

<https://doi.org/10.1016/j.foodchem.2018.10.025>

Received 5 July 2018; Received in revised form 5 October 2018;

Accepted 5 October 2018

✉ Corresponding author.

E-mail addresses: guisiher@hotmail.com (G. Simó), effernan@iaf.uva.es (E. Fernández-Fernández), jvila@pat.uva.es (J. Vila-Crespo), violeta.ruiperez@uva.es (V. Ruipérez), rjosem@iaf.uva.es (J.M. Rodríguez-Nogales).

Food Chemistry 276 (2019) 643–651

Available online 06 October 2018

0308-8146/ © 2018 Elsevier Ltd. All rights reserved.

1 Title page

2 **Effect of stressful malolactic fermentation conditions on the**
3 **operational and chemical stability of silica-alginate encapsulated**
4 ***Oenococcus oeni***

5

6 **Guillermo Simó^a, Encarnación Fernández-Fernández^a, Josefina Vila-Crespo^b,**
7 **Violeta Ruipérez^b, José Manuel Rodríguez-Nogales^{a,*}**

8 ^a Food Technology Department. University of Valladolid. School of Agricultural
9 Engineering. Av. Madrid 44, 34071 Palencia, Spain

10 ^b Microbiology Department. University of Valladolid. School of Agricultural
11 Engineering. Av. Madrid 44, 34071 Palencia, Spain

12

13 ***Corresponding author.** Telephone number: +34 979108478. Fax number: +34
14 979108302. E-mail address: rjosem@iaf.uva.es

15

16 **E-mail addresses for other authors:**

17

18 Guillermo Simó: guisiher@hotmail.com

19 Encarnación Fernández-Fernández: effernan@iaf.uva.es

20 Josefina Vila-Crespo: jvila@pat.uva.es

21 Violeta Ruiperez: violeta.ruiperez@uva.es

22

23 **Running header:** Chemical stability of SiO₂-alginate biocomposites

24

25

26

27 **Effect of stressful malolactic fermentation conditions on the operational and**
28 **chemical stability of silica-alginate encapsulated *Oenococcus oeni***

29

30 **Abstract**

31 *Oenococcus oeni* was encapsulated into inter-penetrated polymer networks of silica-
32 alginate (SiO₂-ALG). Fourier transform infrared spectroscopy analysis proved the
33 presence and the polycondensation of the siliceous material used in SiO₂-ALG capsules.
34 Environmental scanning electron microscopy showed that the structure of SiO₂-ALG
35 biocapsules was rougher than in alginate (ALG) biocapsules. The behaviour of SiO₂-
36 ALG biocapsules was evaluated at pH 3.0-3.6 and alcohol degrees of 12-15%.
37 Repeated-batch malolactic fermentations (MLF) demonstrated that SiO₂-ALG
38 biocapsules can be reused efficiently for five times in either low-pH or high-ethanol
39 wines, while free bacteria only can be used once under the most favourable MLF
40 conditions. The inclusion of siliceous materials into ALG hydrogel improved the
41 stability of the biocapsules, reducing their shrinking and achieving an excellent integrity
42 under winemaking conditions. These results proved the possibility of industrial
43 application of SiO₂-ALG biocapsules in winemaking.

44

45 **Highlights**

- 46 • Combination of ALG and SiO₂ enhances chemical stability of biocapsules in
47 wine
- 48 • Inclusion of SiO₂ improves the operational stability of ALG-based biocapsules
- 49 • SiO₂-ALG biocapsules can be reused efficiently in low-pH or high-ethanol
50 wines

- 51 • SiO₂-ALG entrapped bacteria achieve a successful MLF under adverse
52 conditions

53 **Key words:** Biomaterial, biocatalysis, immobilization, hydrogel.

54 **1. Introduction**

55 In the last decade, cell encapsulation has been reported as an alternative to free
56 cells for the enhancement of long-term bio-efficiency in fermentative processes
57 (Kourkoutas, Manojlović, & Nedović, 2010; Maicas, 2001). Different materials have
58 been applied as carriers for cell encapsulation, mainly organic polymers (such as ALG,
59 carrageenan, chitosan, agarose, pectin, gelatine, and chitin). Ionotropic ALG hydrogel
60 has been extensively used for this purpose due to being easy and cost-effective to
61 prepare, the encapsulation conditions are mild for cell, and its porous structure allows
62 the free diffusion of substrates and products. However, the low chemical stability and
63 the poor mechanical strength of ALG hydrogel hinder its industrial implementation
64 (Ching, Bansal, & Bhandari, 2015). Different alternatives have been assessed to
65 improve these disadvantages based on coating techniques (Simó, Fernández-Fernández,
66 Vila-Crespo, Ruipérez, & Rodríguez-Nogales, 2017a) as well as on the incorporation of
67 other polymers and/or fillers into ALG gel (Neufeld, & Poncelet, 2004).

68 Previous researches have been done for the improvement of the stability of ALG
69 capsules using covalent polymeric networks of SiO₂ obtained by the sol-gel process
70 (Thibaud Coradin, Allouche, Boissiere, & Livage, 2006; Kuncová & Trogl, 2010).
71 These inorganic-organic hybrid materials have the advantages of the organic component
72 (ALG) (biocompatibility, elasticity and flexibility) and the inorganic component (SiO₂)
73 (strength, and thermal and chemical stability) (Hwang & Gu, 2013). From a
74 toxicological point of view, SiO₂-ALG capsules are very suitable for food application

75 since (i) ALG is a natural polymer that has been recognized as a “generally recognized
76 as safe” material (Sosnik, 2014), and (ii) siliceous materials are being used for
77 biomedical applications and hence admitted as safe for humans (Carturan, Dal Toso,
78 Boninsegna, & Dal Monte, 2004).

79 The classic sol-gel process involves the hydrolysis of alkoxide precursors in the
80 presence of water and catalysts (either acids or bases) to form silanol groups (Si-OH)
81 with the releasing of alcohol molecules. Then, condensation of silanol groups takes
82 place to produce siloxanes (Si-O-Si). Finally, polycondensation of silanol and siloxanes
83 occurs, yielding SiO₂ materials (Niederberger & Pinna, 2009). The use of sol-gel
84 chemistry for cell encapsulation presents some risks for cell viability. On one hand, the
85 released alcohol from the hydrolysis of alkoxide precursors may be toxic to cell. On the
86 other hand, the use of cosolvent and catalysts to increase the low water solubility and
87 the reactivity of alkoxide precursors, respectively, may also cause loss of cell viability.
88 Some solutions have been reported to solve these drawbacks based on either the
89 removal of alcoholic by-products before cell encapsulation or the use of an aqueous sol-
90 gel process with non-alkoxide precursors (Coradin et al., 2006; Kuncová & Trogl,
91 2010).

92 A successful strategy was recently proposed by our research team, in which an
93 easy, economic and efficient cell encapsulation procedure based on inter-penetrated
94 polymer networks of SiO₂ and ALG was developed (Simó, Fernández-Fernández, Vila-
95 Crespo, Ruipérez, & Rodríguez-Nogales, 2017b; Simó, Vila-Crespo, Fernández-
96 Fernández, Ruipérez, & Rodríguez-Nogales, 2017). The hybrid composite was obtained
97 by the sol-gel aqueous route using non-alkoxide silicon precursors (colloidal silica and
98 sodium silicate). The resulting SiO₂-ALG biocapsules with entrapped *O. oeni* bacteria
99 were applied in winemaking to develop MLF. SiO₂-ALG biocapsules presented a

100 remarkable increase in mechanical robustness of 328% and 65% in L-malic acid
101 consumption, observing insignificant bacteria release compared to untreated ALG
102 biocapsules (Simó, Vila-Crespo, et al., 2017). Also, SiO₂-ALG encapsulated *O. oeni*
103 showed an enhanced MLF in wines with high concentration of ethanol (13-16%), low
104 pH (3.0-3.3) and at low fermentation temperatures (13-15°C) compared to ALG
105 capsules (Simó, Fernández-Fernández, et al., 2017b).

106 To the best of our knowledge, only two other works using SiO₂-ALG
107 composites for the encapsulation of *O. oeni* have been reported (Callone, Campostrini,
108 Carturan, Cavazza, & Guzzon, 2008; Guzzon, Carturan, Krieger-Weber, & Cavazza,
109 2012). Unlike our strategy, both works noted a non-aqueous method based on double-
110 layer coating of ALG capsules embedded with *O. oeni* using tetraethoxysilane and
111 methyltriethoxysilane as alkoxide silicon precursors. The coating procedure did not
112 increase the metabolic activity of bacteria, although there was a notable improvement of
113 the hardness of biocapsules.

114 One of the potential main advantages claimed for the use of encapsulated cells
115 includes their easy separation and reuse. Industrial implementation of biocapsules
116 depends on the possibility of their recycling for a long period without compromising
117 their metabolic activity and viability. Moreover, biocapsules must be stable along the
118 fermentative period for the success of alcoholic beverage production (Kourkoutas,
119 Bekatorou, Banat, Marchant, & Koutinas, 2004). A deep knowledge of the effect of the
120 hostile wine environment on biocapsules stability needs to be addressed to obtain a full
121 exploitation of this potential advantage.

122 In this context, the encapsulation of *O. oeni* into inter-penetrated polymer
123 networks of SiO₂ and ALG could involve a great enhancement of the operational and

124 chemical stability of the biocapsules to the stressful conditions of MLF. For that, the
125 aim of this study was to evaluate the reusability of Si-ALG biocapsules under
126 unfavorable MLF conditions and to test their long-term chemical stability in harsh wine
127 conditions of low pH and high ethanol concentration. The presence of siliceous
128 materials in the Si-ALG biocomposites and their effect on the external and internal
129 structure of the capsules were also verified by Fourier transform infrared spectra (FTIR)
130 and Environmental scanning electron microscopy (ESEM) analysis.

131

132 **2. Experimental**

133 *2.1. Preparation of SiO₂-ALG biocomposites with encapsulated bacteria*

134 Inter-penetrated polymer networks of silica/silicate and ALG (Simó, Vila-
135 Crespo, et al., 2017) were carried out adjusting the pH of a water solution of 1.23 M
136 colloidal silica (Ludox HS40, Sigma-Aldrich, Spain) and 0.06 M sodium silicate
137 (Sigma-Aldrich) to 6.3 by adding 2N HCl. Then, sodium ALG (Panreac, Spain) was
138 mixed with the silica/silicate solution until obtaining a final concentration of 2% (w/v).
139 Freeze-dried bacteria (*O. oeni* strain LALVIN VP 41[®] MBR, Lallemand, France) were
140 rehydrated in water at 20°C according to the manufacturer's instructions and they were
141 added at a concentration of $\sim 3 \cdot 10^9$ cfu/g of gel. Population of freeze-dried culture was
142 tested in Tomato Juice Agar (Difco, Sparks, MD, USA) at 22°C. Capsules were made
143 by extrusion of the well-mixed siliceous material-ALG-cell suspension with a 10 mL
144 sterile syringe with a nozzle of 1.78 mm of diameter (BD 166 Plastipak, Spain) into
145 sterile 0.2 M CaCl₂ solution under continuous agitation (260 rpm) at 22°C. Capsules
146 were kept in this solution for 2 h and washed with water at 22°C. Capsules of ALG were

147 made using the method described for SiO₂-ALG capsules but omitting the addition of
148 siliceous material. All solutions were autoclaved prior to use.

149 *2.2. Environmental scanning electron microscopy (ESEM)*

150 The prepared SiO₂-ALG and ALG biocapsules were dehydrated by freeze-
151 drying at -40°C. The samples were coated with gold and examined in a Quanta 200FEI
152 ESEM (Hillsboro, Oregon, USA) with a backscattered electron detector at a landing
153 energy of 4.0 KeV.

154 *2.3. Fourier transform infrared spectroscopy analysis (FTIR)*

155 Functional groups of freeze-dried SiO₂-ALG and ALG capsules were
156 characterized by FTIR spectrophotometry. The vibrational spectra of the capsules in
157 the 400-4000 cm⁻¹ spectral range were measured using a Thermo Scientific Nicolet
158 iS50 FTIR spectrometer (Waltham, MA, USA), equipped with a built-in diamond
159 attenuated total reflection (ATR) system. Spectra of the samples were recorded with a
160 1 cm⁻¹ spectral resolution, and 32 scans.

161 *2.4. Operational stability of biocapsules*

162 Reusability of SiO₂-ALG biocapsules was determined using the same inoculum
163 for five consecutive cycles of MLF under different winemaking conditions. Free
164 bacteria were also used as control. Red wine of “Tinta de Toro” variety (3.0 g/L malic
165 acid, pH 3.4 and alcoholic degree of 14.0%, v/v) was partially dealcoholized to obtain a
166 final alcoholic degree of 11% under reduced pressure in a rotary evaporator. A first
167 batch of wine was made adjusting the alcoholic degree of the concentrated wine to
168 12.0%, 13.5% and 15.0% with ethanol and at pH 3.4 using acid or basic concentrated
169 solutions. A second lot of wine was set to an alcoholic degree of 13.0% and pH of 3.0,
170 3.3 and 3.6. MLF was conducted in sterilized red wine (by filtration through a 0.2 µm

171 pore-size) inoculating either free or SiO₂-ALG encapsulated bacteria at a concentration
172 of $\sim 9 \times 10^7$ cfu/mL of wine. Rehydrated culture from freeze-dried *O. oeni* strain
173 (LALVIN VP 41[®] MBR) was used as free inoculum. Bacterial population of free
174 inoculum was tested in Tomato Juice Agar at 22°C (Difco). MLF was carried out in
175 triplicate and its progress was monitored by the determination of L-malic acid
176 concentration (in triplicate) using an enzymatic kit (TDI, Barcelona, Spain). After 120 h
177 of incubation at 22°C, both free and encapsulated bacteria were removed from wine and
178 introduced into a new wine. Free bacteria were recovered by centrifugation (15 min at
179 2300 g) while encapsulated ones were separated from wine by filtration using a metallic
180 sieve (nominal sieve opening of ~ 1.5 mm).

181 *2.5. Long-term chemical stability of biocapsules*

182 For capsule stability evaluation, the same two lots of wine developed for the
183 operation stability study were used. Fifteen SiO₂-ALG biocapsules containing bacteria
184 were immersed in 10 mL of red wine at different pH (3.0, 3.3 and 3.6) and alcoholic
185 degrees (12.0%, 13.5% and 15.0%). ALG capsules were also analysed as control for
186 comparison. Mass and diameter of both SiO₂-ALG and ALG biocapsules were
187 determined along the experimental time at 22°C. The diameter of biocapsules was
188 measured by software IMAGEJ 1.47v (National Institutes of Health, Bethesda, MD,
189 USA). The mass of biocapsules was determined using a semimicro balance with a
190 readability of 0.1 mg (AS 228 220/C/2, Radway, Brancka, Poland). The release of ALG
191 from both capsules was determined by duplicate after 18 days of immersion in wine
192 (fifteen biocapsules in 10 mL of red wine) following the method proposed by Segarra et
193 al. (1995). All experiments were carried out in triplicate.

194 *2.6. Statistical analysis*

195 SPSS v. 17.0 statistical package (SPSS Inc., Chicago, Ill, USA) was used for all
196 statistical analyses. A variance analysis was carried out to determine statistical
197 differences between samples ($p < 0.05$). In figures, error bars were calculated as standard
198 error.

199

200

201 **3. Results and discussion**

202 *3.1. Morphological observations*

203 SiO₂-ALG and ALG biocapsules were nearly spherical and homogeneous in size
204 (Figure 1, images a and b), showing a diameter of 3.24 ± 0.20 and 2.80 ± 0.15 mm,
205 respectively. The sol-gel process using siliceous materials slightly increased the capsule
206 diameter due to the rise of viscosity of ALG solution in presence of colloidal
207 silica/silicate and modified their physical appearance, turning them opaque and white.
208 This behaviour was also observed in alginate-protamine-SiO₂ hybrid capsules where the
209 transparent ALG capsules became opaque and white during the silicification process
210 (Wang et al., 2013).

211 The surface morphology and the internal structure of ALG and SiO₂-ALG
212 biocapsules containing *O. oeni* were studied by ESEM. Both external and internal
213 structures of ALG biocapsules were relatively smooth (Figure 1, images c and d,
214 respectively). However, the morphology of both structures changed in the presence of
215 silica/silicate and became rough (Figure 1, images e and f). A highly porous network
216 was formed by close-packed silica particles. SEM, thermogravimetric analysis and
217 nitrogen adsorption-desorption experiments performed in SiO₂-ALG biocomposites
218 using the sol-gel route with aqueous precursors revealed that ALG polymer filled the

219 macroporosity of hydrogel rather than being strongly entangled with siliceous material
220 (Coradin & Livage, 2003). Compared to ALG biocapsules, the presence of an inorganic
221 network of SiO₂ notably enhanced the bioactivity of SiO₂-ALG encapsulated *O. oeni*
222 (Simó, Fernández-Fernández, et al., 2017b). This efficiency in L-malic acid
223 consumption could be due (at least partly) to the highly porous structure observed in
224 SiO₂-ALG biocapsules that facilitates the substrate accessibility for encapsulated
225 bacteria. This is consistent with the improvement of diffusion characteristics noted in
226 SiO₂-ALG composites (in comparison with ALG hydrogels), evaluating the diffusion
227 coefficients of reduced nicotinamide adenine dinucleotide, an enzymatic cofactor with a
228 molecular weight of about five-times higher than malic acid (Xu, Lu, Li, Jiang, & Wu,
229 2006).

230

231 3.2. Chemical characterization of capsules

232 Chemical composition of ALG and SiO₂-ALG capsules was characterized by FTIR-
233 ATR spectroscopy. The FTIR spectra of the ALG capsules show characteristic bands
234 displayed for alginic acid (Figure 2a). According to literature reports (Gómez-Ordóñez
235 & Rupérez, 2011; Pannier, Soltmann, Soltmann, Altenburger, & Schmitt-Jansen, 2014),
236 the bands at 1612 cm⁻¹ and 1412 cm⁻¹ are assigned to asymmetric and symmetric
237 stretching vibrations of the COO⁻ of alginic acid, respectively. A broad peak near 1000
238 cm⁻¹ corresponds to vibrational modes of the carbohydrate ring between (1-4)-β-D-
239 mannuronic acid and (1-4)-α-L-guluronic acid of alginic acid (Gómez-Ordóñez &
240 Rupérez, 2011). The FTIR spectra of SiO₂-ALG capsules (Figure 2b) present the bands
241 associated with symmetric and asymmetric stretching vibrations of the COO⁻ but in very
242 low intensity, suggesting an interaction of the carboxylate groups of ALG with the

243 siliceous polymer derived from *in situ* gelation of silica/silicate (Xu, Jiang, Lu, Wu, &
244 Yuan, 2006). Under the pH conditions used for bacterial encapsulation (pH 6.3), only
245 hydrogen bonds between hydroxyl groups ($-OH$) of SiO_2 and carboxyl groups ($-$
246 $COOH$) of ALG could take place because the carboxyl groups of ALG are negatively
247 charged and the hydroxyl groups are neutral, whereas SiO_2 presents neutral charge
248 (Coradin & Livage, 2003). A peak at 790 cm^{-1} corresponds to the symmetric stretching
249 of Si-O-Si groups, indicating an extension of SiO_2 polymerization. A peak of high
250 intensity at about 1100 cm^{-1} partially overlapped with the corresponding bands to
251 vibrational modes of the carbohydrate ring of ALG could be assigned to Si-O-Si
252 bending vibration. Finally, the band at 955 cm^{-1} corresponds to Si-OH vibration
253 marking also the presence of silica in this capsule (Xu et al., 2006).

254

255 3.3. Reusability of SiO_2 -ALG biocapsules

256 Bacterial reuse reduces production cost, thus the study of the operational stability of
257 SiO_2 -ALG biocapsules is essential for their industrial implementation. To examine their
258 reusability, repeated-batch MLF were carried out five times using the same inoculum of
259 SiO_2 -ALG encapsulated bacteria. Reusability test of free bacteria was also performed as
260 control for comparison. Operational stability of ALG biocapsules was not studied since
261 malolactic activity of ALG-encapsulated bacteria was notably lower than that observed
262 in SiO_2 -ALG biocapsules (Simó, Fernández-Fernández, et al., 2017b).

263 Figure 3 shows L-malic acid conversion by both SiO_2 -ALG biocapsules and free
264 bacteria along five-repeated cyclic operations in red wine at different pH (3.0, 3.3 and
265 3.6). Wine pH had a marked effect on malolactic activity of both type of inoculum,
266 achieving the best results at pH 3.6. Malolactic conversion by SiO_2 -ALG encapsulated

267 bacteria only decreased by 15.7% after five cycles in wine at pH 3.6 (from 98.4% of
268 conversion in run #1 to 83.0% of conversion in run #5). The decrease of malolactic
269 conversion was calculated as relative percentage as

$$\left(1 - \frac{C_{MLF5}}{C_{MLF1}}\right) \times 100$$

270 where C_{MLF1} and C_{MLF5} are the malolactic conversions of the first and fifth MLF,
271 respectively, observed at each pH, alcoholic degree and type of inoculum. The
272 decreases in L-malic acid conversion were slightly higher at pH 3.3 and 3.0 after five-
273 repeated batches, reaching drops of 20.6% (from 97.6% in run #1 to 77.5% in run #5)
274 and 35.3% (from 98.5% in run #1 to 63.7% in run #5), respectively. Malolactic
275 conversion by free bacteria was stable only at pH 3.6 (13.0% of alcoholic degree) after
276 their reuse for five times. This result was similar to that obtained by four successive
277 reinoculations of cultures of *O. oeni* in Monastrell red wine under favourable MLF
278 conditions (pH 3.5 and 11.1% alcoholic degree) (Sergi Maicas, Pardo, & Ferrer, 2000).
279 L-malic acid conversions by free *O. oeni* drastically dropped by 78.4% (from 84.6% in
280 run #1 to 18.3% in run #5) and a 94.9% (from 78.1% in run #1 to 3.97% in run #5) at
281 pH 3.3 and 3.0, respectively.

282 Free and SiO₂-ALG encapsulated *O. oeni* were also tested in five consecutive
283 MLF processes in wines with different alcoholic degrees (12.0%, 13.5% and 15.0%) to
284 determine if there was deactivation of bacteria after repeated use (Figure 3). Alcoholic
285 degree notably affected L-malic acid conversion of both free and encapsulated bacteria
286 along the successive MLF, achieving the best results at the lowest alcoholic degree
287 (12.0%). L-malic acid conversion by SiO₂-ALG encapsulated bacteria was maintained
288 at 90.7% after five cycles in favourable winemaking conditions (12.0%), corresponding
289 with a slight reduction of 7.8% (a high conversion of 98.3% was achieved in the first

290 cycle). Drops of 29.5% (from 99.4% in run #1 to 70.1% in run #5) and 36.9% (from
291 97.6% in run #1 to 61.6% in run #5) in L-malic acid conversion were observed in wines
292 with alcoholic degrees of 13.5% and 15.0%, respectively, after the fifth cycle of use of
293 SiO₂-ALG biocapsules.

294 L-malic acid conversion by free bacteria remained almost unchanged (above
295 98%) in wine with the lowest alcoholic degree (12.0%) after their reuse for five cycles.
296 However, sharp declines of 95.0% (from 85.9% in run #1 to 4.3% in run #5) and 97.0%
297 (from 78.9% in run #1 to 2.3% in run #5) on L-malic conversion were observed at
298 alcoholic degrees of 13.5% and 15.0%, respectively, corroborating the negative impact
299 of ethanol on malolactic activity of *O. oeni* (Bonomo, Di Tomaso, Calabrone, &
300 Salzano, 2018).

301 Growth and metabolism of *O. oeni* in wine depend on a multitude of parameters
302 (Bauer & Dicks, 2004). It is well documented the adverse effect of low-pH wines as
303 well as high-ethanol wines on the ability of *O. oeni* to survive in this unfavourable
304 environment (Bonomo et al., 2018; Sumbly, Grbin, & Jiranek, 2014). Ethanol causes
305 breaking of cell membrane structure and membrane fluidity alterations, affecting mainly
306 transport of metabolites, and cell wall and membrane biogenesis (Olguín et al., 2015).
307 Wine pH also plays a crucial role in the beginning of MLF as well as the time required
308 to complete MLF (Knoll et al., 2011). pH values lower than 3.5 negatively affect the
309 growth of *O. oeni* and reduce their ability to metabolize L-malic acid (Betteridge, Grbin,
310 & Jiranek, 2015; Rosi, Fia, & Canuti, 2003).

311 Our results have revealed that the encapsulation into SiO₂-ALG hydrogel
312 notably improved the operational stability of bacteria, remaining high malolactic
313 activity after five successive MLF at the harsh environmental conditions of pH 3.0 and

314 13% of alcoholic degree, as well as pH 3.4 and 15.0% of alcoholic degree. These results
315 agree satisfactorily with a previous study of our group. (Simó, Fernández-Fernández, et
316 al., 2017b). In this study, we found that the inclusion of colloidal silica and silicates into
317 ALG capsules markedly enhanced the malolactic activity of *O. oeni* at low pH, high
318 ethanol, and low fermentation temperatures. Guzzon et al. (2012) reported successful
319 MLF after three cycles in Chardonnay wine (ethanol 12.5%, pH 3.3, malic acid 3.5 g/L)
320 inoculating *O. oeni* entrapped into ALG capsules coated with an organic-silica
321 membrane. Encapsulated *O. oeni* in polyvinyl alcohol gel could be successfully reused
322 through six cycles of MLF in Tempranillo wine (ethanol 14.2%, pH 3.7, malic acid 1.9
323 g/L), retaining 75% of efficacy after the sixth batch (Rodríguez-Nogales, Vila-Crespo,
324 & Fernández-Fernández, 2013). Conversely, in Monastrell red wine under gentle
325 winemaking conditions (ethanol 11.0%, pH 3.5, malic acid 3.5 g/L), a drop of about
326 50% of the initial L-malic acid conversion was observed after six consecutive cycles of
327 MLF inoculating *O. oeni* immobilized on positively-charged cellulose sponge (Sergi
328 Maicas, Pardo, & Ferrer, 2001). *Lactobacillus casei* cells immobilized on delignified
329 cellulosic material showed a high L-malic acid conversion (80%) in the first cycle of
330 MLF under mild winemaking conditions (ethanol 11.2% and pH 3.5), declining
331 gradually up a conversion of 14.7% in the sixth batch (Agouridis, Bekatorou, Nigam, &
332 Kanellaki, 2005). These results highlight that the success in the reusability of
333 immobilized bacteria strongly depends on both the type of the selected matrix and the
334 immobilization method, as well as MLF conditions and bacterial strain. In our study, the
335 differences in malic acid conversion found in the first cycle of MLF between both type
336 of inoculum (free and Si-ALG encapsulated bacteria) may be lower if the fermentative
337 behaviour of the selected strain was higher at low pH and high ethanol concentration.

338 The preservation of malolactic activity of SiO₂-ALG encapsulated bacteria along
339 the operation stability assays could be due to the organic-inorganic matrix, which could
340 provide bacterial protection from the unfavourable winemaking conditions (Kourkoutas
341 et al., 2010). An enhancement of the stability of the hydration layer around the cell due
342 to the presence of non-gelling liquid into ALG hydrogel could increase the cell
343 protection (Sun et al., 2007).

344

345 *3.5. Long-term capsule stability*

346 Together with an adequate operational stability of immobilized bacteria, capsule
347 stability studies also play a decisive role for the industrial implementation of
348 encapsulated bacteria. pH and ethanol concentration in model water solutions markedly
349 affected the stability of ALG-based capsule (Simó, Vila-Crespo, et al., 2017). The
350 weight and swelling of capsules in water increased as pH was higher from 3.0 to 3.9,
351 being more notable this behaviour for ALG capsules than for SiO₂-ALG ones. In
352 contrast, both types of capsules shrank in water-ethanol solutions with more intensity at
353 higher ethanol concentrations (from 10% to 16%) and with less severity in SiO₂-ALG
354 capsules than in ALG ones.

355 In this study, we evaluate the impact of pH and ethanol concentration on SiO₂-
356 ALG capsule stability using samples of wine at different pH (3.0, 3.3 and 3.6) and
357 alcoholic degrees (12.0%, 13.5% and 15.0%). Simultaneously, ALG capsules were also
358 tested as control samples to know the effect of silica/silicate on the capsule stability.
359 Pronounced diameter and weight reductions were found when SiO₂-ALG and ALG
360 capsules were exposed to wine, reducing the free volume in the biocapsules (Figures 4
361 and 5). These results are consistent with the shrinking process observed in SiO₂-ALG

362 and ALG materials in model water-ethanol solutions (Simó, Vila-Crespo, et al., 2017).
363 On the whole, shrinking was decreased as wine pH increased, being always the loss in
364 diameter lower for SiO₂-ALG capsules than for ALG ones (Figure 4). After 27.0 days
365 of immersion of capsules in wine, the differences of weight between both capsules were
366 not so clear as those observed for diameter, testing a less drop in weight for SiO₂-ALG
367 capsules than for ALG ones only at pH 3.0. For instance, depletions of 28.4% and
368 35.9% of diameter and weight of ALG capsules were observed at pH 3.0, respectively,
369 while they were lower for SiO₂-ALG capsules (21.3% and 31.0%, respectively). The
370 differences of shrinking between both capsules can be understood from the reduction of
371 the elasticity and the increase of robustness of capsules due to the inclusion of colloidal
372 silica and silicates into ALG hydrogels (Coradin, Nassif, & Livage, 2003; Simó, Vila-
373 Crespo, et al., 2017). Hydrogels based on ALG tend to shrink when they are exposed to
374 acidic environment (Pasparakis & Bouropoulos, 2006). At a pH value lower than 4.0,
375 the carboxylic groups of mannuronic and guluronic acids of ALG are protonated (pK_a
376 3.38 and 3.36, respectively) and therefore the electrostatic repulsions among these
377 groups decrease and shrinkage is favoured (Wu, Zhu, Chang, Zhang, & Xiao, 2010).

378 In parallel with the increase in the alcoholic degree of the wine, the capsules
379 decreased in diameter and weight (Figure 5). The more the ethanol increased, the more
380 the shrinking increased. To illustrate, reductions of 24.5% and 36.3% were observed at
381 15.0% of alcoholic degree for ALG capsule diameter and weight, respectively, being the
382 loss of capsule diameter and weight less for SiO₂-ALG (21.4% and 26.8%,
383 respectively). SiO₂-ALG and ALG capsules are hydrogels which composition is mainly
384 water and they are dehydrated in the presence of alcohols (Torres, Velasquez, & Brito-
385 Arias, 2011). As it was discussed above, the addition of siliceous material into ALG
386 hydrogel causes more rigidity and robustness (Simó, Vila-Crespo, et al., 2017) and

387 hence the degree of shrinking and dehydration of SiO₂-ALG capsules in alcoholic
388 solutions is lower. The higher capacity of SiO₂-ALG capsules to reduce the water loss
389 creates a more hydrophilic microenvironment (Xu et al., 2006), so improving the
390 fermentative performance of bacteria in an ethanol-rich medium as wine.

391 Finally, to evaluate the capsule stability in wine, the release of ALG polymer
392 from SiO₂-ALG and ALG capsules immersed in wines at different pH and alcoholic
393 degree was evaluated after 18 days (Figure 6). Very small amounts of released ALG
394 (less than 0.3 mg/L) were found in all wines assayed. The release of ALG was slightly
395 higher in ALG capsules than in SiO₂-ALG ones in wines at pH 3.0 and 3.3. Neither the
396 increase of pH nor the alcoholic degree modified the levels of total polysaccharides.
397 Likely, pH-dependent changes in the electrostatic interactions between SiO₂ and ALG
398 polymer as well as ethanol-subordinate variations in capsule shrinking are not involved
399 in the release of ALG polymer. The diffusion of unpolymerized ALG from the inside of
400 the capsules to the wine could explain this behaviour (Strand, Skjåk-Bræk, & Gåserød,
401 2004). These results highlight that rupture and chemical decomposition of capsules did
402 not take place under the alcoholic and acid environment of wine indicating the
403 suitability of SiO₂-ALG biocapsules for the development of MLF.

404

405 **4. Conclusion**

406 Aqueous route using colloidal SiO₂ and sodium silicate as non-alkoxide silicon
407 precursors has been proved as a very suitable strategy to improve the chemical and
408 operational stability of ALG-based biocapsules with entrapped *O. oeni* under hard
409 winemaking conditions. Wine pH and ethanol content play a notable role in the
410 operational stability of both SiO₂-ALG biocapsules and free bacteria being their

411 negative impact most severe for unencapsulated bacteria. SiO₂-ALG biocapsules could
412 be reused for at least five cycles in wines at pH 3.0-3.6 as well as with alcoholic degrees
413 of 12.0%-15.0%. However, unencapsulated bacteria could be successfully reused
414 neither in wines at pH lower than 3.6 nor in wines with alcoholic degree higher than
415 12.0%. Combination of ALG and siliceous material enhanced the long-term chemical
416 stability of biocapsules in wines. These results highlight that the implementation of
417 SiO₂-ALG biocapsules with entrapped *O. oeni* may be a promising technique for highly
418 efficient processes in winemaking.

419

420 **Acknowledgements**

421 We are grateful to Diego Madrigal for the analytical work of this study, to
422 Lallemand for the bacteria samples and to Dr. Jesus Martín Gil for his assistance during
423 FTIR analysis.

424 **Declarations of interest:** none

425 **Funding:** this research did not receive any specific grant from funding agencies in the
426 public, commercial, or not-for-profit sectors.

427

428 **References**

429 Agouridis, N., Bekatorou, A., Nigam, P., & Kanellaki, M. (2005). Malolactic
430 fermentation in wine with *Lactobacillus casei* cells immobilized on delignified
431 cellulosic material. *Journal of Agricultural and Food Chemistry*, 53(7), 2546–
432 2551. <http://doi.org/10.1021/jf048736t>

433 Bauer, R., & Dicks, L. M. T. (2004). Control of malolactic fermentation in wine. A

- 434 review. *South African Journal of Enology and Viticulture*, 25(2), 74–88.
- 435 Betteridge, A., Grbin, P., & Jiranek, V. (2015). Improving *Oenococcus oeni* to
436 overcome challenges of wine malolactic fermentation. *Trends in Biotechnology*,
437 33(9), 547–553. <http://doi.org/10.1016/j.tibtech.2015.06.008>
- 438 Bonomo, M. G., Di Tomaso, K., Calabrone, L., & Salzano, G. (2018). Ethanol stress in
439 *Oenococcus oeni*: transcriptional response and complex physiological
440 mechanisms. *Journal of Applied Microbiology*. 125(1), 2-15.
441 <http://doi.org/10.1111/jam.13711>
- 442 Callone, E., Campostrini, R., Carturan, G., Cavazza, A., & Guzzon, R. (2008).
443 Immobilization of yeast and bacteria cells in alginate microbeads coated with silica
444 membranes: procedures, physico-chemical features and bioactivity. *Journal of*
445 *Materials Chemistry*, 18(40), 4839–4848. <http://doi.org/10.1039/b807301e>
- 446 Carturan, G., Dal Toso, R., Boninsegna, S., & Dal Monte, R. (2004). Encapsulation of
447 functional cells by sol-gel silica: actual progress and perspectives for cell therapy.
448 *Journal of Materials Chemistry*, 14(14), 2087–2098.
449 <http://doi.org/10.1039/b401450b>
- 450 Ching, S. H., Bansal, N., & Bhandari, B. (2015). Alginate gel particles-a review of
451 production techniques and physical properties. *Critical Reviews in Food Science*
452 *and Nutrition*, 0. <http://doi.org/http://dx.doi.org/10.1080/10408398.2014.965773>
- 453 Coradin, T., Allouche, J., Boissiere, M., & Livage, J. (2006). Sol-gel biopolymer/silica
454 nanocomposites in biotechnology. *Current Nanoscience*, 2(3), 219–230.
455 <http://doi.org/10.2174/1573413710602030219>
- 456 Coradin, T., & Livage, J. (2003). Synthesis and characterization of alginate/silica

457 biocomposites. *Journal of Sol-Gel Science and Technology*, 26(1–3), 1165–1168.
458 <http://doi.org/10.1023/A:1020787514512>

459 Coradin, T., Nassif, N., & Livage, J. (2003). Silica-alginate composites for
460 microencapsulation. *Applied Microbiology and Biotechnology*, 61(5–6), 429–434.
461 <http://doi.org/10.1007/s00253-003-1308-5>

462 Gómez-Ordóñez, E., & Rupérez, P. (2011). FTIR-ATR spectroscopy as a tool for
463 polysaccharide identification in edible brown and red seaweeds. *Food*
464 *Hydrocolloids*, 25(6), 1514–1520. <http://doi.org/10.1016/j.foodhyd.2011.02.009>

465 Guzzon, R., Carturan, G., Krieger-Weber, S., & Cavazza, A. (2012). Use of organo-
466 silica immobilized bacteria produced in a pilot scale plant to induce malolactic
467 fermentation in wines that contain lysozyme. *Annals of Microbiology*, 62(1), 381–
468 390. <http://doi.org/10.1007/s13213-011-0272-z>

469 Hwang, E. T., & Gu, M. B. (2013). Enzyme stabilization by nano/microsized hybrid
470 materials. *Engineering in Life Sciences*, 13(1), 49–61.
471 <http://doi.org/10.1002/elsc.201100225>

472 Knoll, C., Fritsch, S., Schnell, S., Grossmann, M., Rauhut, D., & Du Toit, M. (2011).
473 Influence of pH and ethanol on malolactic fermentation and volatile aroma
474 compound composition in white wines. *LWT - Food Science and Technology*,
475 44(10), 2077–2086. <http://doi.org/10.1016/j.lwt.2011.05.009>

476 Kourkoutas, Y., Bekatorou, A., Banat, I. M., Marchant, R., & Koutinas, A. A. (2004).
477 Immobilization technologies and support materials suitable in alcohol beverages
478 production: A review. *Food Microbiology*, 21(4), 377–397.
479 <http://doi.org/10.1016/j.fm.2003.10.005>

- 480 Kourkoutas, Y., Manojlović, V., & Nedović, V. A. (2010). Immobilization of microbial
481 cells for alcoholic and malolactic fermentation of wine and cider. In N. Zuidam &
482 V. Nedovic (Eds.), *Encapsulation Technologies for Active Food Ingredients and*
483 *Food Processing* (pp. 327–343). New York, USA: Springer.
484 http://doi.org/10.1007/978-1-4419-1008-0_12
- 485 Kuncová, G., & Trogl, J. (2010). Physiology of microorganisms immobilized into
486 inorganic polymers. In Desiree A. Morrison (Ed.), *Handbook of Inorganic*
487 *Chemistry Research* (pp. 53–101). New York, USA: Nova Science Publishers, Inc.
- 488 Maicas, S. (2001). The use of alternative technologies to develop malolactic
489 fermentation in wine. *Applied Microbiology and Biotechnology*.
490 <http://doi.org/10.1007/s002530100662>
- 491 Maicas, S., Pardo, I., & Ferrer, S. (2000). The effects of freezing and freeze-drying of
492 *Oenococcus oeni* upon induction of malolactic fermentation in red wine.
493 *International Journal of Food Science & Technology*, 35, 75–79.
494 <http://doi.org/10.1046/j.1365-2621.2000.00359.x>
- 495 Maicas, S., Pardo, I., & Ferrer, S. (2001). The potential of positively-charged cellulose
496 sponge for malolactic fermentation of wine, using *Oenococcus oeni*. *Enzyme and*
497 *Microbial Technology*, 28(4–5), 415–419. [http://doi.org/10.1016/S0141-](http://doi.org/10.1016/S0141-0229(00)00339-2)
498 [0229\(00\)00339-2](http://doi.org/10.1016/S0141-0229(00)00339-2)
- 499 Niederberger, M., & Pinna, N. (2009). Aqueous and nonaqueous sol-gel chemistry. In
500 *Metal Oxide Nanoparticles in Organic Solvents: Synthesis, Formation, Assembly*
501 *and Application* (pp. 7–18). London, UK: Springer London.
502 http://doi.org/10.1007/978-1-84882-671-7_2
- 503 Olguín, N., Champomier-Vergès, M., Anglade, P., Baraige, F., Cordero-Otero, R.,

504 Bordons, A., Zagorec, M, Reguant, C. (2015). Transcriptomic and proteomic
505 analysis of *Oenococcus oeni* PSU-1 response to ethanol shock. *Food*
506 *Microbiology*, 51, 87–95. <http://doi.org/10.1016/j.fm.2015.05.005>

507 Pannier, A., Soltmann, U., Soltmann, B., Altenburger, R., & Schmitt-Jansen, M. (2014).
508 Alginate/silica hybrid materials for immobilization of green microalgae *Chlorella*
509 *vulgaris* for cell-based sensor arrays. *Journal of Materials Chemistry B*, 2(45),
510 7896–7909. <http://doi.org/10.1039/c4tb00944d>

511 Pasparakis, G., & Bouropoulos, N. (2006). Swelling studies and in vitro release of
512 verapamil from calcium alginate and calcium alginate-chitosan beads.
513 *International Journal of Pharmaceutics*, 323(1–2), 34–42.
514 <http://doi.org/10.1016/j.ijpharm.2006.05.054>

515 Rodríguez-Nogales, J. M., Vila-Crespo, J., & Fernández-Fernández, E. (2013).
516 Immobilization of *Oenococcus oeni* in Lentikats® to develop malolactic
517 fermentation in wines. *Biotechnology Progress*, 29(1), 60–65.

518 Rosi, I., Fia, G., & Canuti, V. (2003). Influence of different pH values and inoculation
519 time on the growth and malolactic activity of a strain of *Oenococcus oeni*.
520 *Australian Journal of Grape and Wine Research*, 9(3), 194–199.
521 <http://doi.org/10.1111/j.1755-0238.2003.tb00270.x>

522 Segarra, I., Lao, C., Lopez, E., & De La Torre, M. C. (1995). Spectrophotometric
523 methods for the analysis of polysaccharide levels in winemaking products.
524 *American Journal of Enology and Viticulture*, 46(4), 564–570.

525 Simó, G., Fernández-Fernández, E., Vila-Crespo, J., Ruipérez, V., &
526 Rodríguez-Nogales, J. M. (2017a). Research progress in coating techniques of
527 alginate gel polymer for cell encapsulation. *Carbohydrate Polymers*. 170, 1-14.

528 <http://doi.org/10.1016/j.carbpol.2017.04.013>

529 Simó, G., Fernández-Fernández, E., Vila-Crespo, J., Ruipérez, V., &
530 Rodríguez-Nogales, J. M. (2017b). Silica-alginate encapsulated bacteria to enhance
531 malolactic fermentation performance in a stressful environment. *Australian*
532 *Journal of Grape and Wine Research*, 23(3), 342–349.
533 <http://doi.org/10.1111/ajgw.12302>

534 Simó, G., Vila-Crespo, J., Fernández-Fernández, E., Ruipérez, V., & Rodríguez-
535 Nogales, J. M. (2017). Highly efficient malolactic fermentation of red wine using
536 encapsulated bacteria in a robust biocomposite of silica-alginate. *Journal of*
537 *Agricultural and Food Chemistry*, 65(25), 5188–5197.
538 <http://doi.org/10.1021/acs.jafc.7b01210>

539 Sosnik, A. (2014). Alginate particles as platform for drug delivery by the oral route:
540 state-of-the-art. *ISRN Pharmaceutics*, 2014, 926157.
541 <http://doi.org/10.1155/2014/926157>

542 Strand, B. L., Skjåk-Bræk, G., & Gåserød, O. (2004). Microcapsule formulation and
543 formation. In V. A. Nedović & R. Willaert (Eds.), *Fundamentals of cell*
544 *immobilisation biotechnology* (pp. 165–183). Dordrecht, Netherlands: Springer.
545 http://doi.org/10.1007/978-94-017-1638-3_9

546 Sumby, K. M., Grbin, P. R., & Jiranek, V. (2014). Implications of new research and
547 technologies for malolactic fermentation in wine. *Applied Microbiology and*
548 *Biotechnology*, 98(19), 8111–8132. <http://doi.org/10.1007/s00253-014-5976-0>

549 Sun, Z. J., Lv, G. J., Li, S. Y., Yu, W. T., Wang, W., Xie, Y. B., & Ma, X. (2007).
550 Differential role of microenvironment in microencapsulation for improved cell
551 tolerance to stress. *Applied Microbiology and Biotechnology*, 75(6), 1419–1427.

552 <http://doi.org/10.1007/s00253-007-0960-6>

553 Torres, L. G., Velasquez, A., & Brito-Arias, M. A. (2011). Ca-alginate spheres behavior
554 in presence of some solvents and water-solvent mixtures. *Advances in Bioscience
555 and Biotechnology*, 2(1), 8–12. <http://doi.org/10.4236/abb.2011.21002>

556 Wang, J. Y., Yu, H. R., Xie, R., Ju, X. J., Yu, Y. L., Chu, L. Y., & Zhang, Z. (2013).
557 Alginate/protamine/silica hybrid capsules with ultrathin membranes for laccase
558 immobilization. *AIChE Journal*, 59(2), 380–389. <http://doi.org/10.1002/aic.13834>

559 Wu, C., Zhu, Y., Chang, J., Zhang, Y., & Xiao, Y. (2010). Bioactive inorganic-
560 materials/alginate composite microspheres with controllable drug-delivery ability.
561 *Journal of Biomedical Materials Research - Part B Applied Biomaterials*, 94(1),
562 32–43. <http://doi.org/10.1002/jbm.b.31621>

563 Xu, S., Jiang, Z., Lu, Y., Wu, H., & Yuan, W.-K. (2006). Preparation and catalytic
564 properties of novel alginate–silica–dehydrogenase hybrid biocomposite beads.
565 *Industrial & Engineering Chemistry Research*, 45(2), 511–517.
566 <http://doi.org/10.1021/ie050940y>

567 Xu, S. W., Lu, Y., Li, J., Jiang, Z. Y., & Wu, H. (2006). Efficient conversion of CO₂ to
568 methanol catalyzed by three dehydrogenases co-encapsulated in an alginate-silica
569 (ALG-SiO₂) hybrid gel. *Industrial and Engineering Chemistry Research*, 45(13),
570 4567–4573. <http://doi.org/10.1021/ie0514071>

571 Yi, Y., Neufeld, R., & Poncelet, D. (2004). Immobilization of cells in polysaccharide
572 gels. In S. Dumitriu (Ed.), *Polysaccharides: Structural Diversity and Functional
573 Versatility* (pp. 867–891). New York, USA: CRC Press.

574

575

576

577

578

579

580

581

582

583

584

585 **Figure captions**

586 **Fig. 1.** Optical images of (a) alginate and (b) SiO₂-alginate biocapsules with entrapped *Oenococcus oeni*.
587 (c-f) Environmental scanning electron microscopy (ESEM) of alginate (ALG) and SiO₂-alginate (SiO₂-
588 ALG) biocapsules with entrapped *Oenococcus oeni*. (c,e) External structure of the biocapsules. (d,f)
589 Internal structure of the biocapsules.

590 **Fig. 2.** FTIR-ATR spectra of (a) alginate and (b) SiO₂-alginate capsules.

591 **Fig. 3.** Conversion of L-malic acid by SiO₂-alginate encapsulated (a-b) and free (c-d) *Oenococcus oeni* in
592 red wines at different pH (3.0, 3.3, and 3.6) and different alcohol degree (12.0, 13.5, and 15.0%) in five
593 subsequent cycles. MLF conditions: red wine (3.0 g/L of L-malic acid), 0.03 g of biocapsules per mL of
594 wine, bacterial load of $\sim 9 \times 10^7$ cfu/mL of wine, temperature of 22°C, and 120 h of incubation. Standard
595 deviations of the assays are represented by error bars (n=3). Different letters in the bars indicate a
596 significant difference between samples ($p < 0.05$).

597 **Fig. 4.** Evolution of shrinking (expressed as decrease of diameter (in %) and weight (in %)) of alginate
598 (●) and SiO₂-alginate (■) biocapsules loaded with *Oenococcus oeni* submerged in red wines at different
599 pH ((a-b) 3.0, (c-d) 3.3 and (e-f) 3.6). Shrinking was calculated from the average of 20 biocapsules
600 submerged in wine at 22°C. Standard deviations of the assays are represented by error bars. * indicates a
601 statistically significant difference ($p < 0.05$) between both capsules.

602 **Fig. 5.** Evolution of shrinking (expressed as decrease of diameter (in %) and weight (in %)) of alginate
603 (●) and SiO₂-alginate (■) biocapsules loaded with *Oenococcus oeni* submerged in red wines at different
604 alcohol degree ((a-b) 12.0%, (c-d) 13.5% and (e-f) 15.0%). Shrinking was calculated from the average of
605 20 biocapsules submerged in wine at 22°C Standard deviations of the assays are represented by error
606 bars. * indicates a statistically significant difference ($p < 0.05$) between both capsules.

607 **Fig. 6.** Release of alginate (expressed as total polysaccharides) from alginate (■) and SiO₂-alginate (□)
608 biocapsules loaded with *Oenococcus oeni* submerged in red wines at different pH (3.0, 3.3 and 3.6) and
609 alcohol degree (12.0%, 13.5% and 15.0%) after 18 days. Standard deviations of the assays are represented
610 by error bars. Different letters in the bars indicate a significant difference between samples ($p < 0.05$).

611

612

613

614

Guillermo Simó

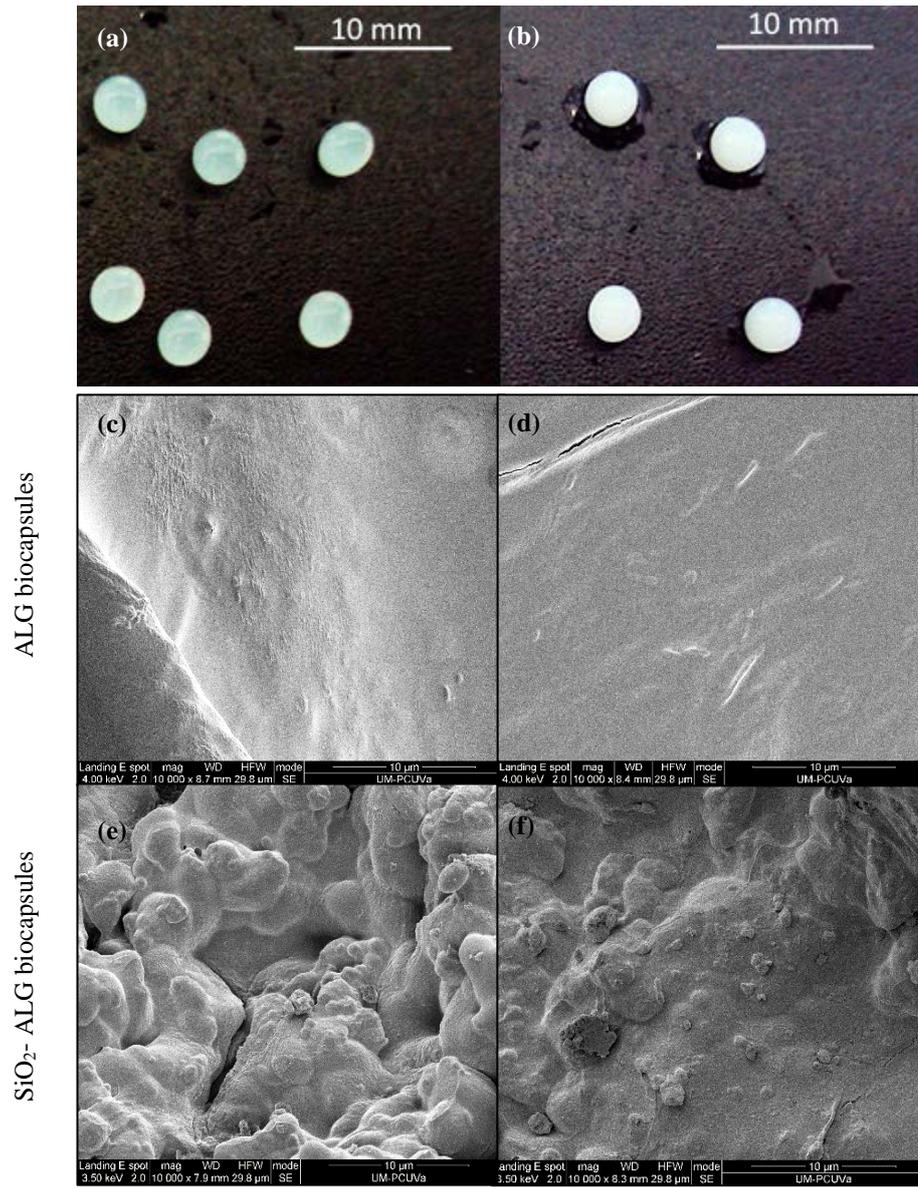


Fig. 1.

Guillermo Simó

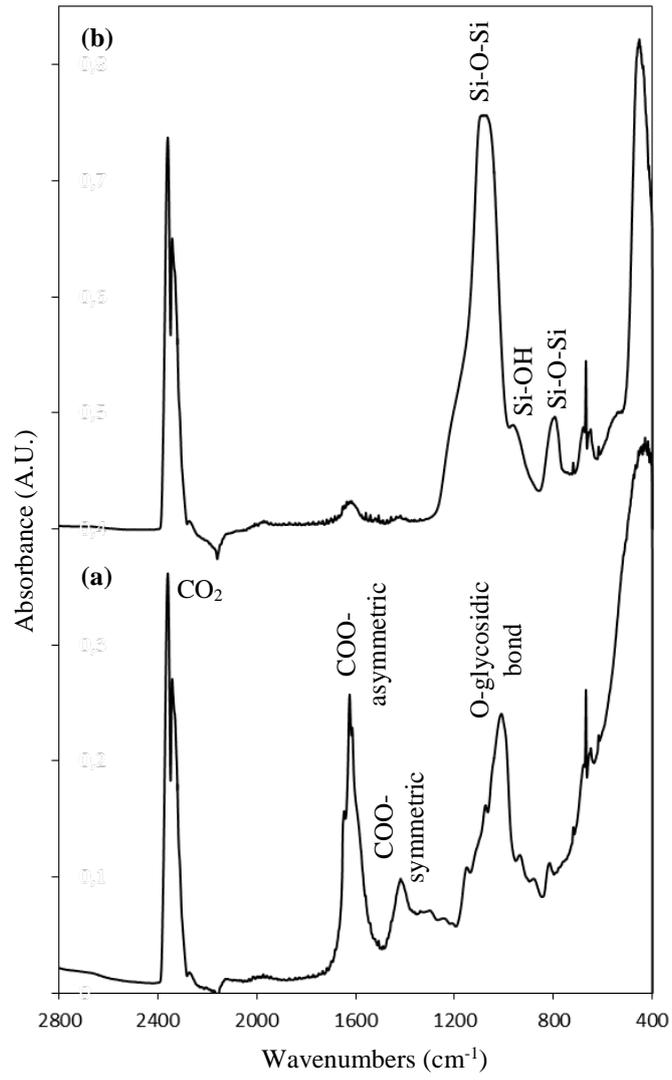


Fig. 2.

Guillermo Simó

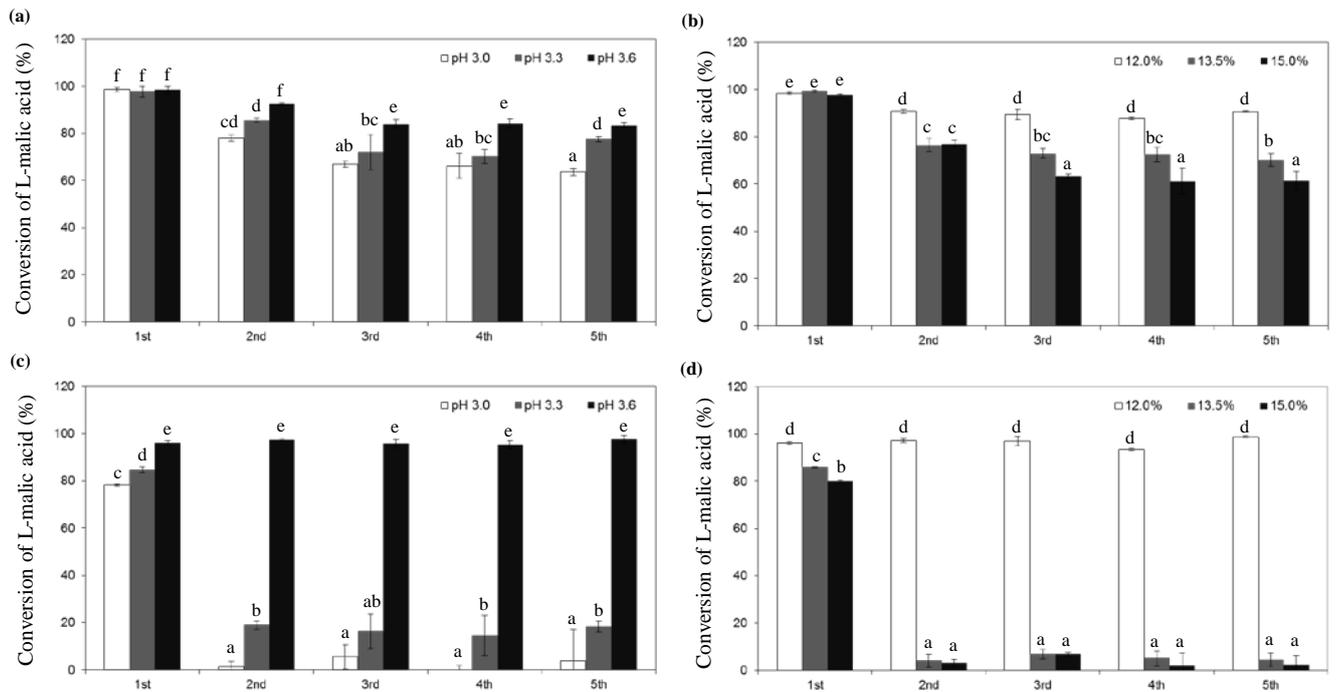


Fig. 3.

Guillermo Simó

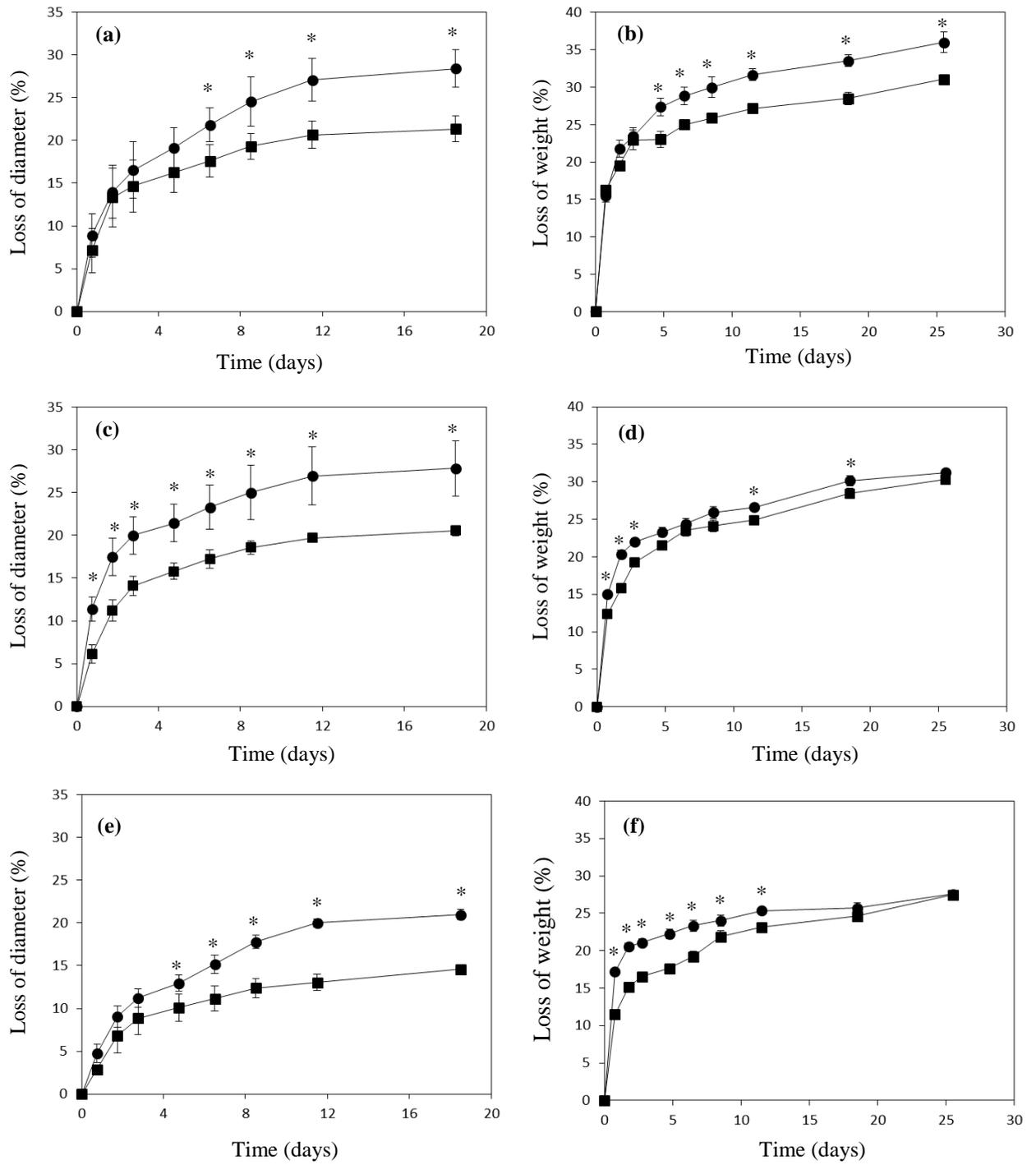


Fig. 4.

Guillermo Simó

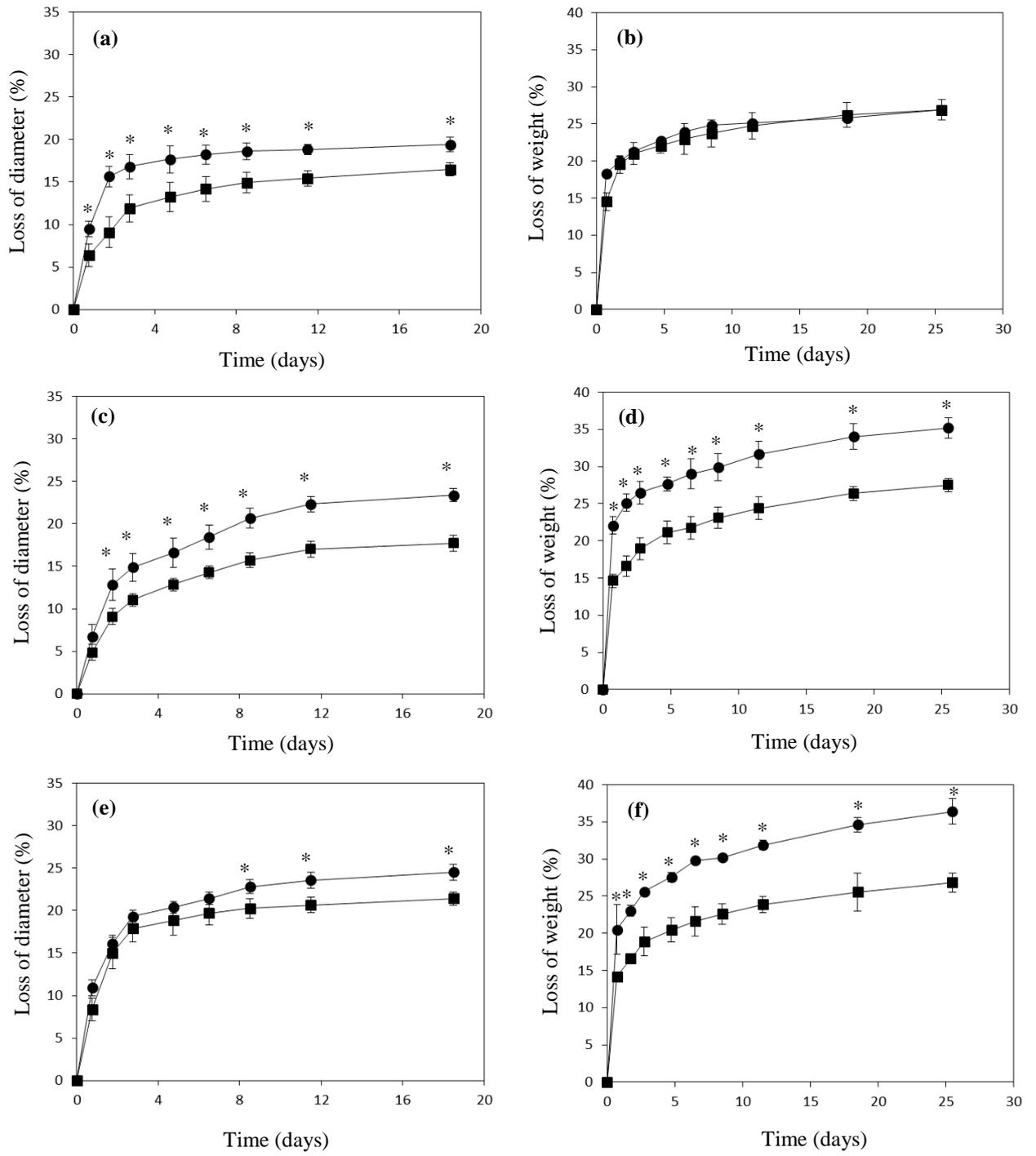


Fig. 5.

Guillermo Simó

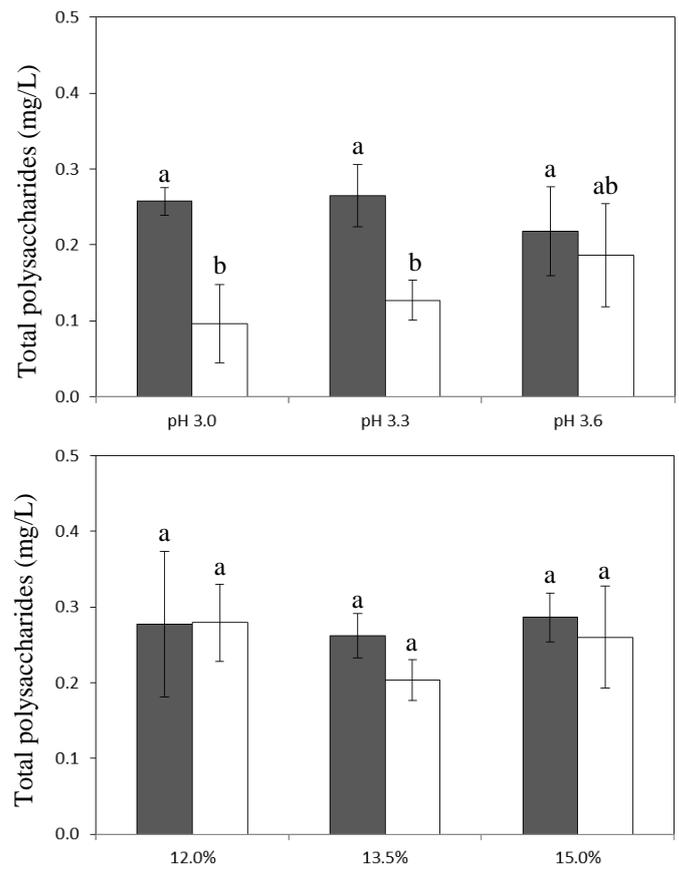


Fig. 6.