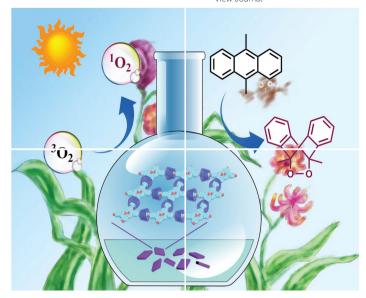
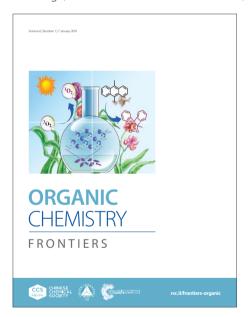
# **ORGANIC**CHEMISTRY

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#### **ARTICLE**

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## NHC-Catalysed [3+2]-Asymmetric Annulation between Pyrazolin-4,5-diones and Enals: Synthesis of Novel Spirocyclic Pyrazolone $\gamma$ -Butyrolactones and Computational Study of Mechanism and Stereoselectivity

Marta Gil-Ordóñez, a Alicia Maestro, \* a Pablo Ortega, b Pablo G. Jambrina, b José M. Andrés \* a

Chiral pyrazolones with a spirocyclic centre at the C4-position are widely found in a large family of medically relevant compounds. In recent years, organocatalysis, particularly that performed with quiral N-heterocyclic carbenes (NHCs), has allowed the enantioselective synthesis of these spirocyclic compounds despite its inherent difficulty. In this work, we describe the fully diastereo- and highly enantioselective synthesis of novel spirocyclic pyrazolone  $\gamma$ -butyrolactones via NHC-catalysed [3+2] annulation reaction of enals and 1H-pyrazol-4,5-diones. To understand the catalytic mechanism and origin of stereoselectivity, electronic structure calculations were carried out. After considering various pathways, we concluded that stereoselectivity-determining step is the formation of the lactone that proceeds after addition of the NHC derived homoenolate to the electrophilic carbonyl group of pyrazolin-4,5-dione. Our calculations predict that the free energy barrier is lower for the (RS) product, which is also the main product experimentally obtained.

#### Introduction

A five-membered heterocycle with two adjacent nitrogen atoms such as the pyrazole ring stands out within the exclusive world of heterocyclic systems. Not only does it play a prominent role in materials and coordination chemistry, its pyrazol-3-one derivatives also exhibit a broad spectrum of potent biological and pharmaceutical activities. In particular, the asymmetric synthesis of pyrazolones with a quaternary spirocyclic centre at the C4 position has attracted the attention from synthetic chemists because spiropyrazolones combine their well-known biological properties with others provided by the rigidity of the skeleton present in spiro compounds (Figure 1).

Over the past few years some organocatalytic strategies have been developed to access spiropyrazolones with a heteroatom incorporated in the chiral spirocentre. Wang group has used quinine-derived thioureas and squaramides for the construction of spiro 4-aminopiropyrazolones from versatile 4-isothiocyanato pyrazolones. There are also not many examples describing access to chiral spirocyclic pyrazolones bearing a 4-oxygen incorporated spirocentre. The diastereodivergent and enantioselective epoxidation of unsaturated pyrazolones developed by the Lattanzi group provide *trans*- or *cis*-spiropyrazolone epoxides using amine-thioureas as effective

Fig 1 Some representative biologically active spiropyrazolones with a quaternary centre in C4 position.

However, to the best of our knowledge there are no examples describing the preparation of spiropyrazolone  $\gamma$ -butyrolactones despite the interest of this cyclic ester unit as a chiral building block for the synthesis of diverse biological active compounds and complex molecules. Instead, various organometallic<sup>10</sup> and organocatalytic<sup>11</sup> approaches have been described for the synthesis of spirooxindole  $\gamma$ -lactones using isatins, 3-hydroxy or methylene oxindoles as starting materials. Among them, chiral *N*-heterocyclic carbene (NHC) catalysed asymmetric reactions<sup>12</sup> constitute an excellent tool to synthesize these spiro

catalysts.<sup>8</sup> Likewise, the bifunctional squaramide-catalysed reaction of in-situ-generated *o*-quinone methides with pyrazolin-5-ones and 4-halo pyrazolones provides easy access to chiral spiro-benzofuran pyrazolones.<sup>9</sup>

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compounds through either NHC-catalysed oxidative [3+2] annulation reaction of dioxindoles and enals,13 the HOBtassisted [3+2] annulation of carboxylic esters with isatins via acyl azolium intermediate, 14 or the enantioselective addition of homoenolate derived enals to isatins by cooperative NHC/Lewis acid catalysis. 15 After reviewing the existing literature, we found only three examples that describe the NHC-catalysed asymmetric synthesis of spirocyclic pyrazolones. The Biju and Yang/Zhong groups have reported the preparation of pyrazolone-fused spirocyclohexanones through [3+3] or [4+2] annulation reactions of enals or γ-chloroenals, respectively, and  $\alpha$ -arylidene pyrazolinones (Scheme 1a and 1b). <sup>16</sup> On the other hand, Enders group has developed a one-pot three-component diastereo and enantioselective synthesis of spirocyclopentane pyrazolones by means of an aldol condensation followed by NHC-catalysed [3+2] annulation reaction (Scheme 1c).<sup>17</sup>

(a) Biju group. [3+3]- Annulation of enals and  $\alpha$ -arylidene pyrazolinones

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(b) Yang/Zhong group. [4+2]- Annulation of  $\gamma\text{-chloro}$  enals and  $\alpha\text{-arylidene}$  pyrazolinones

(c) Enders group. Aldol condensation/[3+2]- Annulation of enals and pyrazolones  $\frac{1}{2}$ 

This work. Experimental and computational study of [3+2]- Annulation of enals and pyrazolediones

**Scheme 1** Asymmetric NHC-catalysed reactions for the synthesis of spiropyrazolones.

Due to the interest of this type of compounds, herein we report their stereoselective synthesis by means a [3+2] annulation of enals and 1*H*-pyrazol-4,5-diones performed with chiral NHCs. To elucidate the mechanism of this cycloaddition reaction, we carried out electronic structure calculations at a density functional theory (DFT) level. These calculations also shed light on the origin of the stereoselectivity experimentally obtained.

#### **Results and Discussion**

We started our studies by combining pyrazolin-4,5-dione  ${\bf 1a}$  and cinnamaldehyde  ${\bf 2a}$  in the presence of different NHC precatalysts  ${\bf A-E}$ , DABCO as base, and toluene as solvent at room temperature (Table 1). The diastereoselectivity for the reaction was excellent in almost all cases, and the spiro  $\gamma$ -butyrolactone  ${\bf 3a}$  was obtained as a single diastereomer (entries 1, 3–5 and 7).

The best enantiomeric ratio (86:14) and yield was achieved when *N*-mesityl triazolium salt **B** (Bode & Catalyse) was used (entry 3).

Table 1 Optimization of reaction conditions.<sup>a</sup>

|                 |         | D                               |                   | E                      |                 |                 |
|-----------------|---------|---------------------------------|-------------------|------------------------|-----------------|-----------------|
| Entry           | pre-NHC | Base                            | Solvent           | Yield (%) <sup>b</sup> | dr <sup>c</sup> | er <sup>d</sup> |
| 1               | Α       | DABCO                           | Toluene           | 41                     | >99:<1          | 37:63           |
| 2               | A1      | DABCO                           | Toluene           | n.r.                   | _               | _               |
| 3               | В       | DABCO                           | Toluene           | 72                     | >99:<1          | 86:14           |
| 4               | C1      | DABCO                           | Toluene           | 38                     | >99:<1          | 79:21           |
| 5               | C2      | DABCO                           | Toluene           | 45                     | >99:<1          | 86:14           |
| 6               | D       | DABCO                           | Toluene           | <10                    | n.d.            | n.d.            |
| 7               | E       | DABCO                           | Toluene           | 31                     | >99:<1          | 50:50           |
| 8               | В       | DABCO                           | Mesitylene        | 65                     | >99:<1          | 87:13           |
| 9               | В       | DABCO                           | Et <sub>2</sub> O | 66                     | >99:<1          | 71:29           |
| 10              | В       | DABCO                           | THF               | 34                     | >99:<1          | 74:26           |
| 11              | В       | DABCO                           | DCM               | 50                     | >99:<1          | 77:23           |
| 12              | В       | DABCO                           | CHCl <sub>3</sub> | 35                     | >99:<1          | 74:26           |
| 13              | В       | DMAP                            | Toluene           | 62                     | >99:<1          | 79:21           |
| 14              | В       | TBD                             | Toluene           | 25                     | >99:<1          | 82:18           |
| 15              | В       | Cs <sub>2</sub> CO <sub>3</sub> | Toluene           | 8                      | >99:<1          | 87:13           |
| 16              | В       | DBU                             | Toluene           | n.r.                   | -               | _               |
| 17              | В       | DIPEA                           | Toluene           | 52                     | >99:<1          | 84:16           |
| 18              | В       | KO <i>t</i> -Bu                 | Toluene           | 29                     | >99:<1          | 88:12           |
| 19              | В       | NaOAc                           | Toluene           | 73                     | >99:<1          | 74:26           |
| 20              | В       | TMEDA                           | Toluene           | 23                     | >99:<1          | 82:18           |
| 21 <sup>e</sup> | В       | DABCO                           | Toluene           | 22                     | >99:<1          | 86:14           |
| 22 <sup>f</sup> | В       | DABCO                           | Toluene           | 62                     | >99:<1          | 79:21           |
| 23 <sup>g</sup> | В       | DABCO                           | Toluene           | 45                     | >99:<1          | 78:22           |
| 24 <sup>h</sup> | В       | DABCO                           | Toluene           | 58                     | >99:<1          | 80:20           |
| 25 <sup>i</sup> | В       | DABCO                           | Toluene           | 39                     | >99:<1          | 76:24           |

<sup>&</sup>lt;sup>a</sup> Reaction conditions: **1a** (0.06 mmol), **2a** (0.06 mmol), pre-NHC (10 mol%), base (0.5 equiv), solvent (1 mL), at rt for 16 h. <sup>b</sup> Yield of **3a** after column chromatography.

<sup>&</sup>lt;sup>c</sup> Dr values determined by <sup>1</sup>H NMR. <sup>d</sup> Er values determined via chiral HPLC analysis.

<sup>&</sup>lt;sup>e</sup> Molar ratio **1a:2a** 1:2. <sup>f</sup> Molar ratio **1a:2a** 1.2:1. <sup>g</sup> LiCl as additive. <sup>h</sup> 0°C. <sup>i</sup> 7 mol% of catalyst **B**.

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When a strong electron-withdrawing group was introduced on the indanol ring (catalyst C2) no change in the enantioselectivity was produced but the yield decreased dramatically (entry 5). A total loss of enantio-discrimination was observed when catalyst E with a pentafluorophenyl group was used (entry 7). Unfortunately, no [3+2] annulation was observed when using pre-catalyst A1 derived from (S)-pyroglutamic acid with a silyl ether substituent (entry 2). 18 Achiral catalyst E afforded racemic adduct 3a with low yield (entry 6). With Bode's catalyst as the most suitable among those examined, solvent screening indicated that toluene was the best solvent for this reaction because although mesitylene kept the same level of enantioselectivity (87:13, entry 8), the yield was slightly lower. Further optimization was carried out to found the best base to access the N-heterocyclic carbene catalyst (entries 13-20). Although the reaction works well with different bases, DABCO was retained as the base of choice. Inorganic bases as Cs<sub>2</sub>CO<sub>3</sub> or KOt-Bu gave the desired product with high diastereo- and enantioselectivity, although with lower yield. On the other hand, in the reactions promoted with bases such as DMAP or NaOAc the enantioselectivity decreased slightly. When DBU was used as the base, no reaction was observed. Finally, other parameters were evaluated (entries 21-25). However, neither the pyrazolin-4,5-dione/enal molar ratio, nor the presence of additives nor the temperature led to an improvement of the previous er and yields.

The scope of the reaction was first examined by modification of the substituents of pyrazolin-4,5-dione. The diastereoselectivity remained excellent (>99:<1) except in the case of substrates having a non-bulky alkyl substituent (R¹), such as a methyl group (**3b**) (Table 2, entry 2).

Table 2 Substrate scope for pyrazolin-4,5-diones 1a-f.a

| Entry | R <sup>1</sup> , R <sup>2</sup> | Base  | Yield (%) <sup>b</sup> | dr <sup>c</sup> | er <sup>d</sup>    |
|-------|---------------------------------|-------|------------------------|-----------------|--------------------|
| 1     | Ph, Ph, <b>3</b> a              | DABCO | 72                     | >99:<1          | 86:14              |
| 2     | Me, Ph, <b>3b</b>               | DABCO | 49                     | 77:23           | 81:19 <sup>e</sup> |
| 3     | <i>t</i> -Bu, Ph, <b>3c</b>     | DABCO | 73                     | >99:<1          | 73:27              |
| 4     | <i>t</i> -Bu, Ph, <b>3c</b>     | TBD   | 63                     | >99:<1          | 87:13              |
| 5     | Naph, Ph, <b>3d</b>             | DABCO | 45                     | >99:<1          | 81:19              |
| 6     | Ph, Me, <b>3e</b>               | DABCO | 65                     | >99:<1          | 80:20              |
| 7     | <i>t</i> -Bu, Me, <b>3f</b>     | TBD   | 58                     | >99:<1          | 84:16              |

 $^{o}$  Reaction conditions: **1a–f** (0.06 mmol), **2a** (0.06 mmol), **B** (10 mol%), DABCO or TBD (0.5 equiv), toluene (1 mL), at rt for 16 h.  $^{b}$  Yield of **3** after column chromatography.  $^{c}$  Dr values determined by  $^{1}$ H NMR.  $^{d}$  Er values determined via chiral HPLC analysis.  $^{e}$  Er for major diastereomer.

It should be noted that a base change is necessary to maintain the level of enantioselectivity in case of 3-tert-butyl substituted pyrazolin-4,5-diones. In this case, 1,5,7-triazabicyclo[4.4.0]dec5-ene (TBD) (entries 3, 4 and 7) was the most suitable base leading to **3c** and **3f** with total diastereoselectivity and good

enantioselectivity (see Supporting Information for optimization of reaction conditions, table S2).

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Next, we studied the influence of the enal by using aliphatic and aromatic  $\alpha,\beta$ -unsaturated aldehydes (Table 3). The reaction of with pyrazolin-4,5-dione **1**a *para*-halo substituted cinnamaldehydes worked well, furnishing the desired spiro compounds 3ab and 3ac in good yields with total diastereoselectivity and good enantioselectivity. As can be seen, the presence of strongly electron-withdrawing (NO2) and electron-donating (MeO) groups in this position of enal led to the lactones 3ad and 3ae in good yields but with lower enantioselectivity. However, C-3 tert-butyl substituted pyrazolin-4,5-dione **3c** reacted with different *p*-substituted cinnamaldehydes affording adducts 3cb-3ce with better enantiomeric ratios independently of the electronic properties of the substituent present in the aromatic ring. The introduction of a methoxy group at the ortho-position of the aryl ring resulted in a significant decrease in both yield and enantioselectivity (3cf). The reaction also tolerates an N-methyl group to provide, in a diastereoselective way, the desired product 3fb in 71% yield with a good er of 90:10. Interestingly, the reaction also works with enals bearing a β-methyl substituent giving the corresponding products (3ag and 3cg) with good yields, very high diastereoselectivity (>99:<1), and moderate to good enantioselectivities (er 80:20-92:8).

Table 3 Substrate scope for enals 2b-g.a

 $<sup>^{\</sup>sigma}$  Reaction conditions: **1** (0.06 mmol), **2b–g** (0.06 mmol), **B** (10 mol%), DABCO or TBD (0.5 equiv), toluene (1 mL), at rt for 16 h.  $^{b}$  Yield of **3** after column chromatography.  $^{c}$  Dr values determined by  $^{1}$ H NMR.  $^{d}$  Er values determined via chiral HPLC analysis.  $^{e}$  **1c** (1.3 mmol), **2b** (1.3 mmol), TBD (0.5 equiv), toluene (15 mL), at rt for 16 h.

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To demonstrate the synthetic utility of this reaction, a gramscale synthesis of spiro compound **3cb** was carried out. Although it was observed a slight decrease in the yield, both diastereoselectivity and enantioselectivity remained excellent (Table 3, entry **3cb**).

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In order to determinate the absolute configuration of products, optically pure (4*R*,5*S*)-4-(4-chlorophenyl)-7,9-diphenyl-1-oxa-7,8-diazaspiro[4.4]non-8-ene-2,6-dione, (–)-**3ac** was isolated after recrystallization from chloroform.<sup>19</sup> X-ray crystallographic analysis of a single-crystal confirmed the absolute configuration as (4*R*,5*S*) for the spirocycle (see the Supporting Information for details).

#### **Computational Methods and Model Reaction**

To elucidate the mechanism of the NHC-catalysed [3+2] asymmetric annulation, and rationalise the origin of the enantio- and diastereoselectivity observed, we carried out electronic structure calculations. The stationary points (minima and saddle points) were optimized at the M06-2X/6-31G(d,p) $^{20}$  level of theory using Gaussian16. $^{21}$  Solvation effects were included using the SMD continuum solvation model, $^{22}$  with the default parameters for toluene.

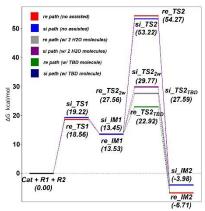
Frequencies were calculated to ensure the convergence and intrinsic reaction coordinate (IRC) calculations were carried out to confirm that the transition states obtained connect reactants and products.<sup>23</sup> Subsequently, single-point calculations of the stationary points were computed at a M062X/maug-cc-pVTZ level, and the free-energies reported were obtained by adding the electronic energy at the M062X/maug-cc-pVTZ level of theory with the thermal free energies computed at the M062X/6-31G(d, p) level. This selection of functional and basis set was done according to recent benchmarks,<sup>24</sup> and the overall procedure is similar to those followed in the literature for the study of other NHC-catalysed [3+2] asymmetric annulations.<sup>25</sup> To study the NHC-catalysed [3+2] asymmetric annulations performed in this work, we used as a model the reaction between R2(1c) and R1(2e), (Scheme 2) for which the highest er (94:6) was experimentally obtained (see Table 3).

Scheme 2 NHC-catalysed [3+2] asymmetric annulation model used in the calculations.

The mechanism that emerges from our calculations is depicted in Scheme 3. In it, we can differentiate two different stages. In the first stage, the NHC catalyst reacts with the aldehyde moiety of **R1(2e)**, forming a new C-C bond, and finally leading to the **IM2**.

Scheme 3 Catalytic cycle of [3+2] cycloaddition reaction.

The free energy profile for the first stage is shown in Scheme 4. According to our calculations, the rate-limiting step in this stage is the hydrogen migration and formation of Breslow intermediate. The free energy barrier for the intramolecular hydrogen migration is too high (> 50 kcal/mol). However, this energy barrier shrinks to 27.56 kcal/mol if assisted by two water molecules that could be present in catalytic concentrations in the bulk, and to 22.92 kcal/mol if hydrogen migration is assisted by TBD.<sup>26</sup> Depending on the side of the carbene attack with respect to the aldehyde, the reaction can proceed via a *Re/Si* path. As it is shown in Scheme 4, the barrier for the *Si* mechanism is larger, regardless of whether the reaction is assisted by water or TBD. Accordingly, from this step, we only focused on the *Re* mechanism.



Scheme 4 Free-energy profile of the first stage of the reaction. The solvation-corrected relative free-energies at the SMD(toluene)/M06-2X/maug-cc-pVTZ level are given in kcal/mol. Cartesian coordinates of all the stationary points are shown in the Supporting Information.

In the second stage, whose free-energy profile is depicted in Scheme 5, **R2(1c)** reacts with the activated aldehyde (*Re\_IM2*), leading to the formation of **P(3ce)** and the release of the NHC

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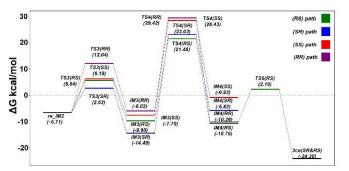
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catalyst. This stage consists of three further steps, addition of the pyrazolin-4,5-dione (3<sup>rd</sup> step), formation of the lactone (4<sup>th</sup> step), and release of NHC (5<sup>th</sup> step). Of these steps, the 4<sup>th</sup> step is the rate-limiting step and also will determine the er and dr ratios. In our calculations, the 4<sup>th</sup> step is divided in two stages. In a first stage, the basis assists the proton transfer from a hydroxyl group at C4-position of the pyrazoldione moiety to the  $\alpha$ -carbon atom of the homoenolate intermediate, a process that is associated with high energy barriers. Then the dissociation of TBD occurs and spontaneous rearrangement to



Scheme 5 Free-energy profile of the second stage of the reaction. The solvation-corrected relative free-energies at the SMD(toluene)/M06-2X/maug-cc-pVTZ level are given in kcal/mol. Cartesian coordinates of all the stationary points are shown in the Supporting Information.

As it is depicted in Figure 2, depending on the attack mode of the pyrazoldione R2(1c) in the 3<sup>rd</sup> step, and as a consequence of the presence of two prochiral faces in both R2 as Breslow intermediate Re IM2, we could find four different mechanisms that will produce four different stereoisomers (SS), (SR), (RR) and (RS). The formation of rotational conformers associated with the Z-configuration of the C=C bond was not feasible due to the high energy barrier associated with C=C torsion (above 30 kcal/mol). Although the free-energy barrier for the four pathways is considerably different ranging from 2.63 to 12.04 kcal/mol, all barriers are well below those obtained in the 4th step. In this step, formation of the lactone proceeds via protonation of the basis (TBD). Our calculations predict that the free energy barrier is lower for the (RS) mechanism, which after NHC release that proceeds with almost no barrier, will produce the P(RS)(3ce) product. This is also the main product obtained experimentally (with an er of 94:6).

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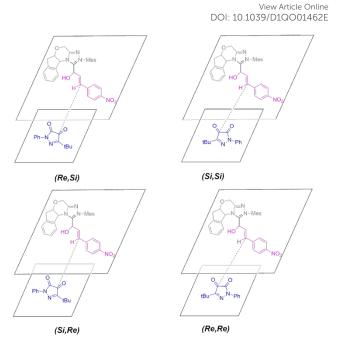


Fig 2 Diagram of the four possible modes of attack of R2(1c) on Re\_IM2. The coordinates of the transition states originated from these modes of attack are shown in the Supporting Information, and their energies are depicted in Scheme 5 (TS3).

To rationalize the origin of the enantioselectivity, we analyse the structure of the four TS4 (shown in Scheme S6 in the Supporting Information). According to TS4 structures, there is a good correlation between the degree of synchronicity of the two proton transfer steps at the saddle point and the energy barrier, as it was previously observed for Diels-Alder reactions.<sup>27</sup> For the TS4(SR) and TS4(RS) structures, both protons are located between IM3 and TBD at the TS, while for TS4(SS) the OH group of pyrazoldione has not been deprotonated, and for **TS4**(RR) this proton has already been transferred to TBD. To quantify the origin of the differences, we combine the distortion-interaction analysis (following the analysis carried out for other NHC-catalysis articles<sup>25a,d,28</sup>) with the energy decomposition analysis (carried out using QChem 5.2<sup>29,30</sup>) and a 6-31G basis set. Leaving aside the entropic contribution, stabilization of the TS4(RS) is traced back from a more favourable interaction, which is induced by the relaxation of the molecular orbitals of the fragments (lower polarization energy. For details see Scheme S7 and Tables S3 and S4 in the Supporting Information).

#### **Conclusions**

In the present work, we have described the first asymmetric synthesis of spiropyrazolone  $\gamma$ -butyrolactones from 1H-pyrazol-4,5-diones and enals by an NHC-catalysed [3+2] annulation process. The developed protocol is mild and tolerates a wide range of substituents on both substrates. The use of Bode's catalyst in this cascade reaction provided the enantioenriched spirocyclic compounds containing two contiguous stereogenic centres in good yields with excellent diastereoselectivity and

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good enantioselectivity. To understand the catalytic mechanism and origin of stereoselectivity, electronic structure calculations were carried out. These indicate that the Br\u00f6nsted base used to generate the carbene NHC, also assists the hydrogen migration for the generation of homoenolate. Furthermore, the formation of the lactone that proceeds after the pyrazolin-4,5-dione attack on the Breslow intermediate is the rate-limiting step and determining of the enantio- and diastereomeric ratios. The pathway responsible for the RS-configuration of spirocyclic pyrazolones has been identified to be the most favourable one.

#### **Author Contributions**

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59 60 Conceptualization and methodology, A. M. and J. M. A. Investigation and writing-original draft, A. M., M. G. O and P. G. J. Electronic structure calculations, P. O. and P. G. J. Writingreview & editing, A. M. and J. M. A. Funding acquisition, J. M. A.

#### **Conflicts of interest**

There are no conflicts to declare.

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