------|---|
| comp | phase components |
| g | gaseous phase |
| m | No. of state variables in the off-gas phase |
| s | solid phase |
| w | Aqueous phase |

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619

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