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Abstract: The present study aimed at maximizing the performance of a standard biotrickling filter (BTF) devoted to the treatment of CH<sub>4</sub> at low concentrations by enhancing the mass transfer using optimum liquid recycling rates and an innovative gas recycling strategy. Internal gas recycling favored CH<sub>4</sub> abatement in the early stages of BTF operation and supported stable elimination capacities (ECs) above 30 g m<sup>-3</sup> h<sup>-1</sup> at an empty bed residence time of 4 min and a liquid recycling velocity of 5 m h<sup>-1</sup>, which represented the highest ECs achieved in single phase BTFs to date. A comprehensive energy analysis confirmed that internal gas recycling could increase CH<sub>4</sub> abatement by 50% at only 10% higher operating costs. The BTF exhibited a high microbial diversity (Shannon-Wiener indices of 2.5-2.8) dominated by Type I methanotrophs, likely due to the presence of high Cu<sup>2+</sup> concentrations. Mass transfer limitations from the aqueous phase to the microorganisms, attributed to biomass accumulation in the packing material, were identified under the long term operation.

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Marc Deshusses  
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Dear Professor Deshusses,

Please find enclosed our unpublished original manuscript "**Methane abatement in a gas-recycling biotrickling filter: evaluating innovative operational strategies to overcome mass transfer limitations**" co-authored by José M. Estrada, Raquel Lebrero, Guillermo Quijano, Rebeca Pérez, Ivonne Figueroa, Pedro A. García-Encina, and Raúl Muñoz. All authors are aware of the CEJ ethics policy, declare no conflict of interest and accept the responsibility for the present manuscript. This manuscript has been prepared according to the CEJ guide for authors.

The increased public awareness of environmental problems and the urgent need to reduce anthropogenic GHG emissions worldwide are promoting an intensive research on the development of cost-effective and environmentally friendly CH<sub>4</sub> abatement technologies. The present study aimed at maximizing the abatement capacity of a standard, single-phase BTF treating dilute CH<sub>4</sub> emissions using an innovative internal gas recycling strategy to overcome previously reported gas-liquid mass transfer limitations. The BTF achieved high CH<sub>4</sub> elimination capacities and unexpected liquid-biofilm mass transfer limitations associated to biomass overgrowth were identified.

In order to avoid any potential conflict of interests based on the current good relationships between Spanish groups working in the field of biofiltration, the authors would highly appreciate the absence of Spanish researchers/academics in the peer-review of this work.

Based on the recent and increasing relevance of anthropogenic GHG emissions and their implications on climate change, we strongly believe that this paper fits well in the scope of *Chemical Engineering Journal* and will certainly attract international attention.

We look forward to your evaluation.  
Best regards,

**Raúl Muñoz**

**José M. Estrada**

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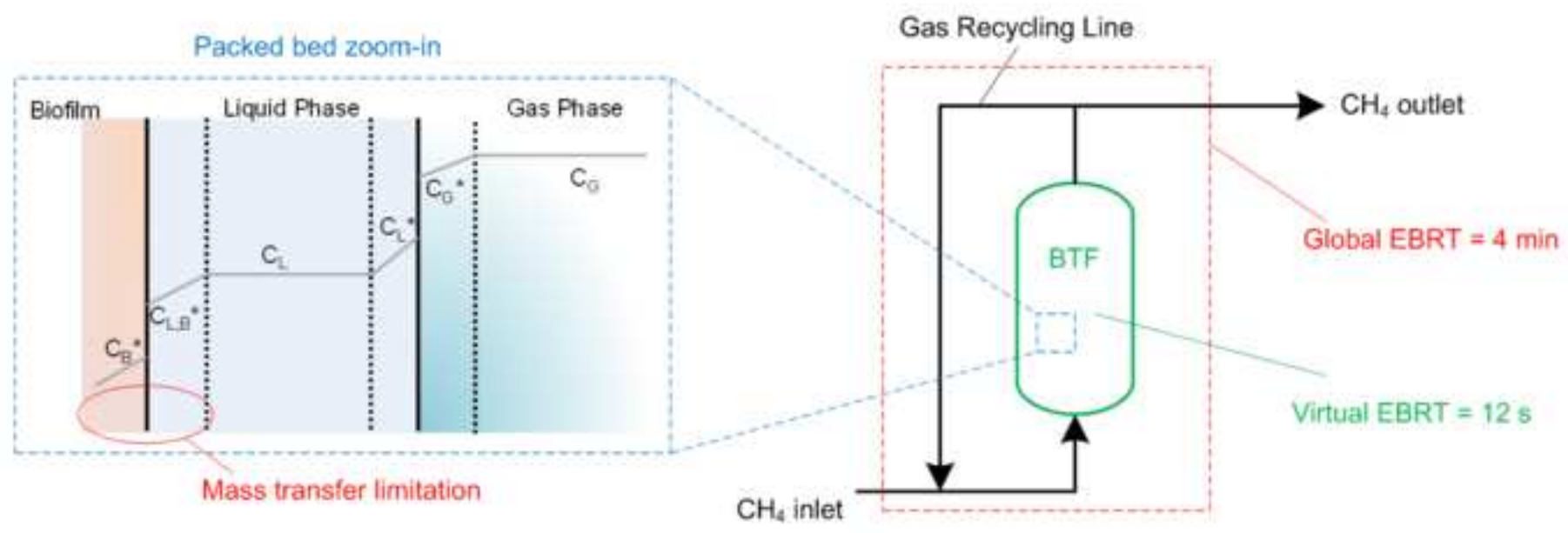
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## \*Highlights (for review)

- Internal gas recycling in a BTF was successful at enhancing CH<sub>4</sub> removal.
- Stable CH<sub>4</sub> elimination capacities above 30 g m<sup>-3</sup> h<sup>-1</sup> were obtained.
- Type I methanotrophs were dominant in the highly diverse community in the BTF.
- The BTF faced non gas-liquid mass transfer limitations due to biomass overgrowth.

1           **Methane abatement in a gas-recycling biotrickling filter:**  
2           **evaluating innovative operational strategies to overcome mass**  
3           **transfer limitations**

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10          **Abstract**

11          The present study aimed at maximizing the performance of a standard biotrickling filter  
12          (BTF) devoted to the treatment of CH<sub>4</sub> at low concentrations by enhancing the mass  
13          transfer using optimum liquid recycling rates and an innovative gas recycling strategy.  
14          Internal gas recycling favored CH<sub>4</sub> abatement in the early stages of BTF operation and  
15          supported stable elimination capacities (ECs) above 30 g m<sup>-3</sup> h<sup>-1</sup> at an empty bed  
16          residence time of 4 min and a liquid recycling velocity of 5 m h<sup>-1</sup>, which represented the  
17          highest ECs achieved in single phase BTFs to date. A comprehensive energy analysis  
18          confirmed that internal gas recycling could increase CH<sub>4</sub> abatement by 50% at only  
19          10% higher operating costs. The BTF exhibited a high microbial diversity (Shannon-  
20          Wiener indices of 2.5-2.8) dominated by Type I methanotrophs, likely due to the  
21          presence of high Cu<sup>2+</sup> concentrations. Mass transfer limitations from the aqueous phase  
22          to the microorganisms, attributed to biomass accumulation in the packing material, were  
23          identified under the long term operation.

25          **Keywords:** biotrickling filter, greenhouse gas, mass transfer, methane, polyurethane  
26          foam.



## 28 **1.Introduction**

29 Methane, with a global warming potential 20 times higher than that of CO<sub>2</sub>, is  
30 nowadays the second most relevant greenhouse gas (GHG) emitted to the atmosphere.

31 Atmospheric CH<sub>4</sub> concentrations in 2011 exceeded pre-industrial levels by 150% [1, 2],  
32 with anthropogenic emissions representing 50-65 % of the total CH<sub>4</sub>emission inventory  
33 worldwide [1]. In this context, the increased public awareness of environmental  
34 problems and the urgent need to reduce anthropogenic GHG emissions worldwide are  
35 promoting an intensive research on the development of cost-effective and  
36 environmentally friendly CH<sub>4</sub> abatement technologies.

37  
38 Methane emissions not suitable for energy recovery (methane content < 30%) have been  
39 traditionally treated using flaring or incineration as end-of-the-pipe technologies [3].  
40 Unfortunately, while these oxidation technologies are only cost-effective for emissions  
41 containing CH<sub>4</sub> concentrations over 20%, more than 50% of the anthropogenic CH<sub>4</sub> is  
42 emitted at concentrations below 3% [4]. Dilute CH<sub>4</sub> emissions are typically found in old  
43 landfills fugitive emissions or gas recovery systems (0-20%), in ventilated coal mines  
44 (0.1 – 1 %) or in covered liquid manure storage tanks (0-3%) [5-9]. In this regard,  
45 biological technologies represent a promising end-of-the-pipe solution for the treatment  
46 of dilute off-gas emissions, biotrickling filtration being one of the most cost-effective  
47 configurations due to its robustness and low operating costs [10, 11].

48  
49 However, pollutant mass transfer limitations often reduce the abatement potential and  
50 hinder the full-scale application of biotrickling filters (BTFs) devoted to the treatment  
51 of highly hydrophobic compounds such as CH<sub>4</sub>[12]. Most recent research studies have  
52 focused on CH<sub>4</sub> mass transfer enhancement by either applying complex bioreactor

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53 configurations such as horizontal biofilm, airlift or tailor flow reactors [13-15] or by  
54 adding non-aqueous phases and surfactants to conventional bioreactor configurations [4,  
55 16]. However, both approaches have resulted in limited elimination capacities and  
56 entailed high operating costs [16]. Therefore, the development of simple and cost-  
57 effective bioreactor configurations and operational strategies devoted to CH<sub>4</sub> abatement  
58 will be crucial in the global fight against climate change.

59  
60 The present study aimed at maximizing the abatement capacity of a standard, single-  
61 phase BTF treating dilute CH<sub>4</sub> emissions. First, the influence of the gas empty bed  
62 residence time (EBRT) and the linear liquid recycling velocity (U<sub>L</sub>) on the abiotic  
63 k<sub>LaCH<sub>4</sub></sub> and pressure drop in the BTF was characterized. Secondly, the influence of U<sub>L</sub>,  
64 internal gas recycling and liquid media renewal rate on the CH<sub>4</sub> biodegradation  
65 performance of the BTF was evaluated. Internal gas recycling constitutes an innovative  
66 mass transfer enhancement approach based on the decoupling of the gas-liquid  
67 turbulence inside the reactor from the actual gas residence time. Finally, the dynamics  
68 of the microbial community structure responsible for CH<sub>4</sub> biodegradation were  
69 elucidated.

## 70 71 **2. Materials and Methods**

### 72 *2.1 Chemicals*

73 The mineral salt medium (MSM) used during the experimentation was a modified  
74 Brunner medium consisting of (g L<sup>-1</sup>): Na<sub>2</sub>HPO<sub>4</sub>·12H<sub>2</sub>O, 6.15; KH<sub>2</sub>PO<sub>4</sub>, 1.52; NaNO<sub>3</sub>,  
75 0.61 (used instead of (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> to prevent the inhibition of methanotrophs by ammonia  
76 [17]); MgSO<sub>4</sub>·7H<sub>2</sub>O, 0.2; CaCl<sub>2</sub>·2H<sub>2</sub>O, 0.05; EDTA, 0.005; FeSO<sub>4</sub>·7H<sub>2</sub>O, 0.002;  
77 H<sub>3</sub>BO<sub>3</sub>, 0.0003; CoCl<sub>2</sub>·6H<sub>2</sub>O, 0.0002; ZnSO<sub>4</sub>·7H<sub>2</sub>O, 0.0001; Na<sub>2</sub>MoO<sub>4</sub>·2H<sub>2</sub>O, 0.00003;

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78  $\text{MnCl}_2 \cdot 4\text{H}_2\text{O}$ , 0.00003;  $\text{NiCl}_2 \cdot 6\text{H}_2\text{O}$ , 0.00002;  $\text{CuCl}_2 \cdot 2\text{H}_2\text{O}$ , 0.00001.  $\text{Cu}^{2+}$  was  
79 supplemented to the MSM from a  $10 \text{ g L}^{-1}$   $\text{CuSO}_4$  stock solution to the target  
80 concentrations in order to avoid copper limitations. All chemicals were purchased from  
81 Panreac (Spain) with a purity higher than 99.0%. Methane (99.5% purity) and nitrogen  
82 (99.9% Purity) were supplied by Abello-Linde, S.A. (Spain), while silicone oil 200 cSt  
83 (99.9% purity) was purchased from Sigma Aldrich (USA).

84

## 85 *2.2 Inoculum*

86 The BTF was inoculated with methanotrophic cultures enriched from aerobic activated  
87 sludge from Valladolid wastewater treatment plant (Valladolid, Spain). Sludge samples  
88 were acclimated separately to  $\text{CH}_4$  degradation for 37 days at  $\text{Cu}^{2+}$  concentrations of 5,  
89 10, 25 and  $50 \mu\text{M}$  in order to assess the influence of copper concentration on methane  
90 biodegradation. Methanotrophic cultures were enriched at  $25 \text{ }^\circ\text{C}$  in 1250 mL bottles  
91 containing 500 mL MSM and batchwise fed (8 amendments) with  $\text{CH}_4$  at initial  
92 headspace concentrations of  $\approx 14 \text{ g m}^{-3}$ . Based on the negligible influence of  $\text{Cu}^{2+}$   
93 concentration on the  $\text{CH}_4$  biodegradation rate (data not shown), the BTF was inoculated  
94 with 300 mL of each culture and further operated at  $10 \mu\text{M Cu}^{2+}$ .

95

## 96 *2.3 Experimental set-up*

97 A laboratory scale BTF consisting of a cylindrical jacketed PVC column (0.08 m inner  
98 diameter) was packed with polyurethane foam (PUF) to a working packed bed volume  
99 of 4 L. The packing material consisted of  $1 \text{ cm}^3$  PUF cubes (Filtren TM 25280, Recticel  
100 Iberica S.L.) with a net density of  $20\text{--}24 \text{ kg m}^{-3}$  and a specific surface area of  $1000 \text{ m}^2$   
101  $\text{m}^{-3}$ . MSM ( $1.2 \pm 0.2 \text{ L}$ ) was continuously recycled into the BTF from an external 1.2 L

102 jacketed holding tank stirred at 700 rpm (Agimatic-S, Selecta<sup>®</sup>, Spain) (Figure 1). All  
103 experiments were carried out at 20°C.

104

#### 105 *2.4 Influence of the EBRT and liquid recycling on $k_L a_{CH_4}$ and pressure drop*

106 The overall volumetric mass transfer coefficients for O<sub>2</sub> were determined at EBRTs of  
107 12, 60, 120, and 240 s and liquid recycling velocities (U<sub>L</sub>) of 0.6, 2, 3, 4, and 5 m h<sup>-1</sup>  
108 using distilled water as the recycling liquid. N<sub>2</sub> was initially supplied to the BTF until  
109 the O<sub>2</sub> concentration in the liquid phase (recorded in the holding tank) reached ≈0 ppm.  
110 Then, air was supplied to the BTF while monitoring the increase in dissolved oxygen  
111 concentration. The experimental data were fitted to the model described by Lebrero et  
112 al. (2012) [18]. The overall  $k_L a$  values for CH<sub>4</sub> were estimated from  $k_L a_{O_2}$  using the  
113 correlation reported by Yu et al. 2006 (Equation 1) [19]:

$$114 \frac{k_L a_{CH_4}}{k_L a_{O_2}} = \frac{(1/V_{m,CH_4})^{0.4}}{(1/V_{m,O_2})^{0.4}} \quad \text{Equation 1}$$

115 where the mass transfer coefficient of a target gas pollutant ( $k_L a_{CH_4}$ ) can be estimated  
116 from the coefficient of a reference gas ( $k_L a_{O_2}$  in the present study) previously  
117 determined in the same reactor under the same operating conditions by means of the  
118 molar volumes of the gaseous compounds ( $V_{m,X}$ ).

119 The pressure drop across the packed bed was also recorded under all the EBRTs and U<sub>L</sub>  
120 tested. Tests in the un-packed BTF were also carried out at all conditions assessed in  
121 order to account exclusively for the pressure drop caused by the packed bed.

122

#### 123 *2.5 Optimization of CH<sub>4</sub> biodegradation in the BTF*

124 The synthetic methane-polluted emission fed to the BTF (15.3±0.5 g CH<sub>4</sub> m<sup>-3</sup>, 2.2±0.1  
125 %) was obtained by mixing a pure methane stream with a pre-humidified air stream in a

126 mixing chamber. The emission flow-rate and CH<sub>4</sub> concentrations were regulated by  
127 means of mass flow controllers (Aalborg, USA), resulting in an EBRT of 4 min and an  
128 overall loading rate of 229±8 g m<sup>-3</sup> h<sup>-1</sup>. The internal gas recycling was carried out using  
129 an EVO 10 compressor (Electro A.D. S.L., Spain) by re-pumping 18 L min<sup>-1</sup> from the  
130 top to the bottom of the BTF and mixing this recycled air flow with the fresh methane-  
131 polluted emission (Figure 1). This innovative operational mode allowed the BTF to  
132 operate with a global EBRT of 4 min and the gas-liquid turbulence at an effective  
133 EBRT of 12.6 s.

134 The MSM renewal rate was set at 50 mL day<sup>-1</sup> (dilution rate, D =0.045 d<sup>-1</sup>) from days 0  
135 to 47, 100 mL d<sup>-1</sup> (D=0.09d<sup>-1</sup>) from days 48 to 66, and 300 mL day<sup>-1</sup> (D=0.27d<sup>-1</sup>) from  
136 days 67 to 110 in order to avoid both nutrient limitation and the accumulation of toxic  
137 inhibitory metabolites in the recycling liquid. The liquid recycling rates tested in the  
138 BTF (200, 500, and 1500 mL min<sup>-1</sup> corresponding to U<sub>L</sub> of 2.3, 5, and 15 m h<sup>-1</sup>) were  
139 controlled by means of a Dosapro series G<sup>TM</sup> A pump (Milton Roy Ltd., USA) and a  
140 520-S pump (Watson Marlow, UK) at the highest flow rate.

141 Gas samples were periodically drawn from the sampling ports located at the inlet and  
142 outlet of the BTF to monitor the CH<sub>4</sub> and CO<sub>2</sub> concentrations. Liquid samples were  
143 periodically drawn from the stirred tank and filtered through 0.45 µm Millipore filters  
144 for the determination of pH, total organic carbon (TOC) and total nitrogen (TN)  
145 concentrations. Bed compaction was directly measured from the variations in the  
146 packed bed height in the BTF. The humidity of the inlet CH<sub>4</sub>-laden air stream was  
147 measured on-line by a thermohygrometer (Testo 605-H1, Testo AG, Germany) and  
148 ranged from 40 to 60%. Water losses by evaporation in the system were balanced by  
149 addition of distilled water to keep a constant recycling liquid volume.

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151 *2.6 Analytical methods*

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2 152 The CO<sub>2</sub> and CH<sub>4</sub> gas concentrations were determined in a Bruker 430 gas  
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4 153 chromatograph (Palo Alto, USA) coupled with a thermal conductivity detector and  
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6 154 equipped with a CP-Molsieve 5A (15m × 0.53 μm × 15 μm) and a CP-PoraBOND Q  
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8 155 (25m × 0.53 μm × 10 μm) columns. The oven, injector and detector temperatures were  
9  
10 156 maintained at 45 °C, 150 °C and 175 °C, respectively. Helium was used as the carrier  
11  
12 157 and make up gas at 6 mL min<sup>-1</sup> and 24 mL min<sup>-1</sup>, respectively. The pH was determined  
13  
14 158 using a pH-meter Basic 20 (Crison, Spain), while the concentrations of TOC and TN  
15  
16 159 were measured using a Shimadzu TOC-VCSH analyzer (Japan) equipped with a TNM-1  
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18 160 chemiluminescence module. Dissolved O<sub>2</sub> concentration in the holding tank was  
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20 161 measured by means of an oxygen probe (Consort<sup>®</sup>, Belgium) connected to a multi-  
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22 162 parameter analyzer C3020 (Consort<sup>®</sup>, Belgium) and a computer data logger as described  
23  
24 163 elsewhere [20]. Pressure drop was measured by means of a U-Tube manometer  
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26 164 connected to the inlet and outlet of the reactor using water as the manometric fluid.  
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36 166 *2.7 Microbiological procedures*

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38 167 Biomass samples from the cultures acclimated at different Cu<sup>2+</sup> concentrations (5, 10,  
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40 168 25 and 50 μM corresponding to samples A, B, C and D, respectively), from the mixed  
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42 169 BTF inoculum (sample E) and from the BTF at days 38 (sample F) and 104 (sample G)  
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44 170 were collected and stored at -20°C in order to evaluate the richness and composition of  
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46 171 the bacterial communities. The genomic DNA was extracted according to Lebrero et al.  
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48 172 (2011) [23] . The PCR mixture (50 μL) was composed of 25 μL of BIOMIX ready-to-  
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50 173 use 2× reaction mix (Bioline, Ecogen) containing reaction buffer, magnesium chloride,  
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52 174 deoxynucleotide triphosphates (dNTPs), Taq polymerase and additives, 1 or 2 μL of the  
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54 175 extracted DNA, PCR primers 968-F-GC and 1401-R (10μM) (Sigma- Aldrich, St.  
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176 Louis, MO, USA) for bacterial 16S rRNA gene amplification, and Milli-Q water up to a  
177 final volume of 50  $\mu$ L. The PCR thermo-cycling program used was previously  
178 described in Lebrero et al. (2011). The DGGE analysis of the amplicons was performed  
179 with a D-Code Universal Mutation Detection System (Bio Rad Laboratories) using 8%  
180 (w/v) polyacrylamide gel with a urea/formamide denaturing gradient from 45 to 65%.  
181 The DGGE running conditions were applied according to Roest et al. (2005) [21]. The  
182 gels were stained with GelRed Nucleic Acid Gel Stain (biotium) for 1 h and the  
183 obtained DGGE patterns processed using the GelCompar IITM software (Applied  
184 Maths BVBA, Sint-Martens-Latem, Belgium). After image normalization, bands were  
185 defined for each sample using the band search algorithm within the program. Similarity  
186 indices of the compared profiles were calculated from the densitometric curves of the  
187 scanned DGGE profiles by using the Pearson product–moment correlation coefficient  
188 [22]. The peak heights in the densitometric curves were also used to determine the  
189 Shannon–Wiener diversity index (H).

190 The most relevant bands were excised from the DGGE gel in order to identify the  
191 bacteria present in the samples above described. The procedure was previously  
192 described in Lebrero et al. (2011) [23]. The taxonomic position of the sequenced DGGE  
193 bands was obtained using the RDP classifier tool (50% confidence level) [24]. The  
194 closest matches to each band were obtained using the BLAST search tool at the NCBI  
195 (National Centre for Biotechnology Information) [25]. Sequences were deposited in  
196 GenBank Data Library under accession numbers KJ002507- KJ002532.

### 198 **3. Results and Discussion**

#### 199 *3.1 Influence of the EBRT and liquid recycling on $k_L a_{CH_4}$ and pressure drop*

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200 The overall  $k_{L\text{aCH}_4}$  values increased when increasing the liquid recycling velocity and  
201 decreasing the EBRT, with a maximum of  $280\pm 15\text{ h}^{-1}$  recorded at  $5\text{ m h}^{-1}$  and  $12\text{ s}$   
202 (Figure 2A).  $\text{CH}_4$  mass transfer exhibited a low sensitivity towards variations in the gas  
203 EBRT at low liquid recycling velocities. For instance, the overall  $k_{L\text{aCH}_4}$  increased from  
204  $37$  to  $85\text{ h}^{-1}$  (130%) when decreasing the EBRT from  $240$  to  $12\text{ s}$  at a  $U_L$  of  $0.6\text{ m h}^{-1}$ ,  
205 while this increase accounted for 220% at a  $U_L$  of  $5\text{ m h}^{-1}$ . These empirical findings  
206 were in agreement with the data reported by Kim and Deshusses (2008), where the  
207 volumetric mass transfer coefficient in the liquid film ( $k_{L\text{a}_w}$ ) for  $\text{CO}_2$  was not sensitive to  
208 variations in the gas flow-rate at the lowest  $U_L$  tested ( $0.1\text{ m h}^{-1}$ ) [26]. This also  
209 suggested that the process was limited by mass transfer in the liquid side under low  $U_L$ ,  
210 since only a moderate mass transfer improvement was observed when increasing the  
211 turbulence in the gas side. An increasing influence of the EBRT on the overall  $k_{L\text{aCH}_4}$   
212 was recorded at higher  $U_L$  ( $3, 4,$  and  $5\text{ m h}^{-1}$ ) likely due to a decrease in the mass  
213 transfer resistance in the liquid film (as a result of the higher liquid turbulence),  
214 concomitant with an enhanced transport in the gas side. Similarly, a reduced  $k_{L\text{aCH}_4}$   
215 sensitivity towards variations in  $U_L$  was recorded at high EBRTs. Therefore, these  
216 results suggest that attempts to overcome mass transfer limitations in BTFs by  
217 increasing the liquid recycling rate might be only cost-effective at low EBRTs. Internal  
218 gas-recycling can help reducing the mass transfer resistances while operating at high  
219 EBRTs, which can eventually boost  $\text{CH}_4$  abatement in BTFs [13].

220 On the other hand, no significant influence of the EBRT and liquid recycling velocity  
221 on the pressure drop across the packed bed was recorded, with a maximum pressure  
222 drop variation from  $0.1$  to  $0.3\text{ Pa m}^{-1}_{\text{bed}}$  under the conditions tested (Figure 2B). This  
223 finding was of key relevance for the implementation of internal gas-recycling strategies,  
224 since any additional energy requirement in this innovative operational mode would



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225 derive from the higher circulating flow rates rather than from an additional pressure  
226 drop mediated by the internal gas recycling.

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### 228 *3.2 Optimization of CH<sub>4</sub> biodegradation in the BTF*

229 When the BTF was operated at an EBRT of 4 min and  $U_L = 2.3 \text{ m h}^{-1}$ , the EC remained  
230 below  $2 \text{ g m}^{-3} \text{ h}^{-1}$  with stable  $\text{CO}_2$  production rates of  $\approx 10 \text{ g m}^{-3} \text{ h}^{-1}$  (Figure 3). The high  
231  $\text{CO}_2$  production, above the expected levels according to the low  $\text{CH}_4$  EC recorded, was  
232 attributed to the inoculum endogenous respiration. Then,  $U_L$  was increased to  $5 \text{ m h}^{-1}$  by  
233 day 13 in order to overcome possible mass transfer limitations in the system. This  
234 change in operational conditions corresponded to an overall abiotic  $k_{L\text{aCH}_4}$  increase  
235 from  $28 \text{ h}^{-1}$  to  $88 \text{ h}^{-1}$  and resulted in fluctuating ECs ( $4.2$  to  $16.3 \text{ g m}^{-3} \text{ h}^{-1}$ ) from day 14  
236 to 31, which confirmed the occurrence of mass transfer limitation in the liquid side  
237 during the previous operational stage.  $\text{CO}_2$  production rates gradually decreased in this  
238 period, which was attributed to the increasing contribution of anabolism (biomass  
239 growth) to  $\text{CH}_4$  biodegradation compared to process start-up where endogenous  
240 respiration was the predominant process (Figure 3B). This fact was confirmed by visual  
241 observation of the significant biomass growth on the packing material and the low  $\text{CH}_4$   
242 mineralization observed in this period ( $24 \pm 12\%$ ).

243 Internal gas recirculation was implemented on day 31 at a rate of  $18 \text{ L min}^{-1}$ , resulting  
244 in a virtual EBRT of 12 s in the packing material while maintaining a global EBRT of 4  
245 min and a  $\text{CH}_4$  loading rate of  $229 \pm 8 \text{ g m}^{-3} \text{ h}^{-1}$ . This strategy was expected to increase  
246 the mass transfer coefficient for  $\text{CH}_4$  by a factor of 3, which agreed with the EC of up to  
247  $29 \text{ g m}^{-3} \text{ h}^{-1}$  recorded by day 32 (2.5 fold EC increase). This fact confirmed the potential  
248 of internal gas recirculation to enhance  $\text{CH}_4$  mass transfer from the gas to the liquid  
249 phase. However, a sharp decrease in the EC to  $0 \text{ g m}^{-3} \text{ h}^{-1}$  by day 35 was observed,

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250 which was attributed to nutrient limitation in the system. Thus, 500 mL of recycling  
251 cultivation broth were replaced by fresh MSM at day 38, which allowed to recover an  
252 EC of  $27 \text{ g m}^{-3} \text{ h}^{-1}$  on day 39. The EC was then allowed to gradually decrease again in  
253 order to confirm a potential nutrient limitation. The absence of  $\text{CH}_4$  biodegradation  
254 recorded by day 42 coincided with negligible TN concentrations in the recycling liquid  
255 medium. Hence,  $\text{NO}_3^-$  concentrations in the BTF were daily restored from days 43 to 47  
256 by adding 12 mL of a  $100 \text{ g L}^{-1}$  stock nitrate solution, which entailed TN concentrations  
257 of  $72 \pm 32 \text{ mg L}^{-1}$  and steady EC of  $21.7 \text{ g m}^{-3} \text{ h}^{-1}$  by day 48. Therefore, nitrogen was  
258 identified as the key limiting factor under internal gas recycling at  $18 \text{ L min}^{-1}$  and  $U_L$  of  
259  $5 \text{ m h}^{-1}$ , and an increase in the MSM renewal to a  $D$  of  $0.09 \text{ day}^{-1}$  was implemented. This  
260 higher frequency in MSM exchange was able to maintain ECs of  $18.5 \pm 3.0 \text{ g m}^{-3} \text{ h}^{-1}$  for  
261 only 7 days.  $\text{CH}_4$  biodegradation performance started to decrease again by day 55 likely  
262 due to the accumulation of inhibitory biodegradation metabolites. Hence, while a  $D$  of  
263  $0.09 \text{ day}^{-1}$  was able to maintain TN concentrations at  $149 \pm 28 \text{ mg L}^{-1}$ , TOC  
264 concentration in the recycling liquid increased from 77 to  $161 \text{ mg L}^{-1}$  from day 55 to  
265 day 66. The present empirical findings were in agreement with the deterioration in  $\text{CH}_4$   
266 oxidation activity observed by Mancebo et al. (2012) in an organic packing-based  
267 biofilter at high dissolved organic carbon concentrations [27]. Therefore, a new MSM  
268 dilution rate of  $0.27 \text{ day}^{-1}$  was implemented from day 66 onward, which allowed to  
269 maintain TOC concentrations at  $\approx 100 \text{ mg L}^{-1}$  and TN concentrations  $>100 \text{ mg L}^{-1}$ . In  
270 this context, the EC gradually recovered to steady values of  $22.2 \pm 1.8 \text{ g m}^{-3} \text{ h}^{-1}$  from  
271 day 72 to 82 concomitant with a rise in  $\text{CO}_2$  production up to  $47.3 \pm 4.1 \text{ g m}^{-3} \text{ h}^{-1}$  (89%  
272  $\text{CH}_4$  mineralization).  
273 Internal gas recirculation was stopped at day 82 in order to confirm the potential of this  
274 operational strategy to enhance  $\text{CH}_4$  mass transport in the BTF operating at the real

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275 EBRT of 4 min. Surprisingly, no deterioration in CH<sub>4</sub> biodegradation was observed  
276 following the interruption of the gas recirculation, with average ECs and CO<sub>2</sub>  
277 production rates of  $22.2 \pm 2.4 \text{ g m}^{-3} \text{ h}^{-1}$  and  $51.9 \pm 2.3 \text{ g m}^{-3} \text{ h}^{-1}$ , respectively, from day  
278 82 to day 94. Based on the previous abiotic mass transfer characterization,  $k_{L,CH_4}$   
279 decreased from 280 to  $88 \text{ h}^{-1}$  when the EBRT increased from 12 s to 4 min. It can be  
280 hypothesized that biomass growth in the packed bed modified both the hydrodynamics  
281 and mass transfer processes in the BTF. Popat and Deshusses (2010) recently reported  
282 the complex and significant influence of biomass growth on mass transfer mechanisms  
283 in BTFs, where shifts in rate-governing steps were associated to a biomass-mediated  
284 modification of the interfacial area available for pollutant mass transfer [28]. Likewise,  
285 Arellano-García et al. (2013) demonstrated that the accumulation of biomass can  
286 modify the hydrodynamics of the recycling liquid from plug flow, affecting pollutant  
287 biodegradation in the BTF [29].

288 The influence of  $U_L$  on the EC in the absence of internal gas recirculation was further  
289 assessed by increasing the liquid recycling rate from 5 to  $15 \text{ m h}^{-1}$  on day 94. Despite an  
290 increase in mass transfer was expected from the extrapolation of the results presented in  
291 Figure 2A, the reactor maintained stable ECs of  $22.5 \pm 1.7 \text{ g m}^{-3} \text{ h}^{-1}$  from days 94 to 110  
292 showing a slight increase in the CO<sub>2</sub> production rates with an average value of  $54.2 \pm$   
293  $5.4 \text{ g m}^{-3} \text{ h}^{-1}$  (Figure 3). A mass transfer limitation test was carried out on day 97 in  
294 order to elucidate the rate-limiting step by increasing the inlet gas CH<sub>4</sub> concentration  
295 from 15 to  $41 \text{ g m}^{-3}$  for 3.5 h (2.8 fold increase) (Figure 4). The EC rapidly increased  
296 from 24.3 to  $63.8 \text{ g m}^{-3} \text{ h}^{-1}$  (2.6 times increase) during this step CH<sub>4</sub> load increase, and  
297 concomitantly decreased to previous steady state values of  $19 \text{ g m}^{-3} \text{ h}^{-1}$  when the inlet  
298 CH<sub>4</sub> concentration was decreased to  $15 \text{ g m}^{-3}$ . This test confirmed that CH<sub>4</sub> abatement in  
299 the BTF was mass transfer limited, ruling out a potential biological limitation [30]. In

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300 addition, the determination of CH<sub>4</sub> concentration in the liquid phase at the bottom of the  
301 column by day 100 revealed values of  $0.22 \pm 0.04 \text{ g m}^{-3}$ , which were close to the  
302 theoretical equilibrium concentration of  $0.40 \text{ g m}^{-3}$  calculated by the Henry's law. This  
303 confirmed that CH<sub>4</sub> was effectively transferred to the liquid phase but suggested that the  
304 diffusive transport through the biofilm was the limiting mass transfer process. In view  
305 of the above-mentioned results, the occurrence of a mass transfer limitation between the  
306 liquid phase and the biofilm colonizing the packed bed might be hypothesized (Figure  
307 5). This would explain the absence of increase in EC when increasing the gas velocity,  
308 and the high CH<sub>4</sub> concentrations recorded in the recycling liquid phase. Therefore, any  
309 operational modification to enhance the gas-liquid mass transfer would be unfruitful to  
310 increase CH<sub>4</sub> abatement. However, the increase in the  $U_L$  was also unsuccessful in  
311 enhancing the liquid-biofilm mass transfer, probably due to interfacial area decrease  
312 caused by biomass growth [28].

313 Finally, a significant packed bed compaction was recorded during BTF operation. A  
314 22% bed compaction was recorded by day 87, increasing up to 34% by day 98 and to  
315 35% by day 105. This high bed height decrease determined the effective EBRT and  
316 consequently the CH<sub>4</sub> load. Thus, the average ECs reported ( $\approx 22 \text{ g m}^{-3} \text{ h}^{-1}$ ) during the  
317 final stages of the experiment (days 94-110) should be corrected to account for the real  
318 packed bed volume, resulting in stable real ECs above  $30 \text{ g m}^{-3} \text{ h}^{-1}$ [31]. To the best of  
319 our knowledge, these ECs were higher than any of the previously reported ECs in the  
320 scarce literature available to date for single phase BTFs treating CH<sub>4</sub>. For instance,  
321 Avalos et al. (2012) found maximum ECs of  $\approx 10 \text{ g m}^{-3} \text{ h}^{-1}$  in a stone-based BTF  
322 operated at 4.25 min of EBRT and CH<sub>4</sub> loads of  $62 \text{ g m}^{-3} \text{ h}^{-1}$ , while Rocha-Rios et al.  
323 (2009) reached  $22 \text{ g m}^{-3} \text{ h}^{-1}$  in a polyurethane foam-packed BTF operated at 4.8 min of  
324 EBRT and CH<sub>4</sub> loads of  $140 \text{ g m}^{-3} \text{ h}^{-1}$ [32]. BTFs operated an EBRTs of 4-5 min

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325 support ECs similar to those obtained in full scale biofilters (BFs) operated at EBRTs  
326 sometimes exceeding 60 min ( $20\text{-}80\text{ g m}^{-3}\text{ h}^{-1}$ ) [3], while low EBRTs (4.3 min) in BFs  
327 often result in lower ECs (e.g.  $\approx 19\text{ g m}^{-3}\text{ h}^{-1}$ ) [33].

### 329 *3.3 Energy considerations during operation with internal gas recycling*

330 A cost-benefit analysis was conducted in order to evaluate the environmental  
331 sustainability of this operational strategy under two scenarios based on the overall  
332  $k_{L\text{ACH}_4}$  previously determined. Under reference scenario 1 ( $U_L = 5\text{ m h}^{-1}$ , EBRT = 240 s,  
333 no internal gas recycling) the BTF would require 1 energy unit and remove 1  $\text{CH}_4$  unit  
334 (or 20  $\text{CO}_2$  equivalent units) (Table 1). Thus, 20  $\text{CO}_2$  equivalents could be removed per  
335 unit of energy applied. The implementation of internal gas recycling to achieve a virtual  
336 EBRT of 12 s (by recycling 18 times the inlet flow rate) would increase the  $k_{L\text{ACH}_4}$  by a  
337 factor of 3 (Figure 2A), removing 60  $\text{CO}_2$  equivalents. In this particular scenario, the  
338 energy consumption associated to gas pumping would increase by a factor of 18 (based  
339 on the fact that pressure drop remains constant, Figure 2B), and the efficiency would  
340 decrease from 20 to 3.2 units of  $\text{CO}_2$  equivalents removed per unit of energy applied.  
341 This would result in an increase in the annual operating cost of 260 %, based on the fact  
342 that energy consumption in BTFs accounts for 22 % of the total operating costs (Estrada  
343 et al. 2012). On the other hand, an increase in the EBRT by a factor of 3 to achieve  
344 comparable ECs to those obtained under internal gas recycling would entail an increase  
345 in the operating cost of 150% (Estrada et al. 2011, 2012).

346 Nevertheless, internal gas recycling might be economically and energetically favorable  
347 under different operating conditions. The BTF would hypothetically remove 5  $\text{CO}_2$   
348 equivalents under reference scenario 2 ( $U_L = 2\text{ m h}^{-1}$ , EBRT = 240 s and no internal gas  
349 recycling) and 15  $\text{CO}_2$  equivalents when internal gas recycling decreases the virtual

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350 EBRT to 120 s. Under this internal gas recycling, the BTF would double the energy  
351 consumption, resulting in removals of 7.5 units of CO<sub>2</sub> equivalent per unit of energy  
352 applied compared to 5 in reference scenario 2 (Table 1). In this case, the increase in the  
353 annual operating costs under internal gas recycling would account for only 10%, while  
354 an increase in the EBRT by a factor of 3 to achieve comparable ECs to those obtained  
355 under gas recycling would entail an increase in the operating cost of 150% (Estrada et  
356 al. 2011, 2012). However, all these are hypothetical considerations and were not  
357 observed in the experimental operation of the BTF probably due to a biomass excessive  
358 growth which led to additional mass transfer limitations. Thus, finding optimal biomass  
359 content can be of key relevance for further optimization of the gas recycling strategy  
360 here proposed.

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### 362 *3.4 Bacterial population dynamics*

363 The structure of the bacterial communities in the inocula and BTF was elucidated by  
364 sequencing 21 bands from the DGGE gel (Figure 6). The closest matches for each band,  
365 along with its similarity percentage and sources, are shown in Table 2 (Supplementary  
366 material). The phylum *Proteobacteria* was predominant in the cultures acclimated to  
367 different copper concentration and in the biotrickling filter regardless of the operational  
368 stage with 17 bands belonging to this phylum (DGGE bands 1 to 7 and 13 to 22, Fig. 6).  
369 Most of these *Proteobacteria* were closely related to methane oxidizing bacteria  
370 (methanotrophs) and are often found at the anoxic/oxic interface of landfills, wastewater  
371 treatment plants, soils, rice paddies, peat bogs, wetlands and sediments [34, 35].  
372 Aerobic methanotrophic bacteria obtain energy via CH<sub>4</sub> to CO<sub>2</sub> oxidation based on their  
373 ability to synthesize methane mono-oxygenases [36, 37]. Most methanotrophs belonged  
374 to the bacterial phylum *Proteobacteria*, in the classes *Gammaproteobacteria* (Type I) and

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375 Alphaproteobacteria (Type II) [38]. Type I methanotrophs, which include genera such  
376 as *Methylomonas*, *Methylobacter* and *Methylococcus*, [2] produce particulate methane  
377 monooxygenase (pMMO) and possess a more efficient CH<sub>4</sub>-oxidizing metabolism than  
378 their type II counterparts [11, 35].

379 Both type I and II methanotrophs were identified in this work, type I being by far the  
380 most abundant type of methanotrophs. The closest relatives for DGGE bands 2-18 were  
381 type I methanotrophs [2], although the similarity was as low as 84 % for band 2 and 92  
382 % for band 15. The preferential enrichment of type I methanotrophs was likely due to  
383 the high Cu<sup>2+</sup> content (10 μM) in the mineral salt medium, which has been often  
384 correlated with the production of pMMO instead of the soluble form of the enzymes  
385 MMO [39]. Bands 19 and 20, which were relevant in the BTF but not in the initial  
386 inoculum belonged to the *Xanthomonagaceae* family and were related to other liquid or  
387 gas pollutant degraders. The DGGE band 21, specifically affiliated to the *Methylocystis*  
388 genus, was the only one belonging to type II methanotrophs. Bacteria belonging to the  
389 *Methylocystis* genus have been also found in biofilters treating methane [40]. Finally,  
390 while the DGGE band 22 was 100% similar to a *Betaproteobacteria* isolated from a  
391 petroleum hydrocarbon-contaminated water [41], bands 23 to 26 were affiliated with the  
392 phylum *Chlamydiae*, *Firmicutes*, *Gemmatimonadetes* and *Verrucomicrobia*, but their  
393 similarity to the closest relatives ranged from 88 to 95%.

394 The Shannon-Wiener diversity indices for the methanotrophic cultures enriched at  
395 different Cu<sup>2+</sup> concentrations revealed an increase in bacterial diversity at increasing  
396 Cu<sup>2+</sup> concentrations in the cultivation medium (1.8, 2.3, 2.4 and 2.6 at Cu<sup>+2</sup>  
397 concentrations of 5, 10, 25 and 50 μM, respectively). A high bacterial diversity (2.5-2.8)  
398 was maintained in the biotrickling filter from inoculation throughout the complete  
399 experimental period (Figure 6, samples E, F and G). However, a low similarity was

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400 observed between the inoculum and the community established in the reactor by day 38  
401 (31.8% similarity) or by the end of the experiment (27.4% similarity). A significant  
402 evolution (66.4% similarity) of the communities governing CH<sub>4</sub> oxidation in the BTF  
403 was also observed from day 38 to 104. These results showed the progressive  
404 establishment of bacterial populations with high functional resilience and redundancy  
405 capable of maintaining high ECs despite the changes in their structure [42].

406

#### 407 **4. Conclusions**

408 Internal gas recycling enhanced the performance of the BTF in the early stages of  
409 operation, which agreed with the data obtained in the abiotic mass transfer tests. Under  
410 certain operating conditions this strategy can theoretically result in a 50% improvement  
411 in the BTF energy use efficiency (CH<sub>4</sub> removal per unit of energy applied) at  
412 significantly lower operating costs than other strategies such as EBRT increase.  
413 However, the reactor faced additional mass transfer limitations not easily overcome by  
414 increasing the gas-liquid mass transfer as a result of biomass accumulation in the  
415 packed bed. CH<sub>4</sub> mass transfer from the liquid phase to the biofilm was identified as the  
416 limiting step during CH<sub>4</sub> abatement in our particular BTF. ECs higher than 30 g m<sup>-3</sup> h<sup>-1</sup>  
417 were achieved at an EBRT of 4 min and U<sub>L</sub> of 5 m h<sup>-1</sup>, which represent the highest ECs  
418 recorded in a single-phase BTF treating CH<sub>4</sub>. A high diversity was found in the reactor  
419 through the experimental period showing high functional resiliency and redundancy and  
420 maintaining high EC despite changes in the bacterial community. Type I methanotrophs  
421 were dominant, proving that high Cu<sup>2+</sup> concentrations in the recycling MSM was a  
422 successful strategy to promote their growth.

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#### 424 **5. Acknowledgements**



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552 **Figure captions**

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2 553 **Figure 1.** Schematic representation of the experimental set-up. 1. Ambient air  
3 554 compressor, 2. Humidifying column, 3. CH<sub>4</sub> reservoir, 4. Mass flow controllers, 5.  
4 555 Mixing chamber, 6. Gas sampling ports, 7. Stirred tank, 8. Liquid recycling pump, 9. Gas  
5 556 recycling compressor, 10. Liquid sampling port.  
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11 558 **Figure 2.** Influence of the gas EBRT and liquid recycling velocity ( $U_L$ ) on  $k_{La}$  for  
12 559 CH<sub>4</sub>(A) and on the pressure drop in the packed bed (B).  
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17 561 **Figure 3.** Time course of (A) Loading rate (grey line) and EC (white diamonds), and (B)  
18 562 CO<sub>2</sub> production rate (black circles) during CH<sub>4</sub> biodegradation in a BTF. Horizontal  
19 563 arrows indicate the MSM exchange rates, while vertical lines indicate the different  
20 564 operational stages: 1.  $U_L = 2.3 \text{ m h}^{-1}$ , 2.  $U_L = 5 \text{ m h}^{-1}$ , 3. Internal gas recycling at  $18 \text{ L}$   
21 565  $\text{min}^{-1}$ , 4. Gas recycling stopped, 5.  $U_L = 15 \text{ m h}^{-1}$ .  
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27 567 **Figure 4.** Time course of the EC (white diamonds) and loading rate (black squares)  
28 568 during the mass transfer limitation test.  
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33 570 **Figure 5.** CH<sub>4</sub> concentration profile representing the mass transfer processes occurring  
34 571 in the BTF and the hypothetical mass transfer resistance governing the process.  $C_G =$   
35 572 bulk concentration in the gas phase,  $C_G^* =$  Gas phase concentration at the gas-liquid  
36 573 interphase,  $C_L^* =$  Liquid phase concentration in equilibrium with the gas phase,  $C_L =$   
37 574 bulk concentration in the liquid phase,  $C_{L,B}^* =$  Liquid phase concentration at the liquid-  
38 575 biofilm interphase,  $C_B^* =$  Biofilm concentration in equilibrium with the liquid phase.  
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45 577 **Figure 6.** Bacterial DGGE profile of the four cultures acclimated at different Cu<sup>2+</sup> (A,  
46 578 B, C and D corresponding to 5, 10, 25 and 50  $\mu\text{M}$ , respectively), the mixed microbial  
47 579 inoculum (E), the population present in the BTF at day 38 (F) and 104 (G). The  
48 580 Shannon-Wiener diversity indices are indicated in the upper part of the gel. The  
49 581 sequenced bands are indicated by “▶” and the corresponding number of each band.  
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**Table 1.** Comparative evaluation of energy consumption and GHG mitigation efficiency of internal gas recycling under two different scenarios. A reference (without gas recycling) and a gas recycle operation mode are considered for each scenario.

	Scenario 1		Scenario 2	
	Ref.	Recycle	Ref	Recycle
<b>Gas recycling ratio</b> (Recycling flow/Feed flow)	0	18	0	1
<b>Conditions</b>	$V_L$ (m h <sup>-1</sup> )	5	2	2
	<b>Virtual EBRT</b> (s)	240	12	240
<b>Energy consumed</b> (energy units)	1	19	1	2
<b>CH<sub>4</sub> removed*</b> (CH <sub>4</sub> units)	1	3	0.25	0.75
<b>CO<sub>2</sub> eq. removed</b> (CO <sub>2</sub> units)	20	60	5	15
<b>CO<sub>2</sub> eq. removed / energy consumed</b>	20	3.2	5	7.5
<b>Annual Operating Costs with gas recycling</b>	1	3.6	1	1.1
<b>Annual Operating Costs with EBRT increase**</b>	1	2.5	1	2.5

\*Comparative data based on the experimental data from the present study.

\*\*Theoretically calculated by increasing the EBRT of the unit to achieve a similar CH<sub>4</sub> removal to that obtained under internal gas recycling.

Figure 1  
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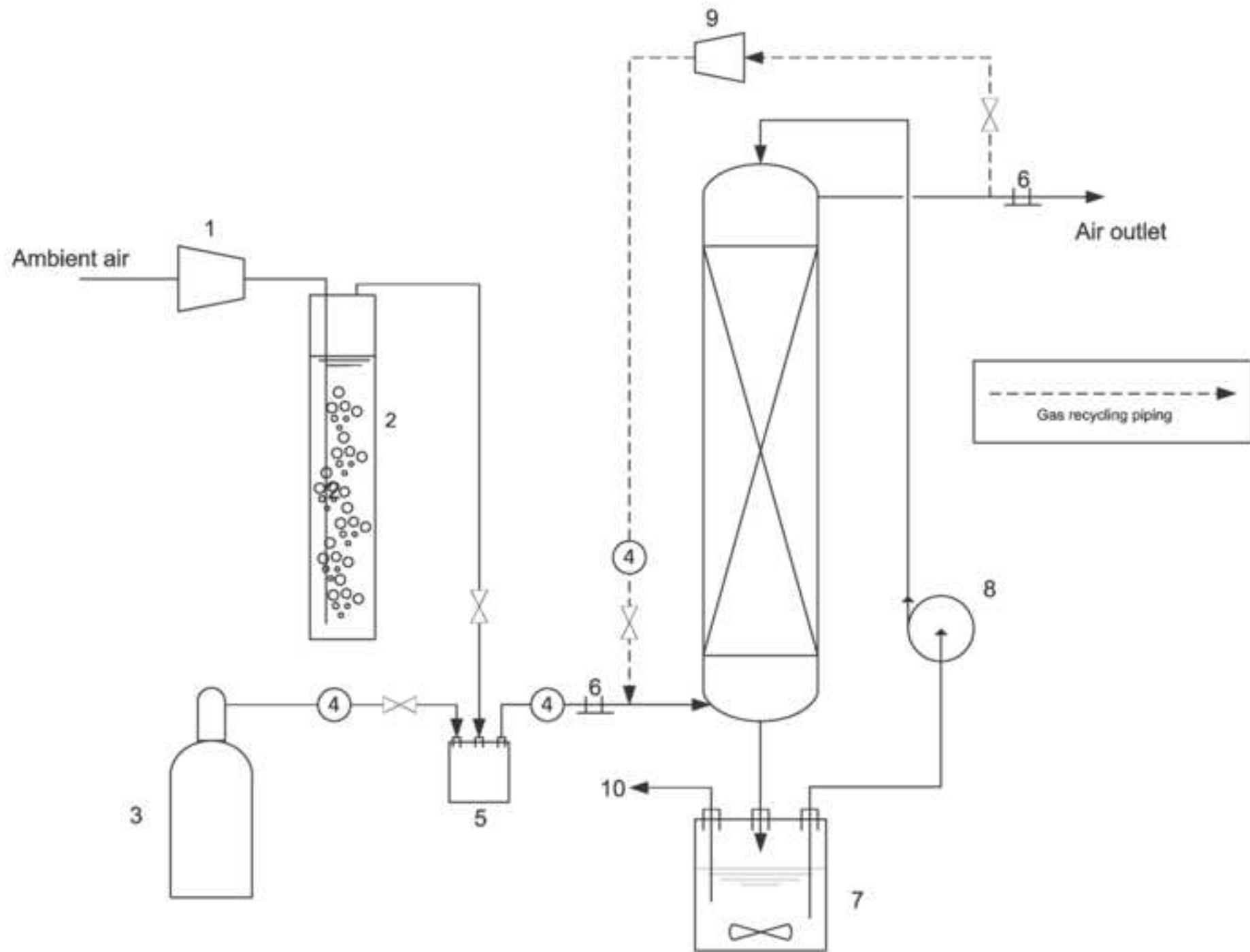




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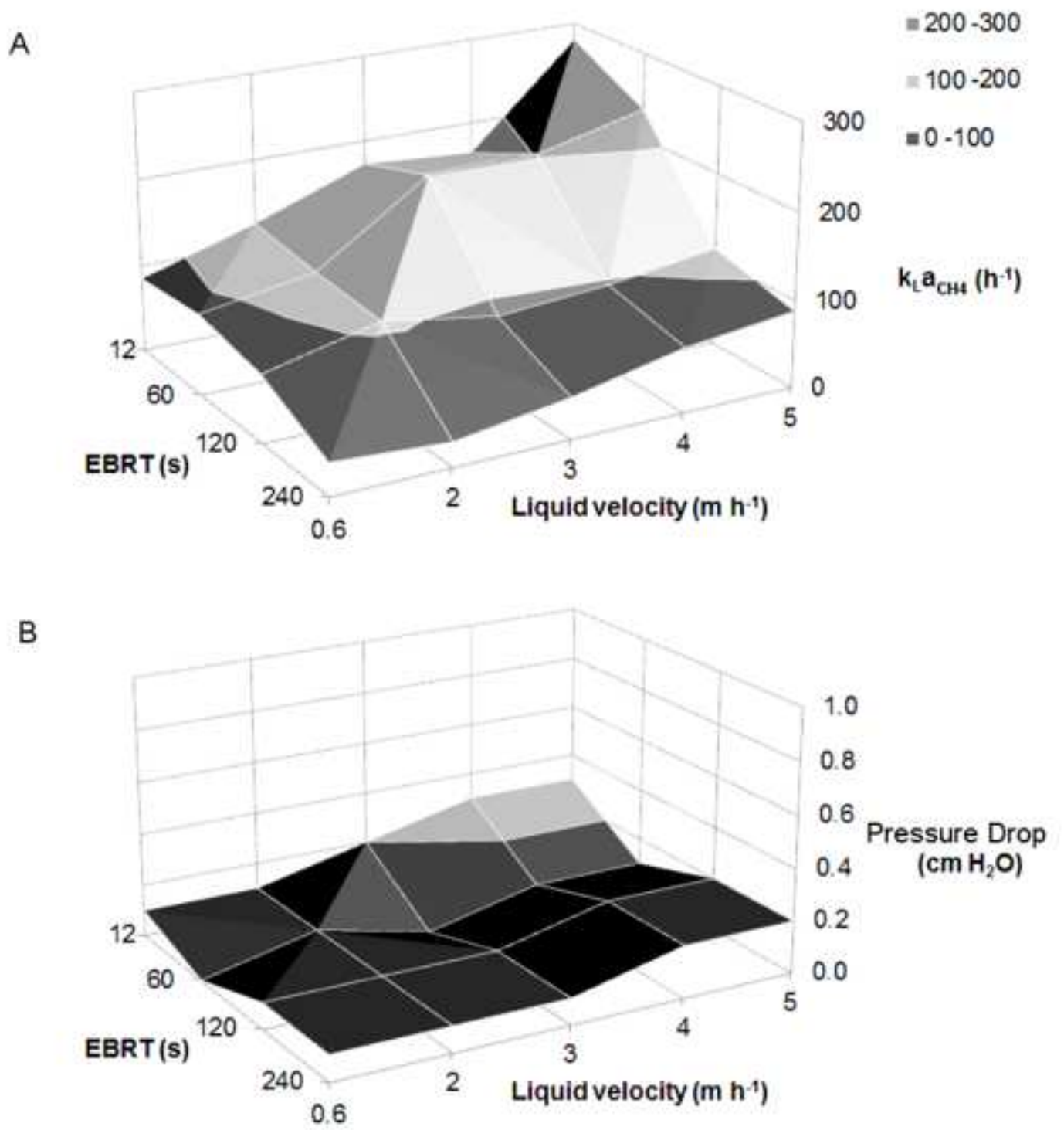


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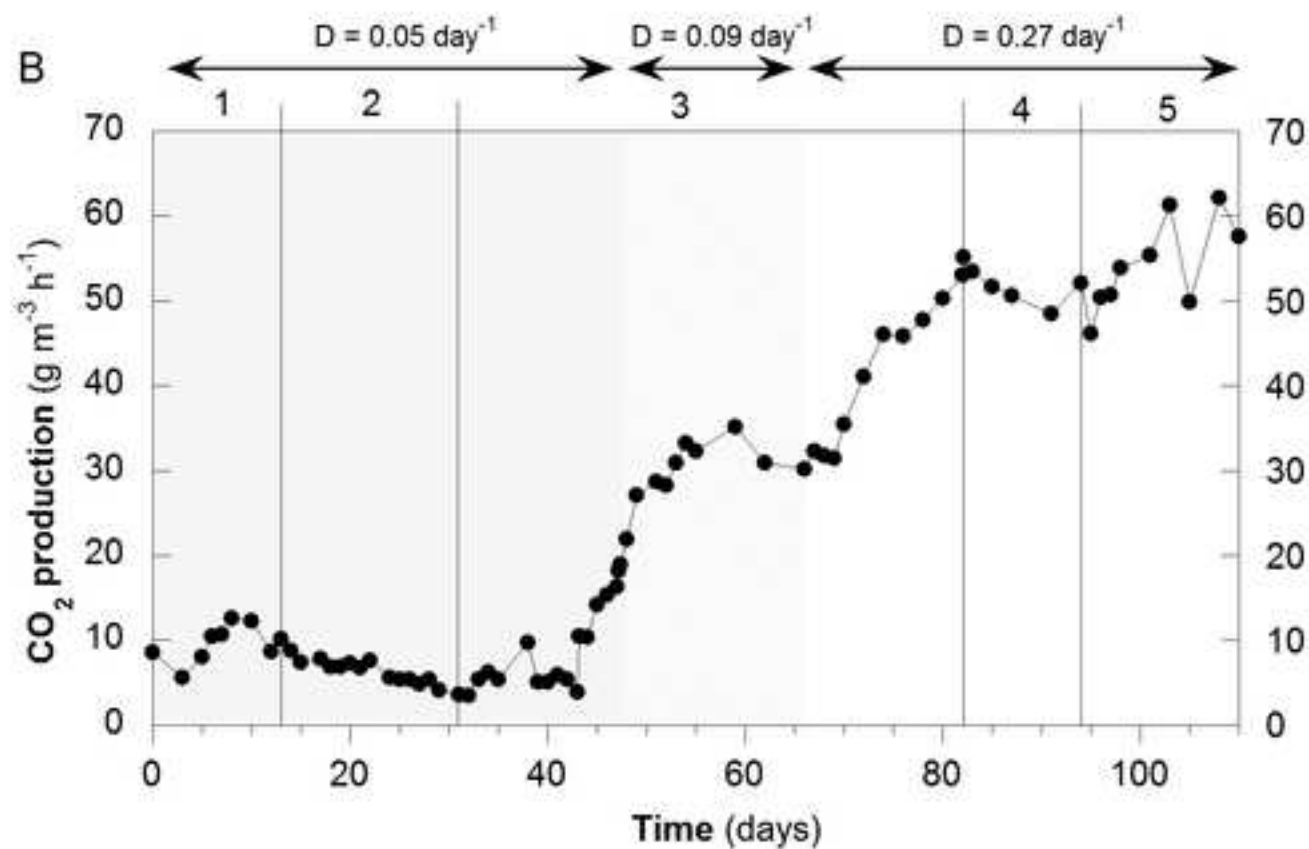
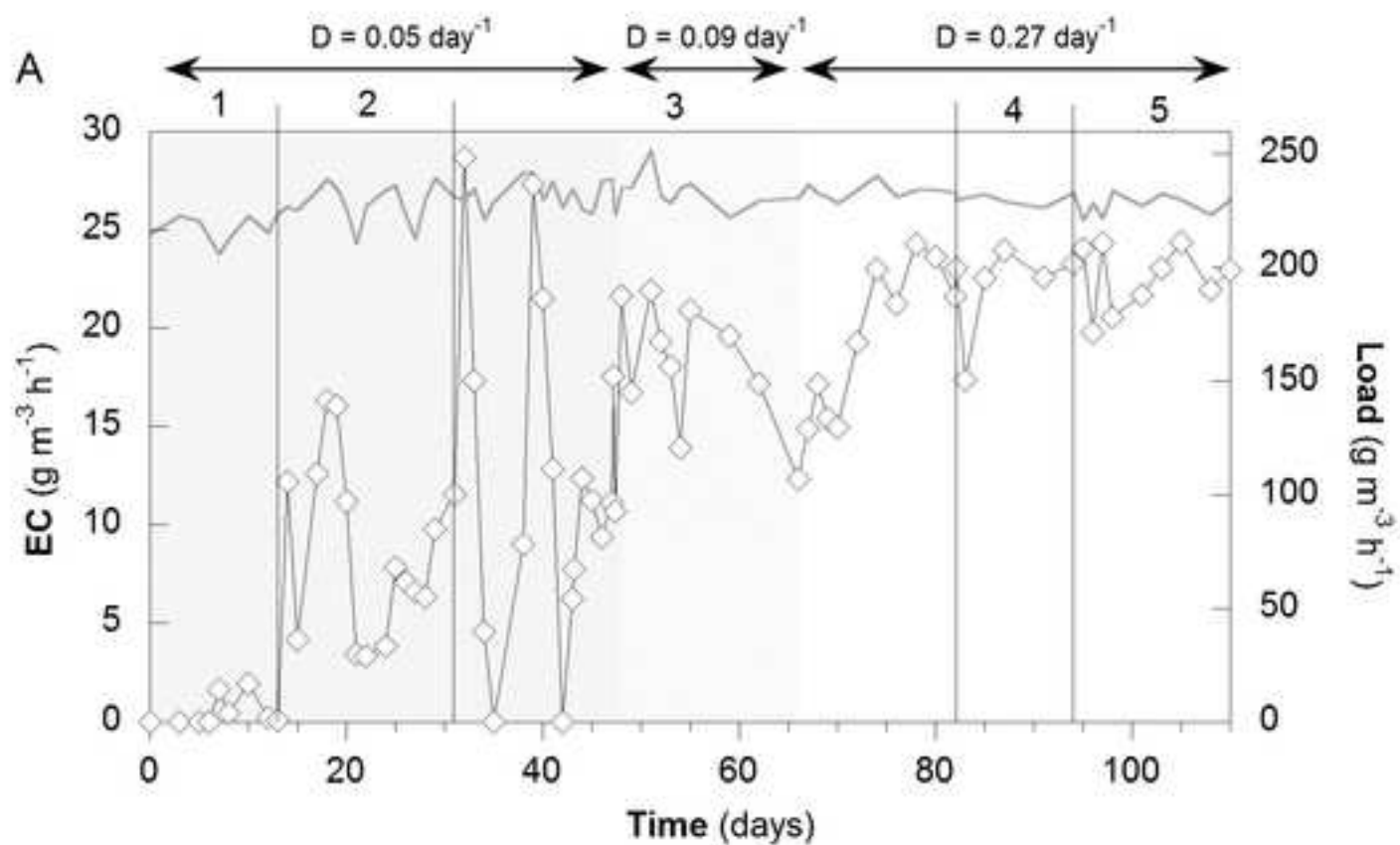


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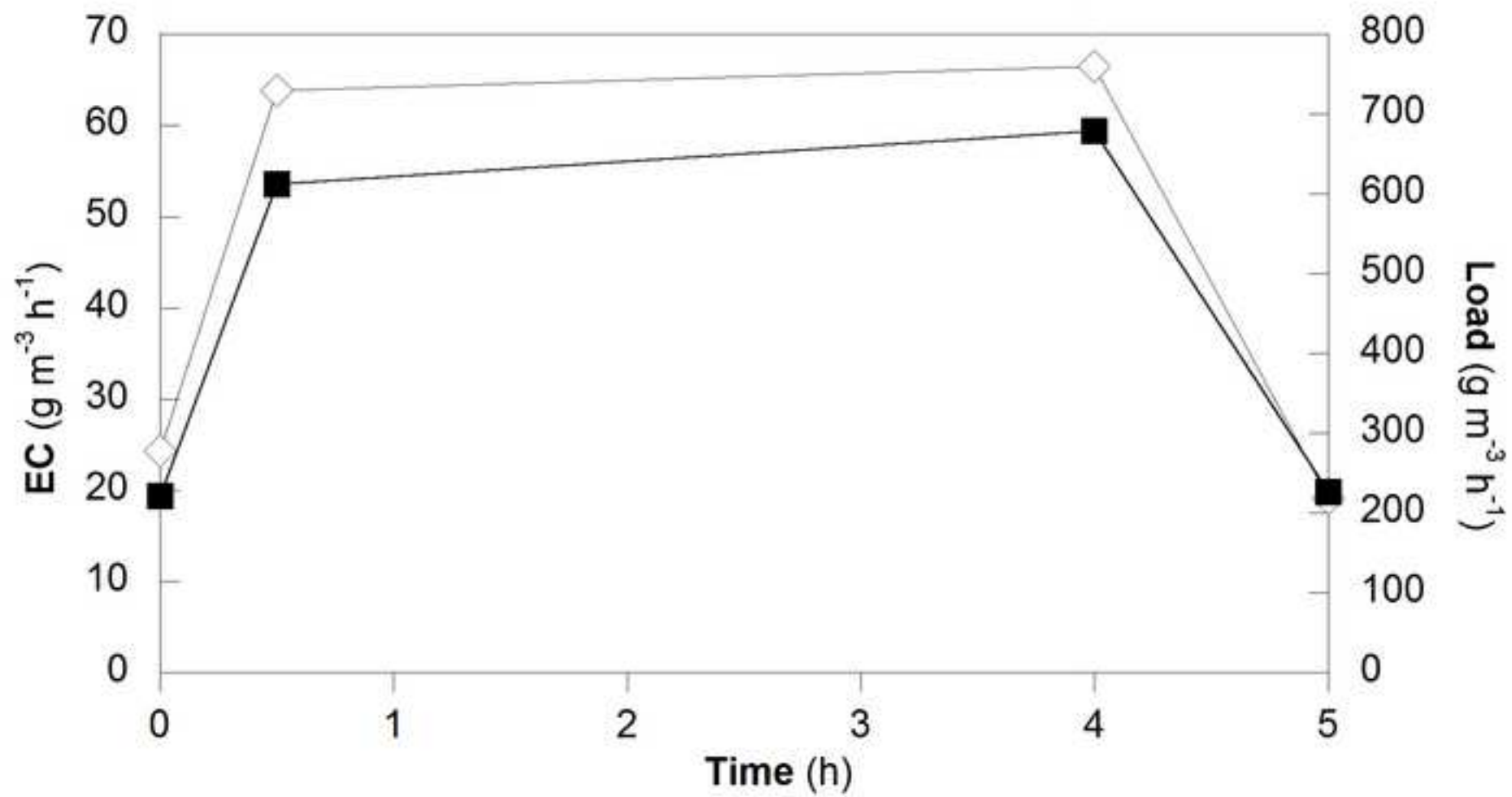
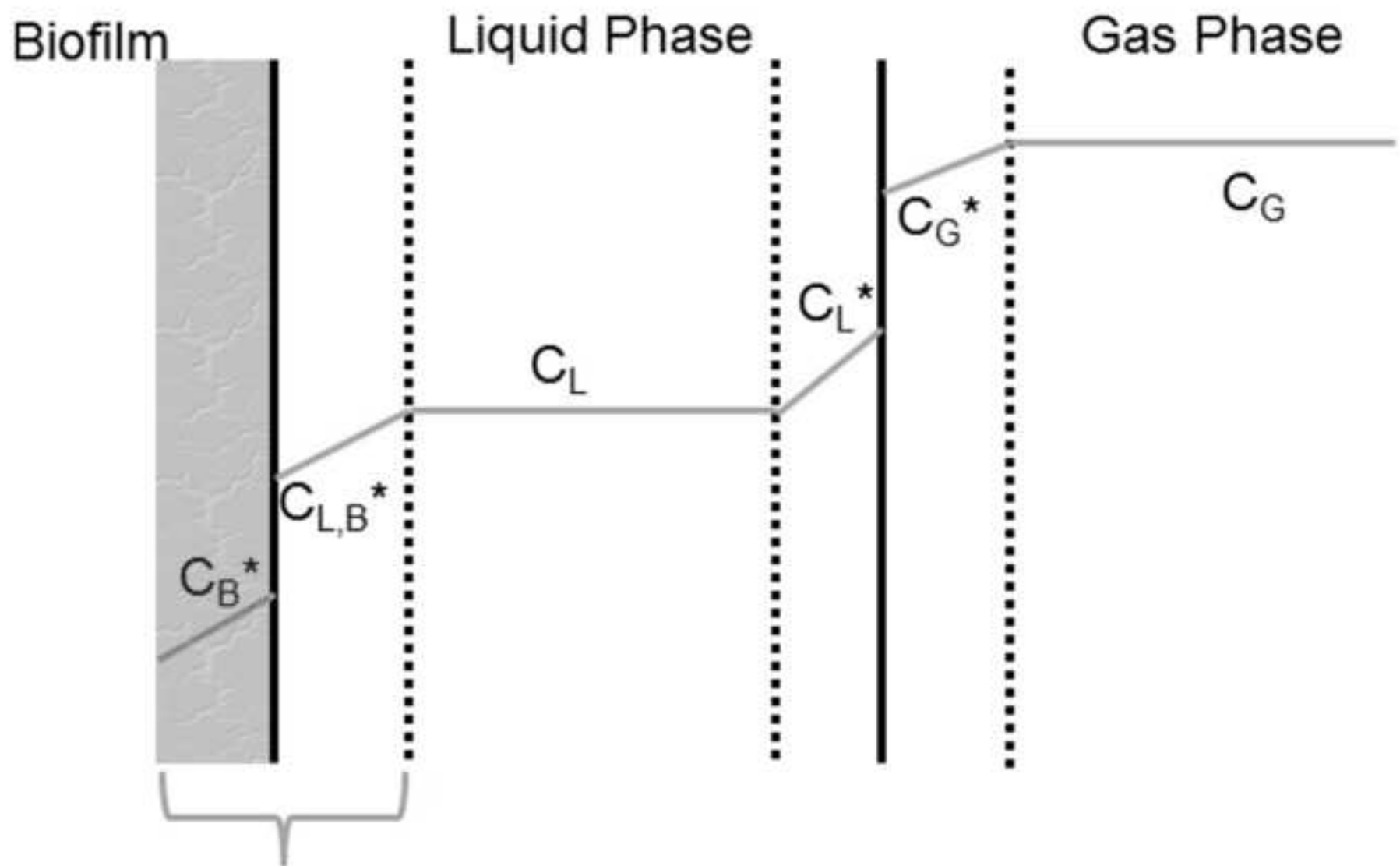
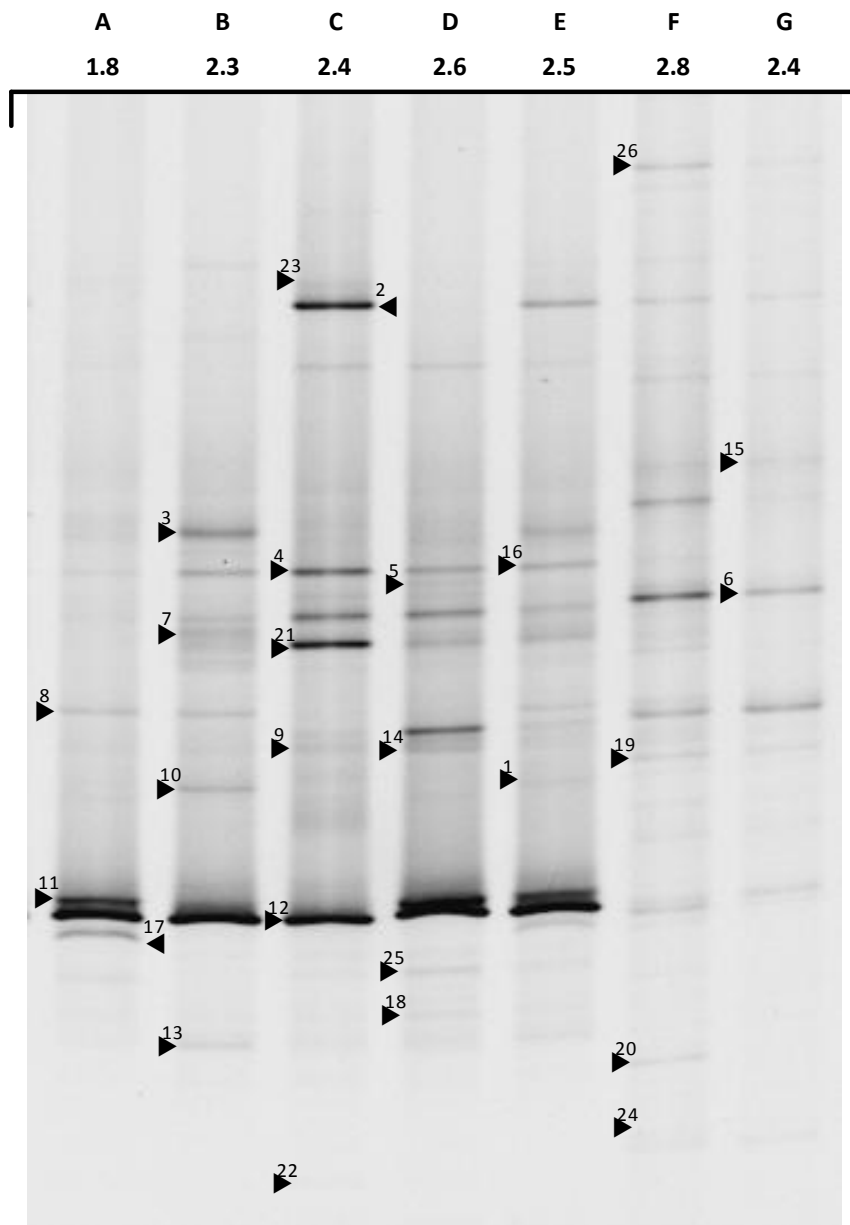


Figure 5  
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Location of the hypothetical limitation governing the overall mass-transfer rate.

Figure 6.



**Supplementary Material Tble 2**

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